



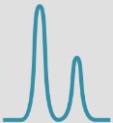
## Cutting Through the Haze: Real-Time Air Monitoring in Action

Kraig Kmiotek  
Product Manager, Process Wet Chemistry  
Metrohm USA

In today's presentation we will cover...



Current state of air quality monitoring



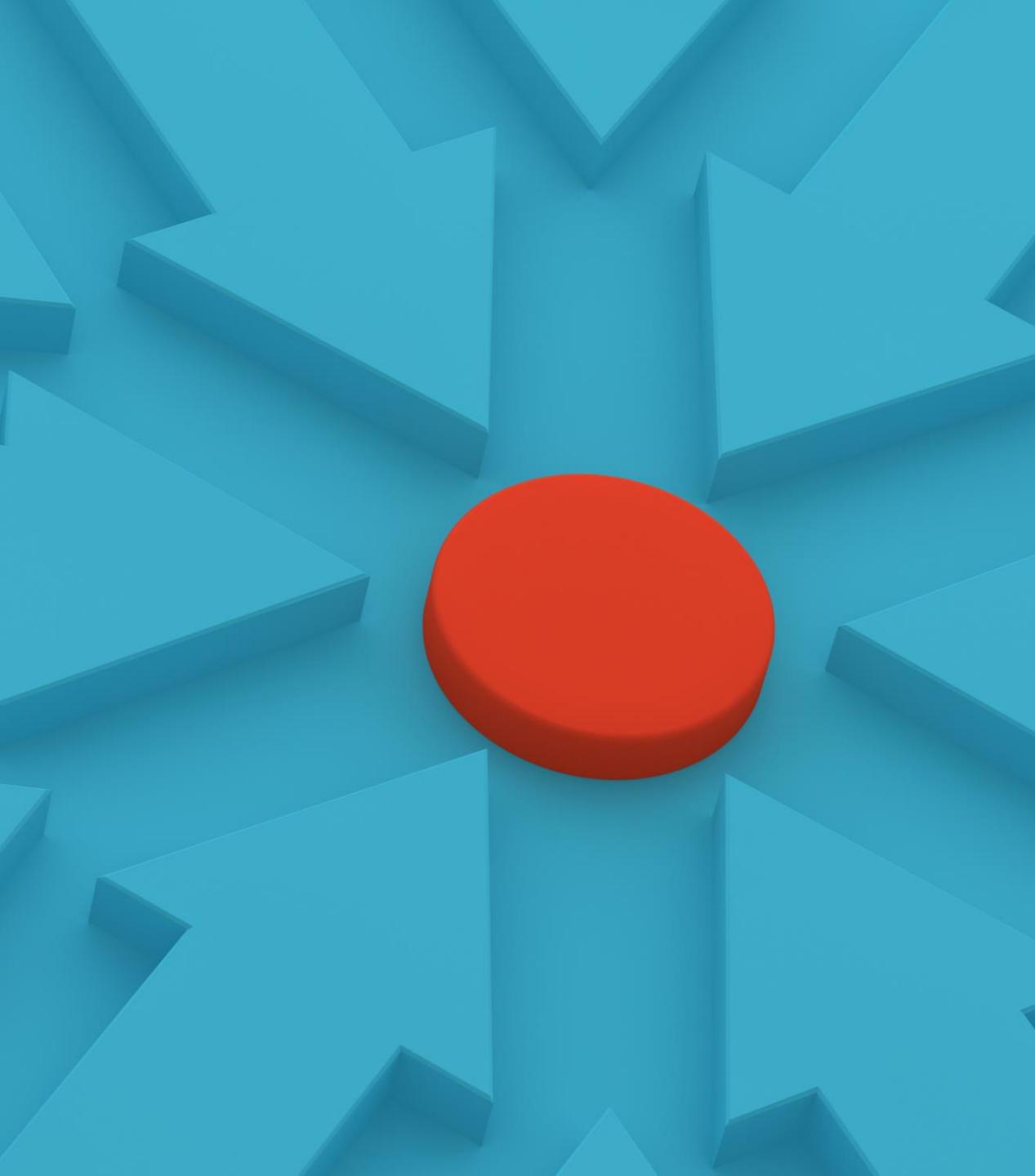
Monitor for AeRosols and Gases in ambient Air (MARGA)



Case study: Firework-related pollution events on New Year's Eve



Field Studies: MARGA in action

The background of the left half of the slide features a complex, abstract pattern of blue 3D geometric shapes, including triangles and rectangles, creating a sense of depth. A single, solid red circle is positioned in the center of the pattern.

# Current State of Air Quality Monitoring

# Air Quality Monitoring Today



- ✓ In recent years, there's been a shift toward **modernizing air quality monitoring**.
- ✓ New regulations and health data highlight the need for **higher time resolution and chemical specificity**.
- ✓ The EPA's **Air Quality Monitoring 2.0** initiative promotes hybrid networks, **pairing low-cost sensors with reference-grade instruments**.
- ✓ Traditional methods still play a role but often **miss short-term pollution events**.
- ✓ Emerging technologies are now making it possible to **track both gases and aerosols in real time**, with far greater detail than before.

# Current Methodologies

## Filter-Based Sampling

- **Use:** Measuring PM2.5 or PM10 mass and composition.
- **How:** Air is drawn through filters for 24 hours. Filters are weighed and analyzed in a lab (e.g., for sulfates, nitrates, metals).
- **Common in:** Regulatory monitoring (EPA, EEA), long-term studies.

## Continuous Gas Analyzers

- **Use:** Real-time monitoring of gases like NO<sub>x</sub>, SO<sub>2</sub>, O<sub>3</sub>, and CO.
- **How:** Instruments use specific detection principles (e.g., UV photometry, chemiluminescence).
- **Common in:** Urban air monitoring stations, compliance sites.

# Current Challenges in Air Quality Monitoring



**Daily / 24-hour  
integrated samples**



**Delayed data availability  
due to required lab  
processing**



**Labor intensive due to  
manual collection,  
transportation and lab  
work**



**Low resolution of events  
due to result being an  
average over time**



# Monitor for AeRosols and Gases in ambient Air (MARGA)

# 2060 MARGA Process Analyzer

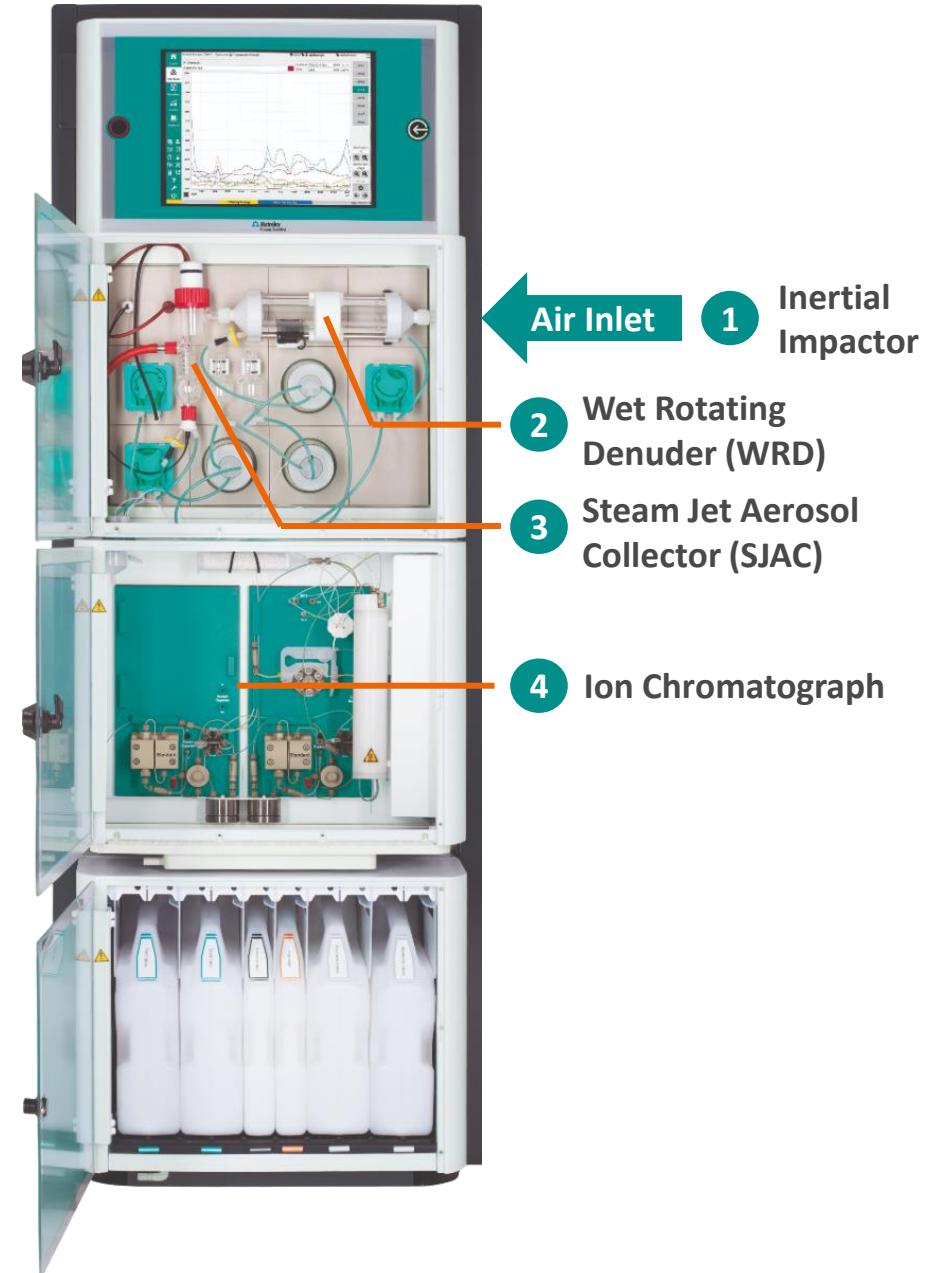
## Aerosols

— Cl <sup>-</sup>	— Na <sup>+</sup>
— NO <sub>3</sub> <sup>-</sup>	— Ca <sup>2+</sup>
— SO <sub>4</sub> <sup>2-</sup>	— Mg <sup>2+</sup>
— NH <sub>4</sub> <sup>+</sup>	— K <sup>+</sup>

— F <sup>+</sup>
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## Gases

— HCl	— SO <sub>2</sub>
— HNO <sub>3</sub>	— NH <sub>3</sub>
— HONO	— HF



# 2060 MARGA Configurations



## 2060 MARGA R

Ideal for campaigns  
IC can be used as stand-alone lab instrument

Ideal for campaigns

IC can be used as stand-alone lab instrument



## 2060 MARGA M

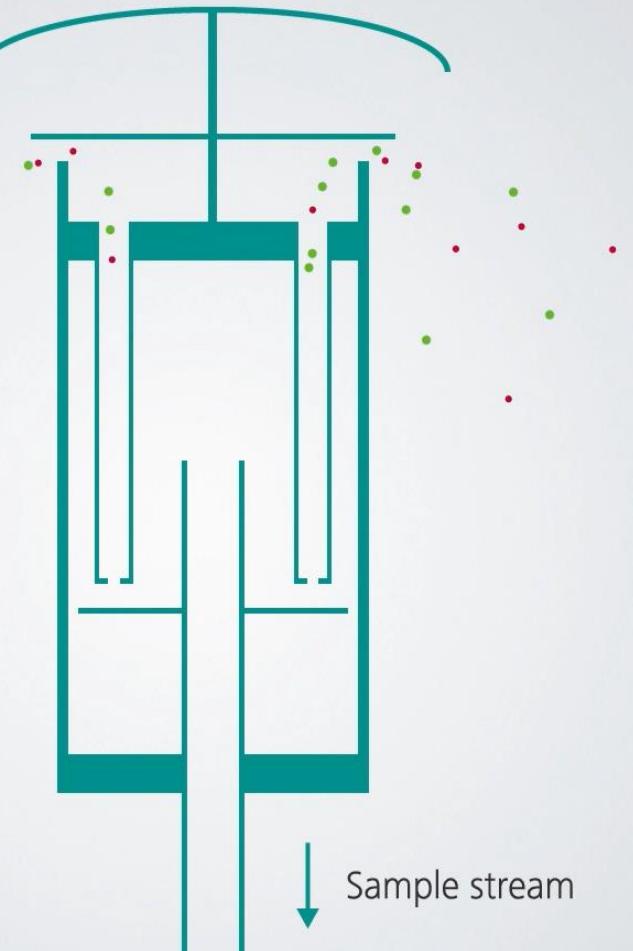
Continuous monitoring at  
permanent site

# Inertial Impactor

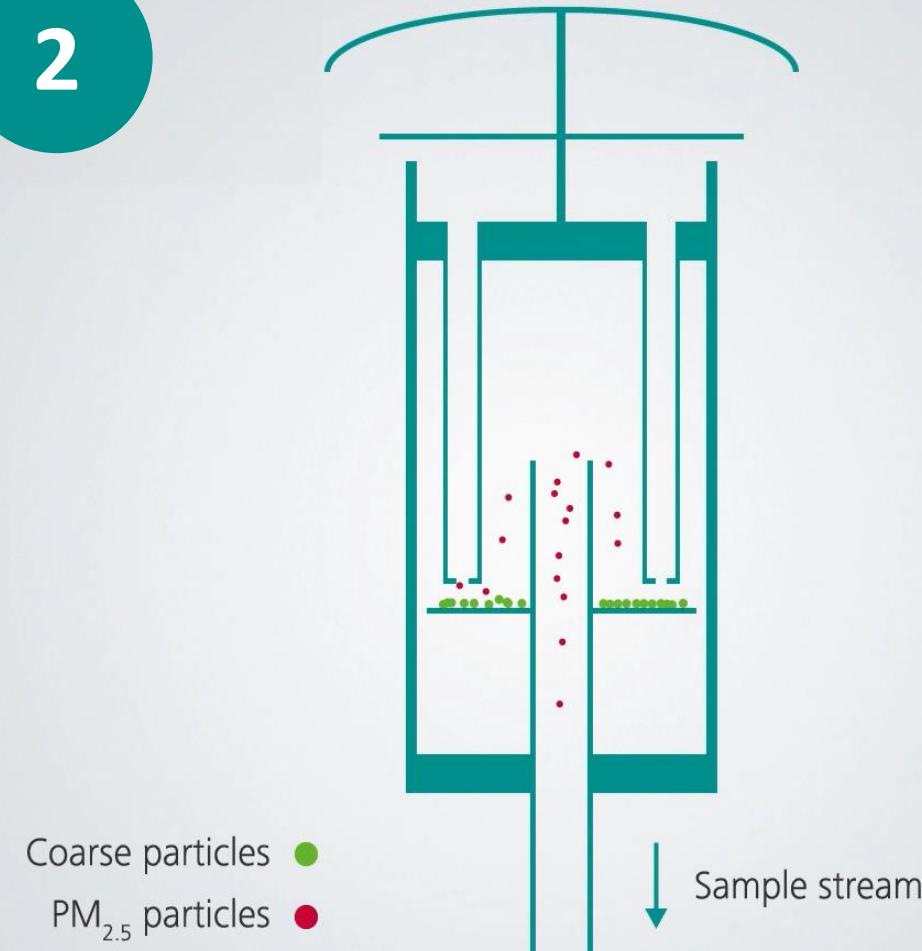


# Air Inlet – Inertial Impactor

1



2

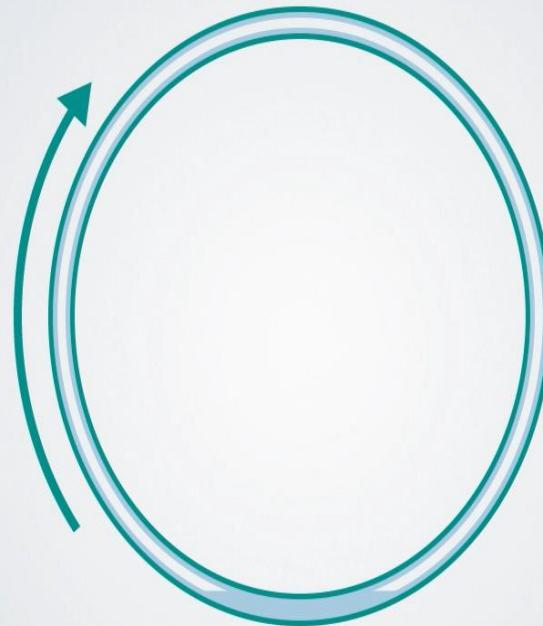


# Wet Rotating Denuder (WRD)



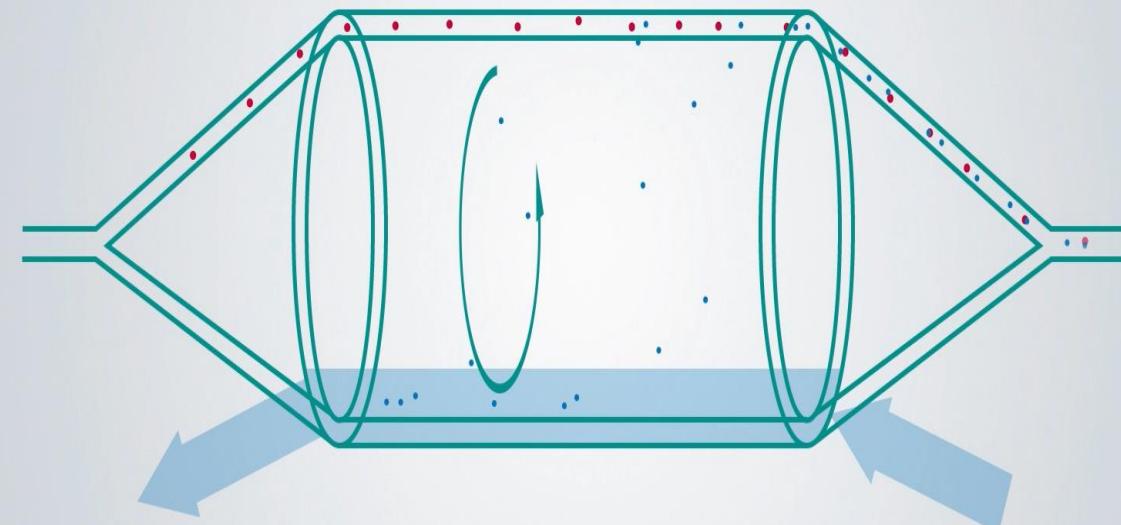
# Wet Rotating Denuder (WRD)

1



The WRD rotates to create a large surface area liquid film to absorb the gases

2



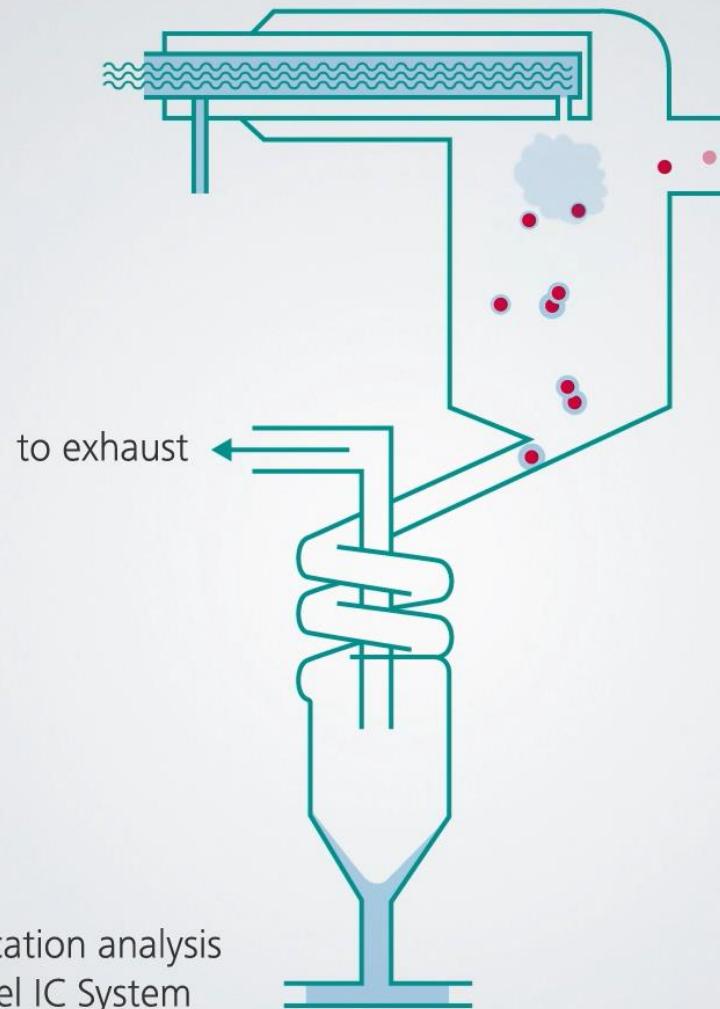
To anion and cation analysis by  
dual-channel IC System

# Steam Jet Aerosol Collector (SJAC)



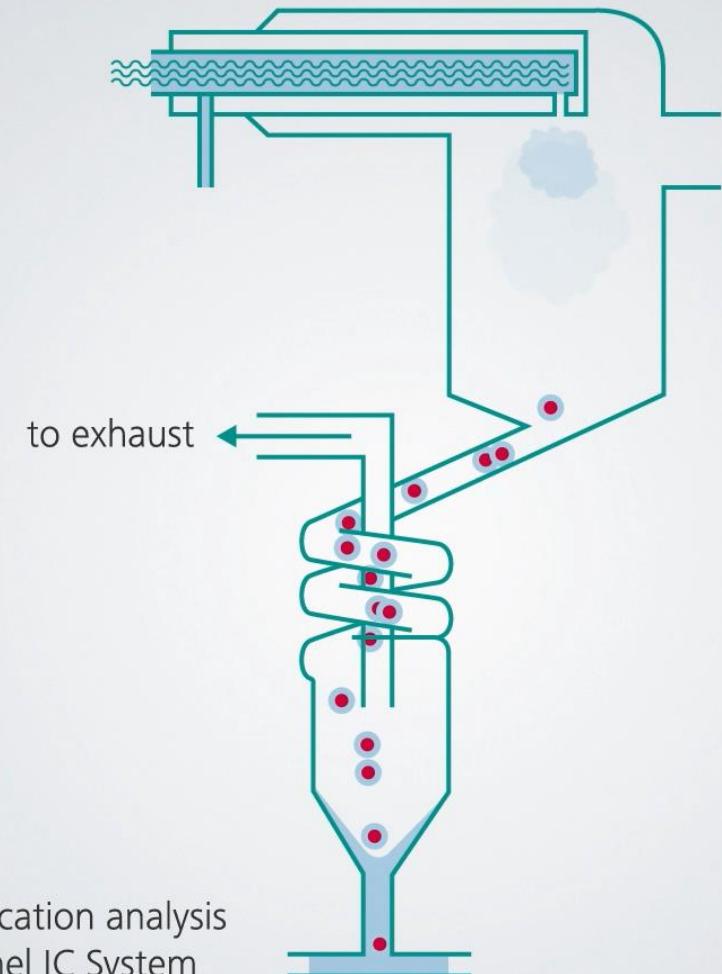
# Steam Jet Aerosol Collector (SJCA)

1



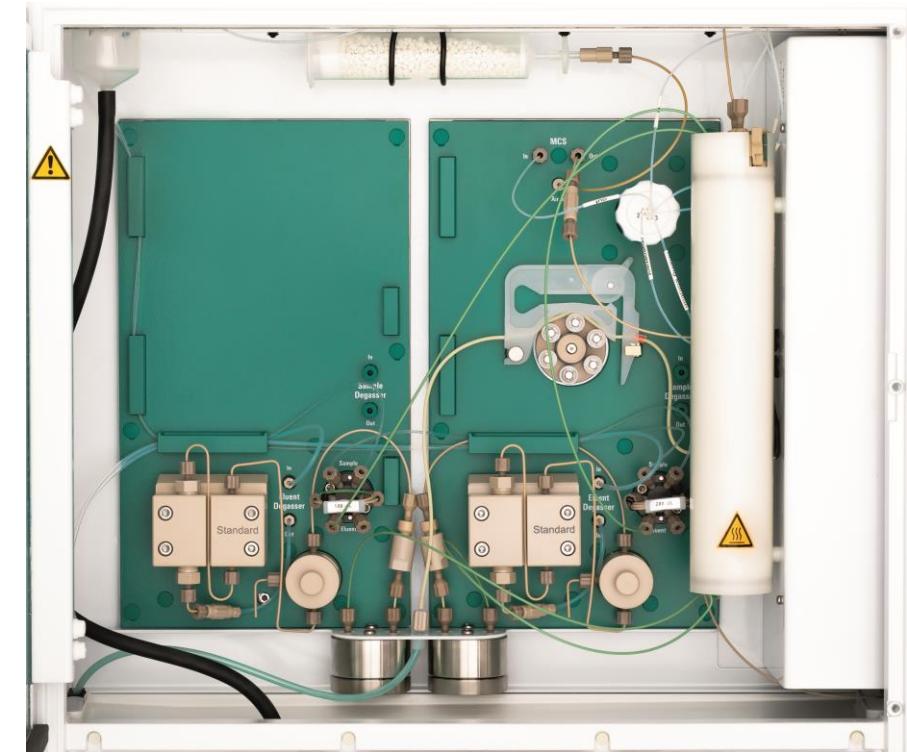
To anion and cation analysis  
by dual-channel IC System

2

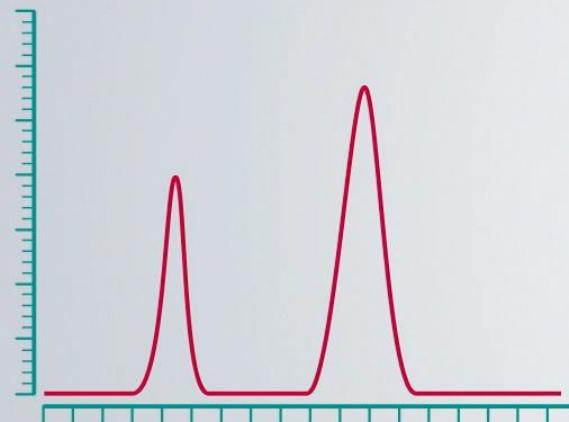


To anion and cation analysis  
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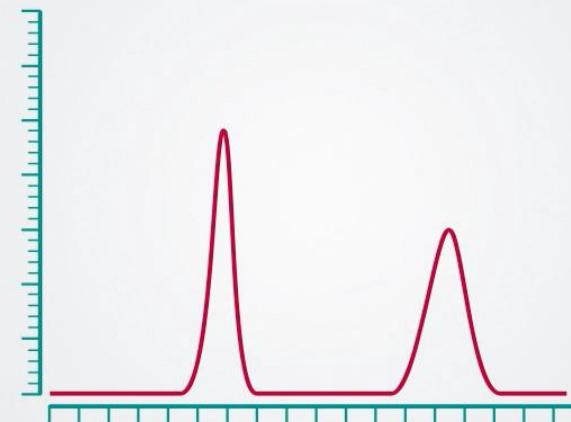
# Ion Chromatograph



# Dual Channel Ion Chromatography



Anion analysis  
(of aerosol and gas fraction)



Cation analysis  
(of aerosol and gas fraction)



# Limits of Detection

- The 2060 MARGA offers much lower detection limits compared to the original MARGA, with improvements up to 10 times more sensitive.
- Chloride, nitrate, and sulfate are now detectable at concentrations as low as 0.001 to 0.002  $\mu\text{g}/\text{m}^3$ .
- The 2060 MARGA adds support for new analytes like lithium and bromide, which were not measurable with the previous system.
- These lower limits allow for better detection of background levels, earlier identification of pollution events, and more precise source analysis.

**Sample:** 1  $\text{m}^3$  ambient air

Water soluble components absorbed into 15 mL absorbance liquid.

Component	LOD in air ( $\mu\text{g}/\text{m}^3$ )	
	2060 MARGA	MARGA
Lithium	0.005	---
Sodium	0.02	0.05
Ammonium	0.02	0.05
Potassium	0.05	0.09
Magnesium	0.02	0.06
Calcium	0.05	0.09

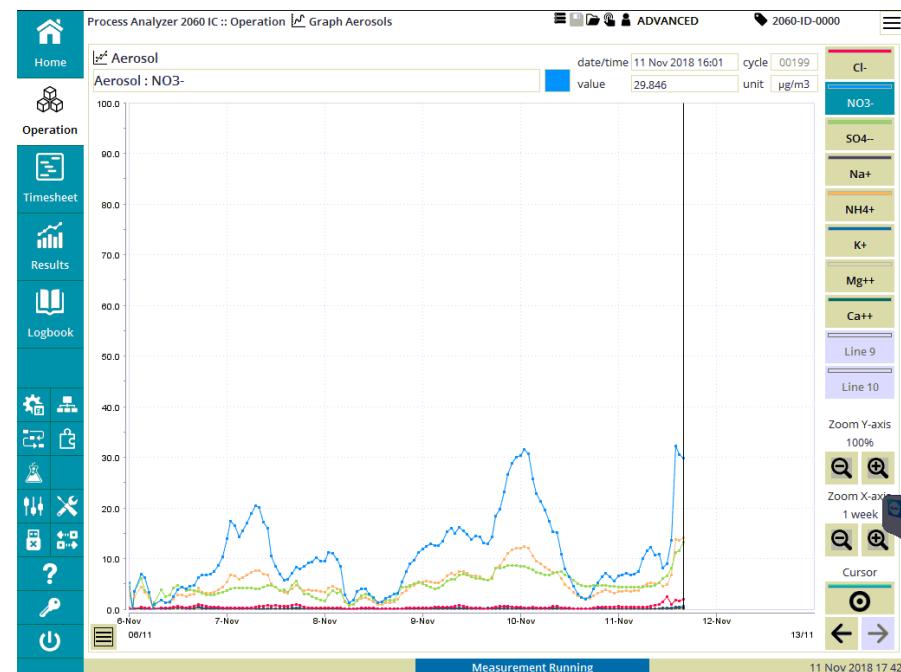
Component	LOD in air ( $\mu\text{g}/\text{m}^3$ )	
	2060 MARGA	MARGA
Chloride	0.001	0.01
Nitrite	0.001	0.02
Bromide	0.002	---
Nitrate	0.002	0.05
Sulfate	0.002	0.03



- Real-time data** – process data is collected and displayed on the touchscreen user interface instantly
- Data integrity** – All data are stored in an encrypted database to prevent data tampering
- Remote capabilities** – Access the analyzer from anywhere using TeamViewer or VNC (cellular modem required)

# Data Handling with IMPACT

## Trend Graph - Aerosols



## Trend Graph - Gases





## Case study: Firework-related pollution events on New Year's Eve

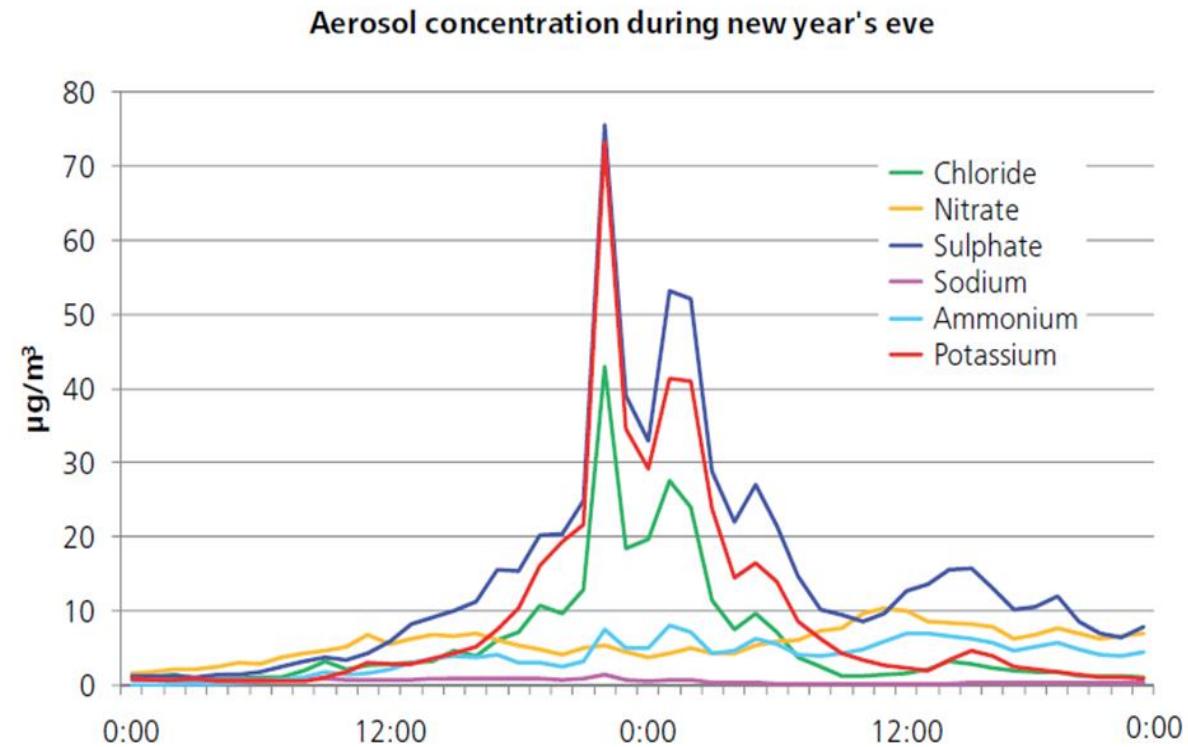
# Air Pollution from Fireworks



- Fireworks release a rapid burst of air pollutants, including particulate matter (PM), metallic compounds, and gaseous byproducts.
- Short-term pollution spikes in PM<sub>2.5</sub> and PM<sub>10</sub> are common and concentrations can rise by several hundred percent within minutes.
- **Typical cations detected:**
  - K<sup>+</sup> (Potassium) – Used as an oxidizer; prominent in colored fireworks
  - Na<sup>+</sup> (Sodium) – Produces yellow color
  - Mg<sup>2+</sup>, Ca<sup>2+</sup>, Ba<sup>2+</sup>, Sr<sup>2+</sup>, Cu<sup>2+</sup> – From metal salts used for color effects
- **Typical anions detected:**
  - Cl<sup>-</sup> (Chloride) – Combines with metals to form colorant salts
  - NO<sub>3</sub><sup>-</sup> (Nitrate) – Common oxidizer
  - SO<sub>4</sub><sup>2-</sup> (Sulfate) – From sulfur-containing compounds used as fuel or stabilizers
- **Gaseous species may include:**
  - SO<sub>2</sub>, NO<sub>x</sub>, CO, and sometimes ozone precursors

# 2060 MARGA and Fireworks

- The 2060 MARGA captured a clear, time-resolved pollution event tied to fireworks at midnight.
- Multiple aerosol components increased simultaneously, showing the complex chemical signature of firework emissions.
- The event was short-lived but intense, underscoring the need for hourly resolution in air quality monitoring.
- This type of data enables source identification, event attribution, and supports regulatory and public health analysis.
- Traditional 24-hour methods would have missed both the timing and severity of the event.





# Field Studies: MARGA in Action

# The importance of vehicle emissions as a source of atmospheric ammonia in the megacity of Shanghai

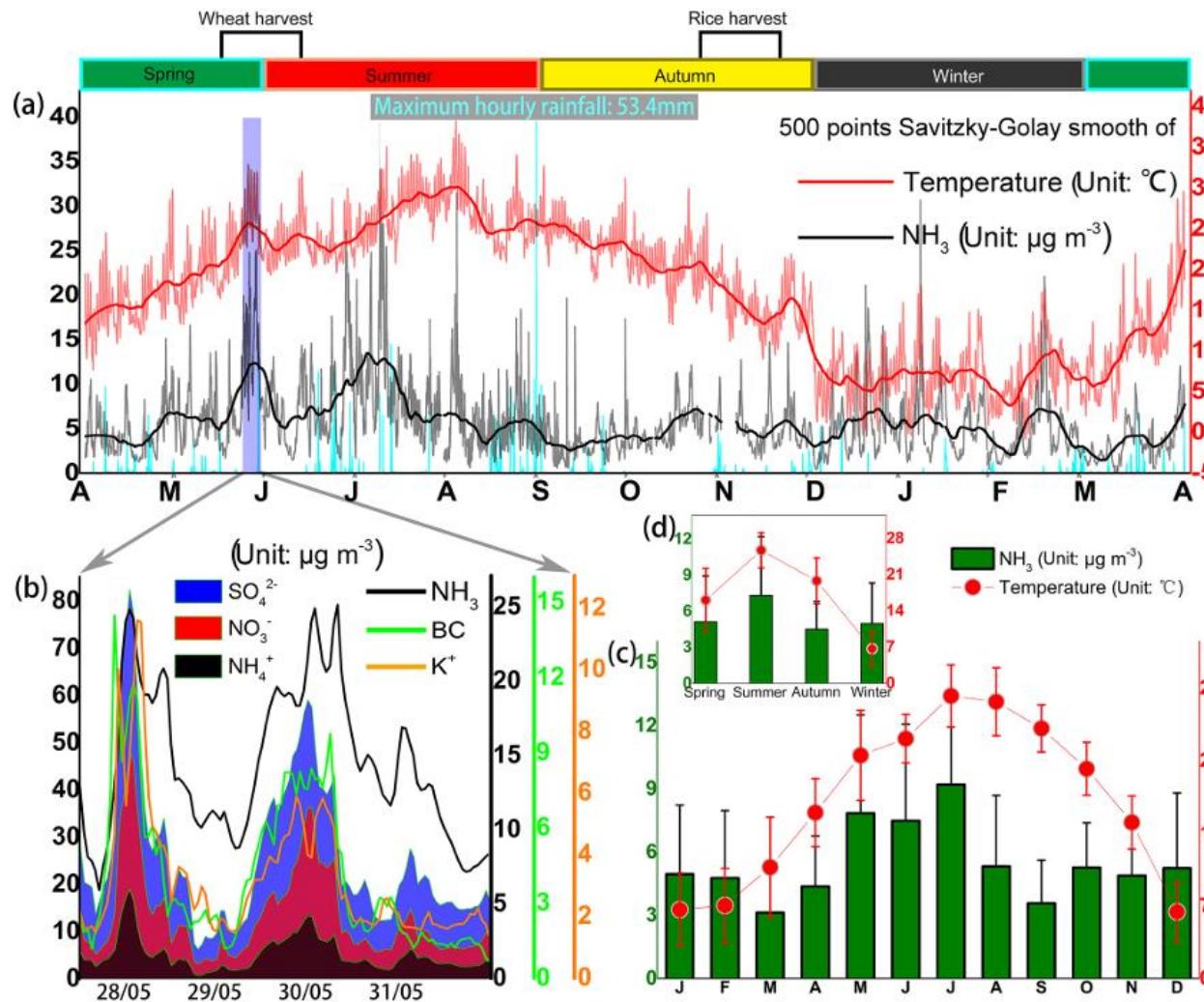
Researchers used the MARGA system to measure hourly concentrations of  $\text{NH}_3$ ,  $\text{NH}_4^+$ , and related aerosol species in the North China Plain over a full year.

The data revealed clear seasonal variation in  $\text{NH}_3$ , with highest levels during summer and harvest periods, likely tied to fertilizer use and temperature-driven volatilization.

A strong positive correlation between temperature and  $\text{NH}_3$  was observed, reinforcing the role of thermally driven emissions.

Episodes of high secondary aerosol formation were linked to elevated  $\text{NH}_3$ ,  $\text{SO}_4^{2-}$ , and  $\text{NO}_3^-$  concentrations.

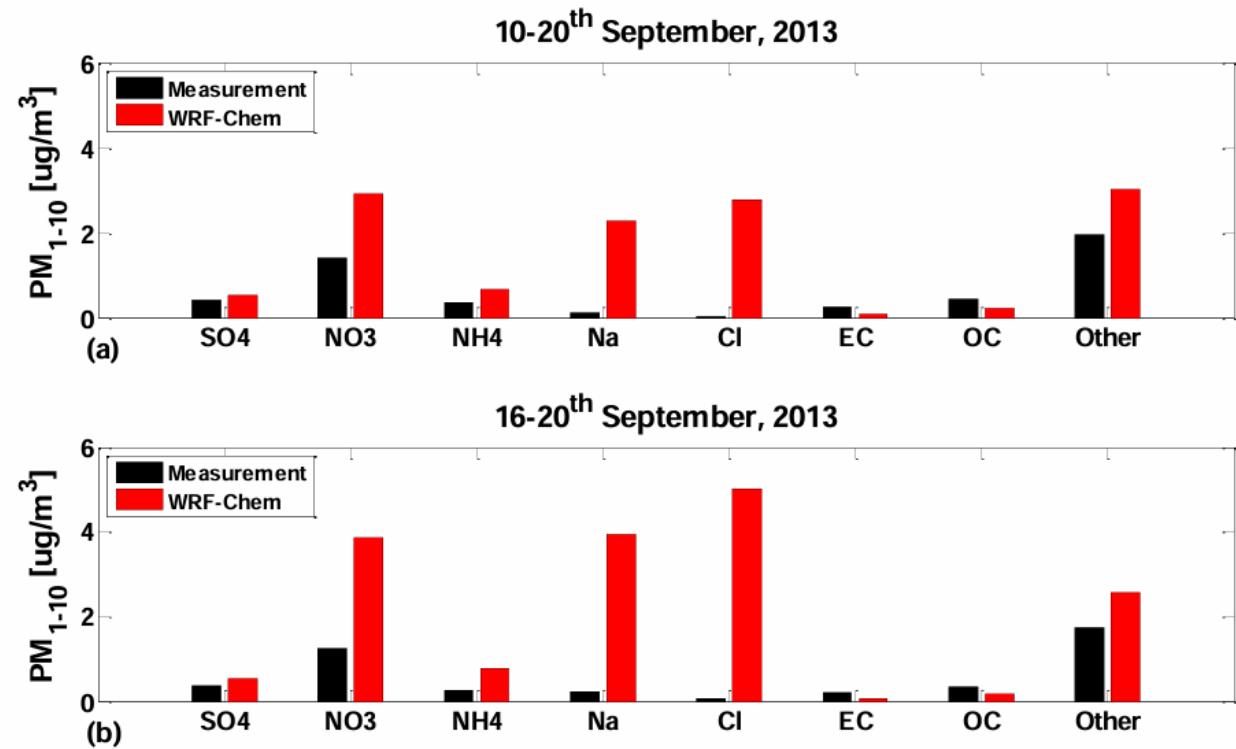
These results highlight the importance of continuous gas-aerosol monitoring for understanding ammonia chemistry and its role in regional haze formation.



(a) Temporal variations of hourly  $\text{NH}_3$  concentrations (gray) and temperature (red), along with 500-point Savitzky-Golay smoothed records in Shanghai from 3 April 2014 to 2 April 2015. Rainfall is shown in cyan. The vertical blue rectangle highlights  $\text{NH}_3$  pollution episodes that occurred during the wheat harvest season. (b) Time series of  $\text{NH}_3$ ,  $\text{BC}$ ,  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$ ,  $\text{NH}_4^+$ , and  $\text{K}^+$  concentrations during periods of pollution associated with biomass burning. Monthly (c) and seasonal (d) variations of  $\text{NH}_3$  average concentrations and temperature.

# Sea salt emission, transportation and influence on nitrate simulation: a case study in Europe

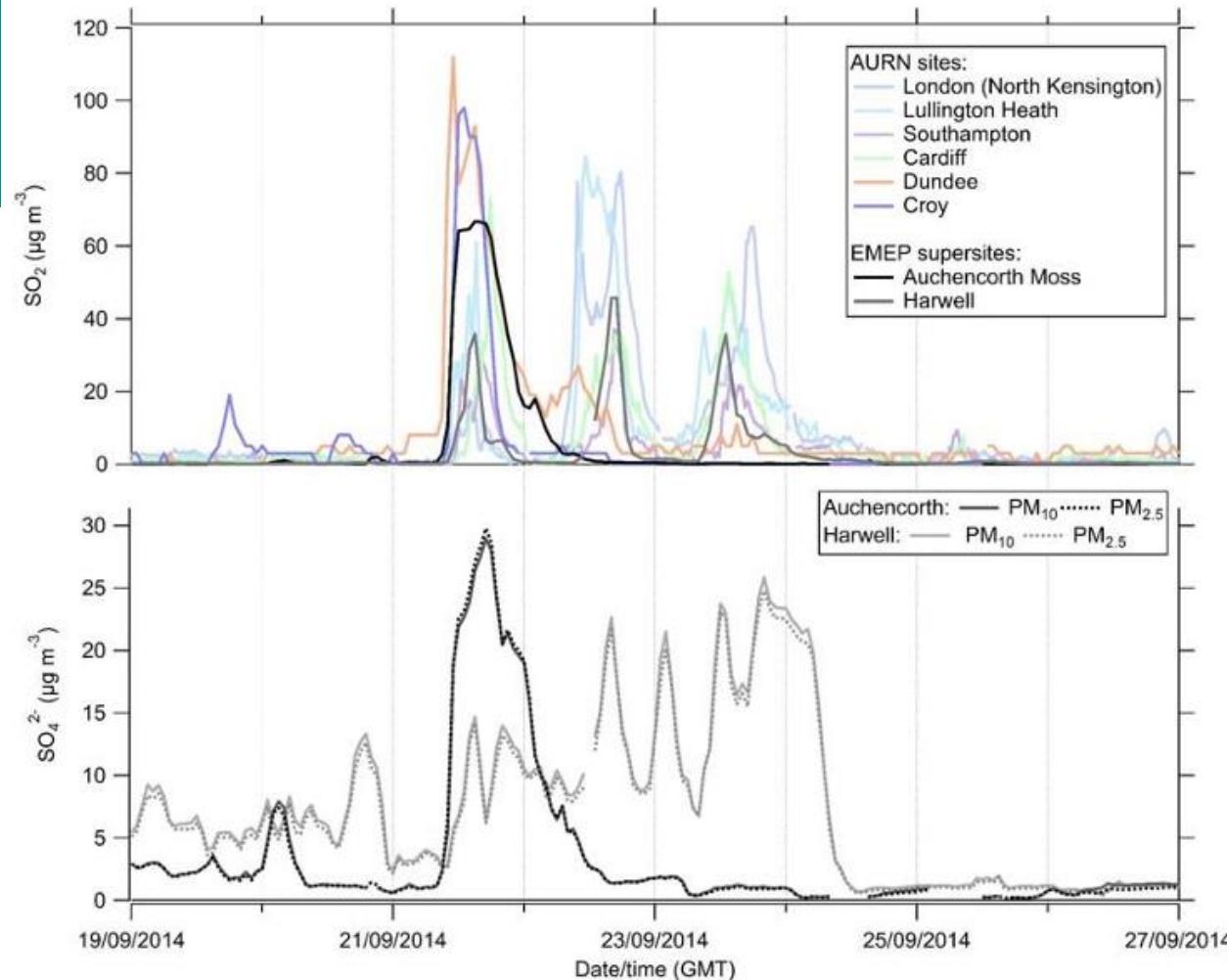
- Researchers used the MARGA system in Hong Kong to measure hourly concentrations of water-soluble ions in  $\text{PM}_{1-10}$  during a regional pollution event.
- The MARGA provided time-resolved data for sulfate, nitrate, ammonium, sodium, and chloride to evaluate model accuracy.
- These measurements were compared to predictions from the WRF-Chem atmospheric model.
- WRF-Chem overpredicted key species such as nitrate and chloride, especially during the later sampling period.
- The study shows how MARGA data can support model validation and improve future air quality forecasts.



Comparison of coarse mode aerosol ( $\text{PM}_{1-10}$ ) chemistry compounds between WRF-Chem model results and Melpitz measurements. (a) averaged in the HOPE-Campaign period of September 10-20, 2013; (b) averaged in the marine air mass period of September 16-20, 2013.

# Impacts of the 2014-2015 Holuhraun eruption on the UK atmosphere

- The study investigated a large-scale SO<sub>2</sub> and sulfate pollution event over the UK in September 2014.
- Researchers used the MARGA system at EMEP supersites (Auchencorth and Harwell) to measure hourly sulfate concentrations.
- The event was linked to industrial emissions transported from mainland Europe, confirmed by trajectory analysis.
- MARGA data showed a strong correlation between elevated SO<sub>2</sub> and secondary sulfate, supporting evidence of regional transformation.
- The results highlight the value of high-resolution, chemically speciated data for identifying and characterizing cross-border pollution events.



Time series of SO<sub>2</sub> hourly measurements made at 6 AURN sites in the UK and the two UK EMEP supersites measurements of SO<sub>2</sub> and PM<sub>10</sub>/2.5 SO<sub>4</sub><sup>2-</sup>. (NOTE: SO<sub>2</sub> at Auchencorth Moss is underestimated between 11:00 and 22:00 (GMT) on the 21/09/14)

# Metrohm 2060 MARGA Advantages



- 1 Measures gases and aerosols simultaneously
- 2 Captures hourly, time-resolved data
- 3 Fully automated with minimal manual handling
- 4 Remote access via cellular modem
- 5 Designed for real-world conditions

# Thank You



**Metrohm USA**

9250 Camden Field Parkway  
Riverview, FL 33578



**Email**

[kraig.kmiotek@metrohmusa.com](mailto:kraig.kmiotek@metrohmusa.com)



**LinkedIn**

<https://www.linkedin.com/in/kraigkmiotek/>