

Analysis of Per- and Polyfluoroalkyl Substances (PFAS) in Leachate by EPA Method 1633A Using LC-MS/MS

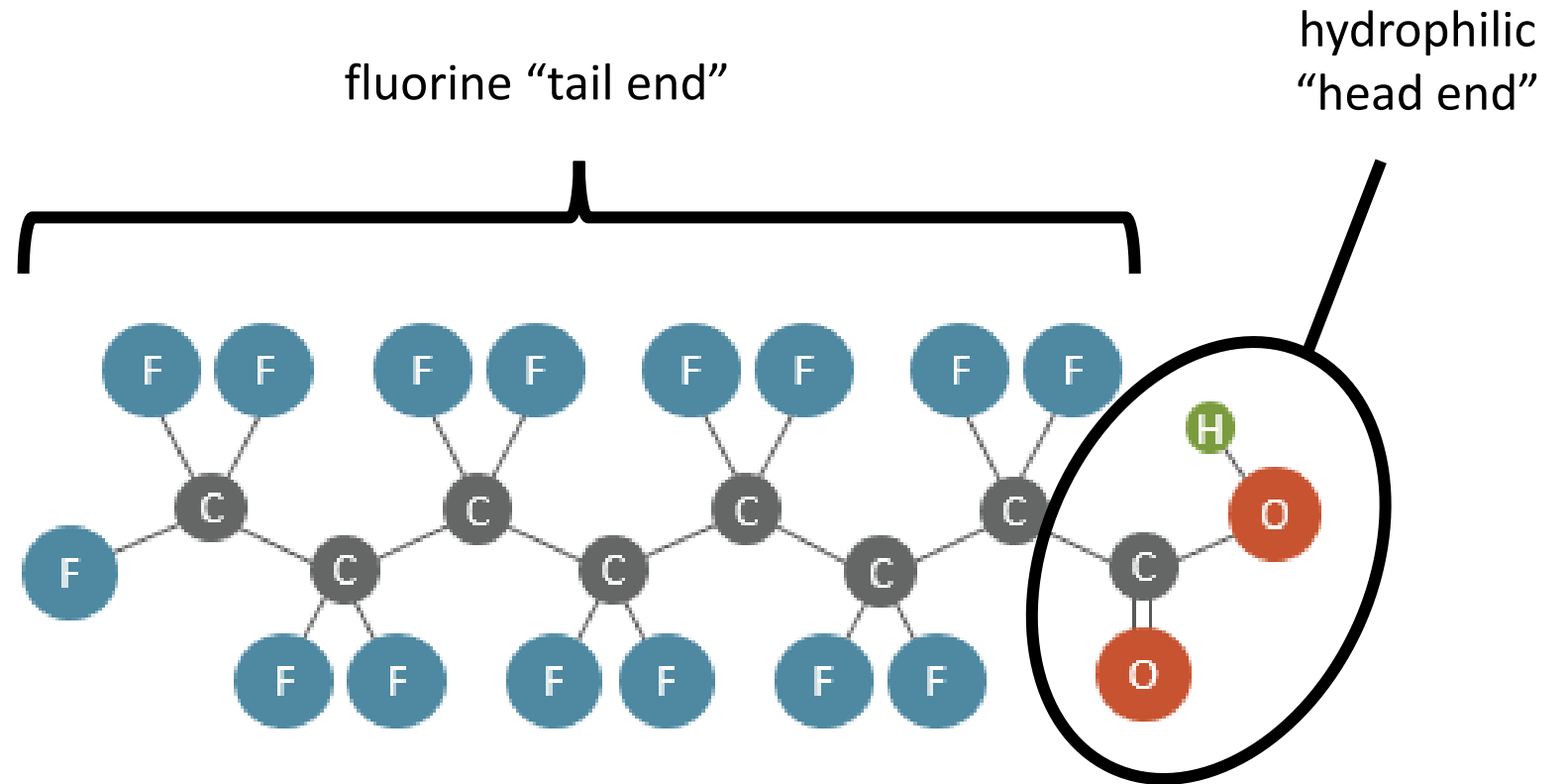
August 4, 2025

Tiffany Liden, PhD



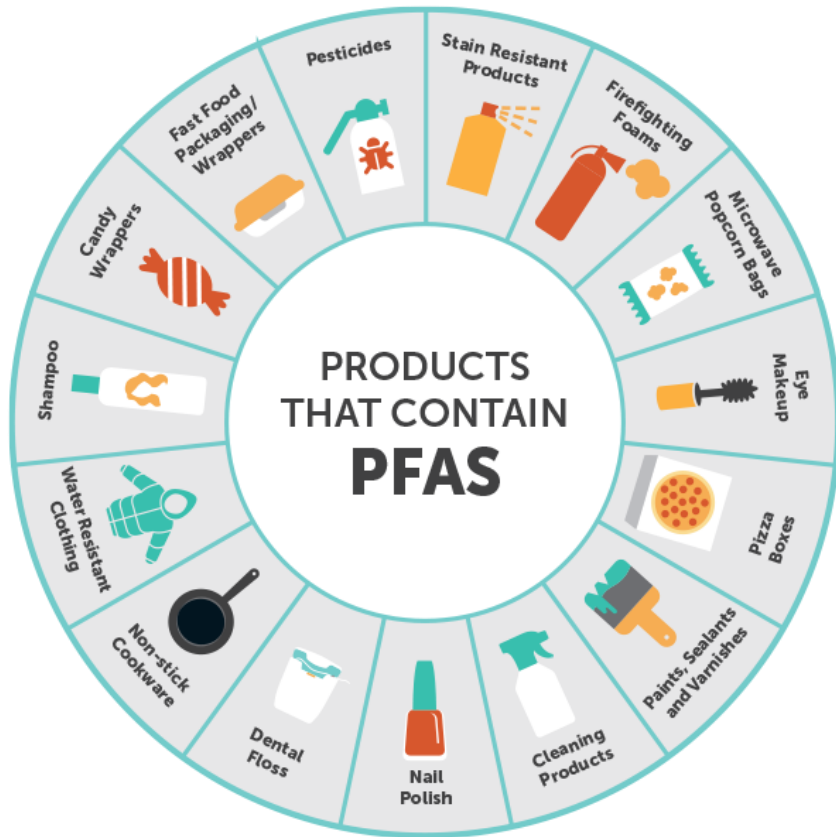
PFAS & Leachate

Per- and Polyfluoroalkyl
Substances (PFAS)

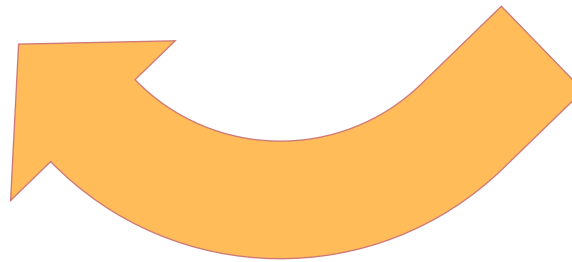
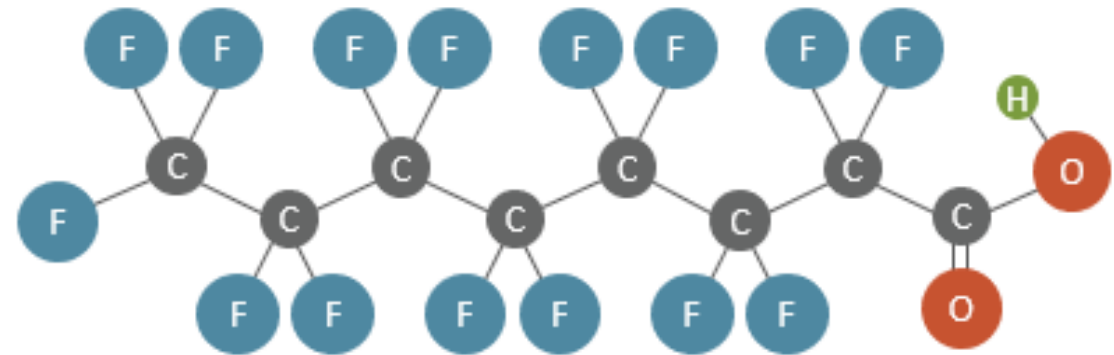


PFAS & Leachate

Per- and Polyfluoroalkyl Substances (PFAS)
are used to make household products.

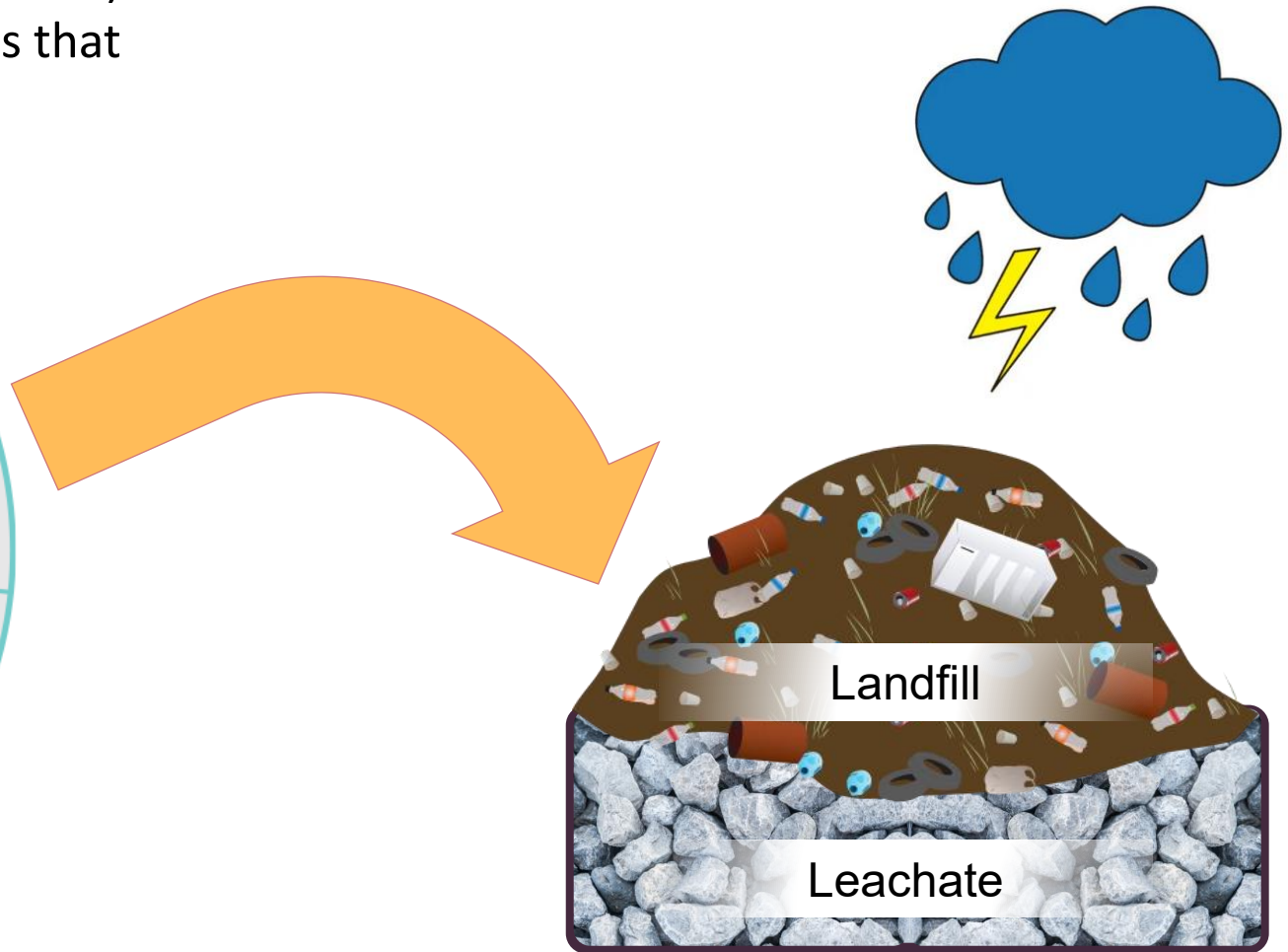
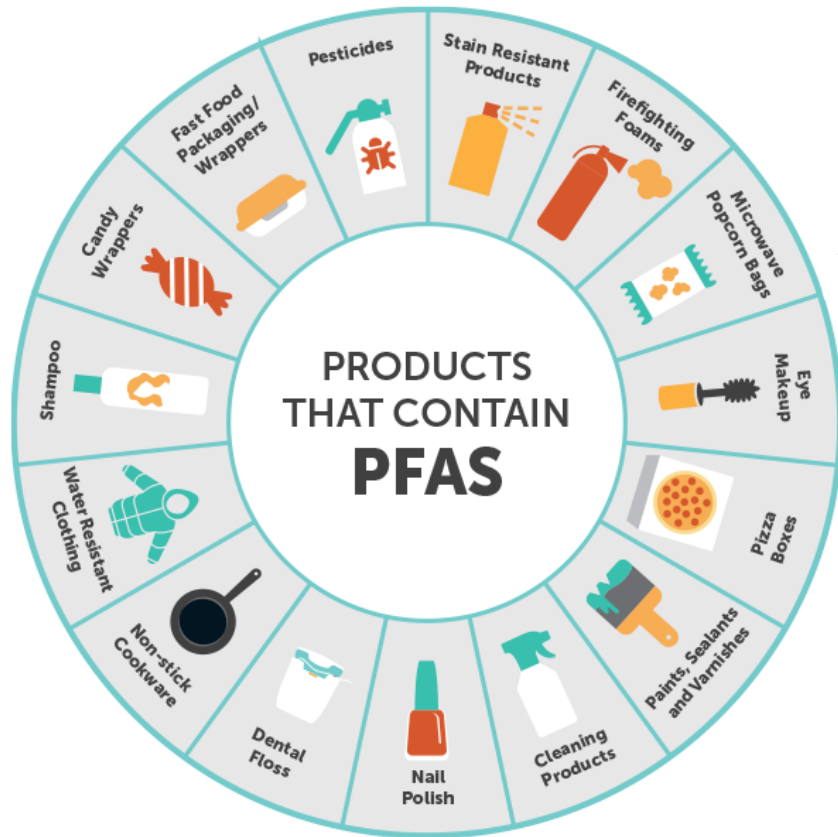


PWC Fayetteville's Home Town Utility, <https://www.faypwc.com/pfas-facts/>



PFAS & Leachate

Per- and Polyfluoroalkyl Substances (PFAS) are used to make household products that are discarded.



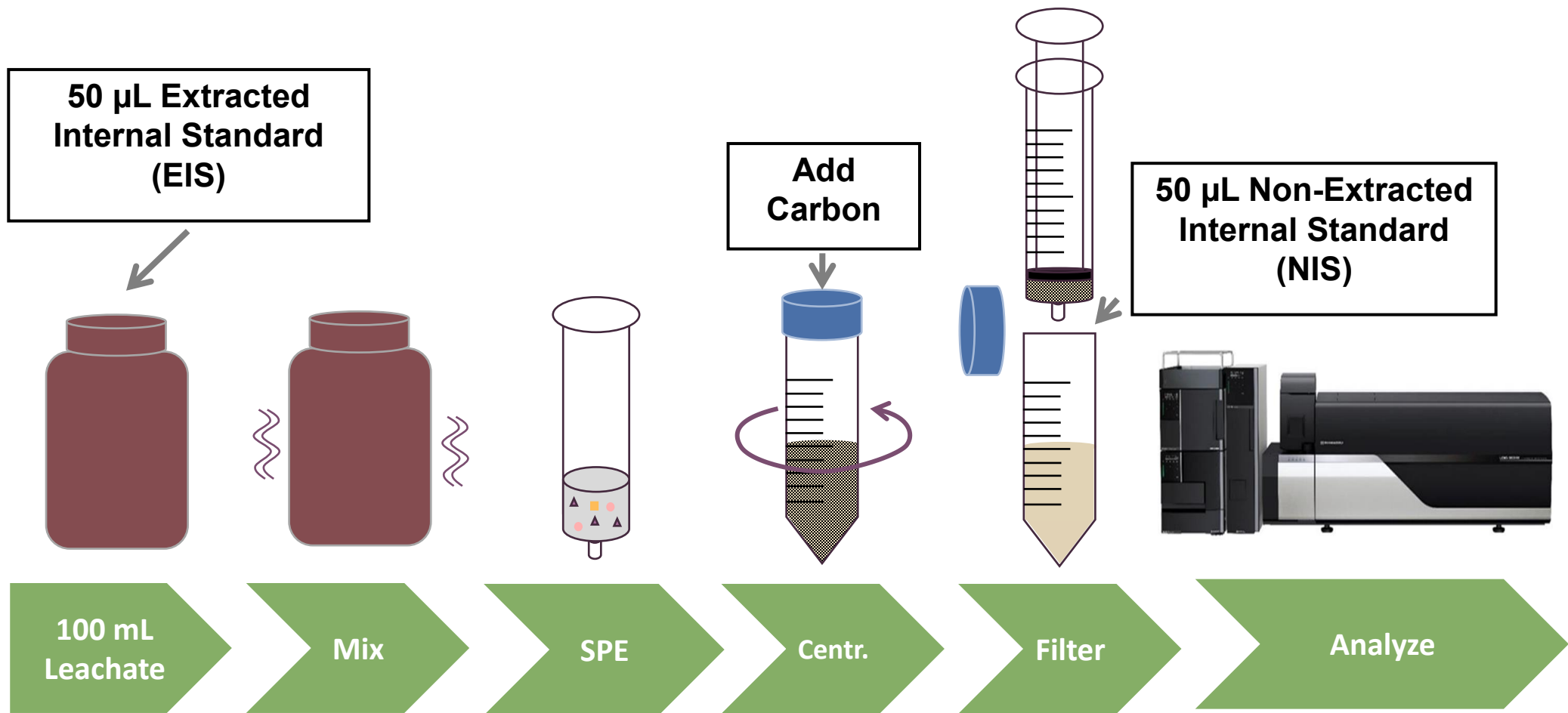
PWC Fayetteville's Home Town Utility, <https://www.faypwc.com/pfas-facts/>

Leachate Source



- Pooled Sample
- From the central-southern U.S.

EPA 1633A Sample Prep



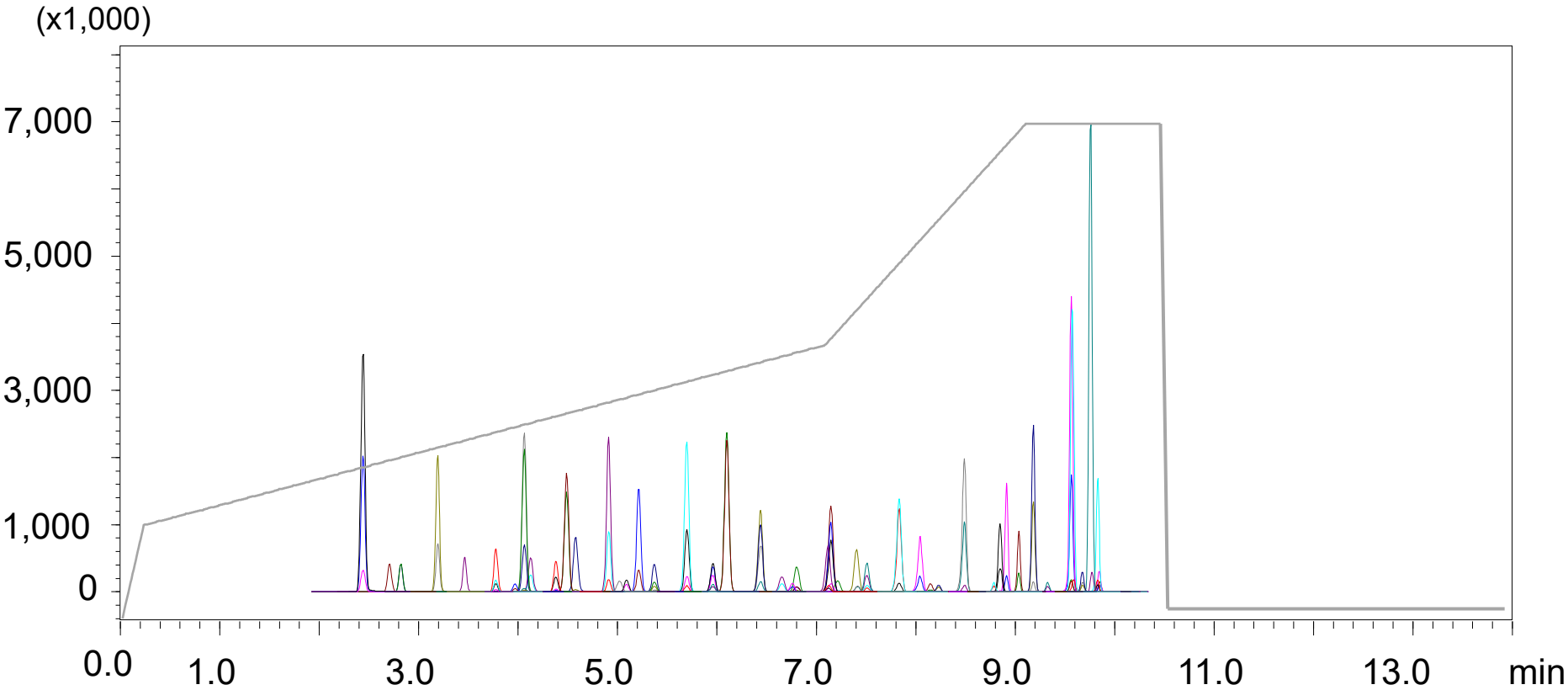
Instrument Parameters



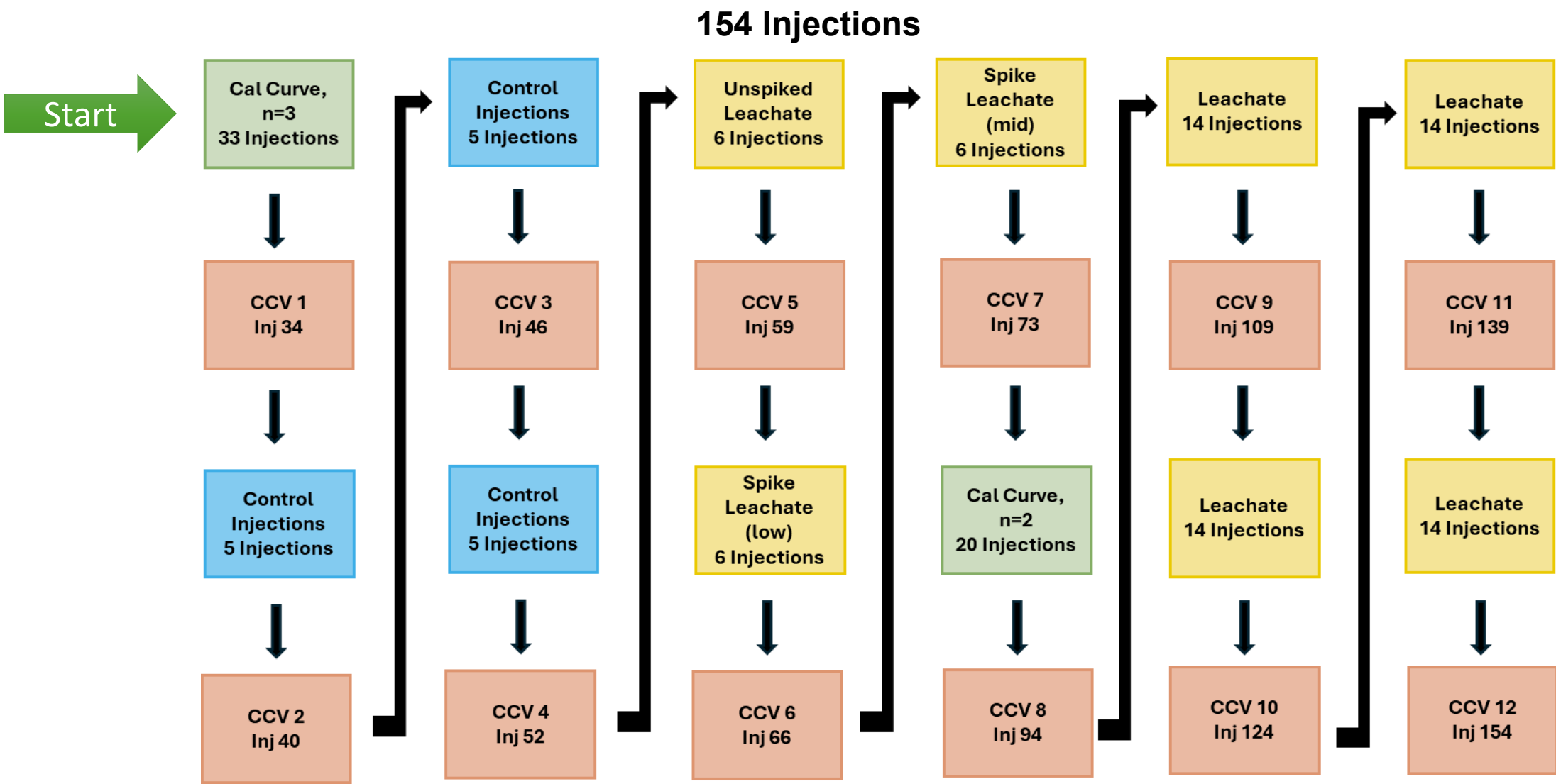
LCMS-8060	Parameters	LC	Parameters
Nebulizing gas	3.0 L/min	Delay Column	Nexcol PFAS Delay Column (50 x 3.0mm, 5 µm)
Heating gas	15.0 L/min	Analytical Column	Shim-pack Scepter C18-120 Column (50 × 2.1 mm, 3 µm)
Drying gas	5.0 L/min	Flow rate	0.4 mL/min
Interface Temperature	250 ° C	Mobile phase A	2mM Ammonium Acetate in Water
DL Temperature	200° C	Mobile phase B	Acetonitrile
Heat Block Temperature	300° C	Column oven temperature	40 ° C
Diluent	Methanol with 4% water, 1% ammonium hydroxide and 0.625% acetic acid		

PFAS- EPA 1633A

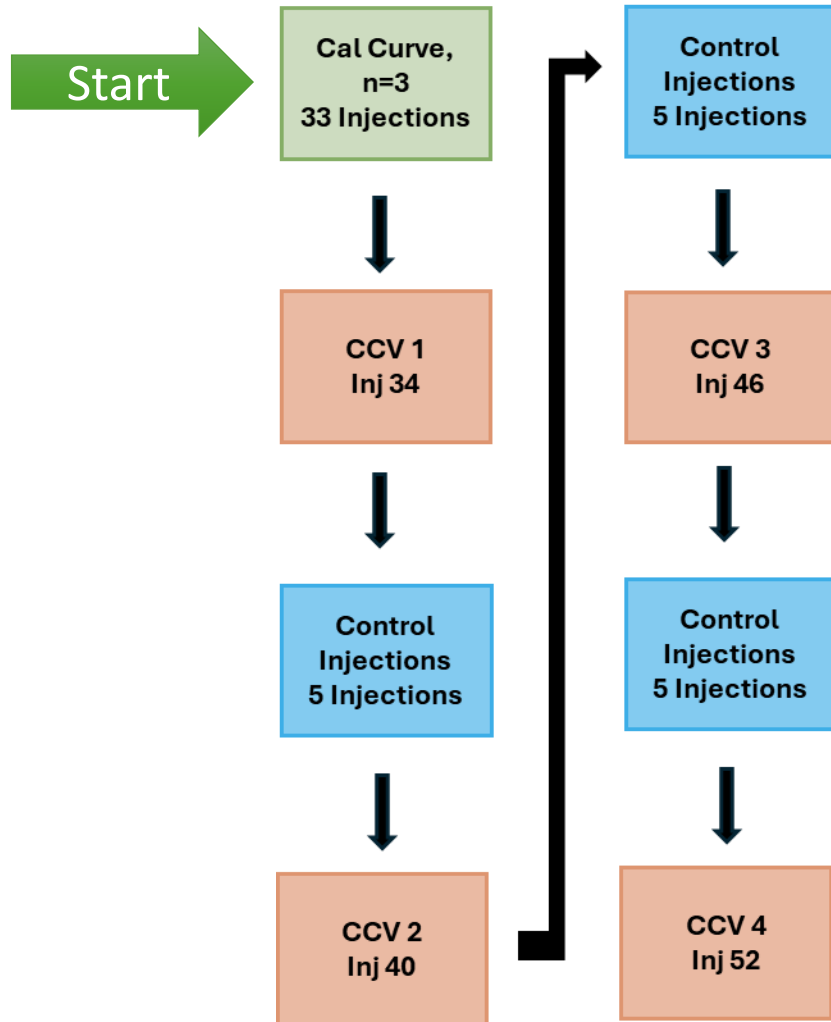
LC Gradient	
Time (min)	%B
0	2
0.20	20
7.00	55
9.00	98
10.00	98
10.01	2
14.00	Stop



Injection Sequence



Injection Sequence



Calibration Curve

- Based on EPA 1633A ranges
- Repeated 3X included blanks

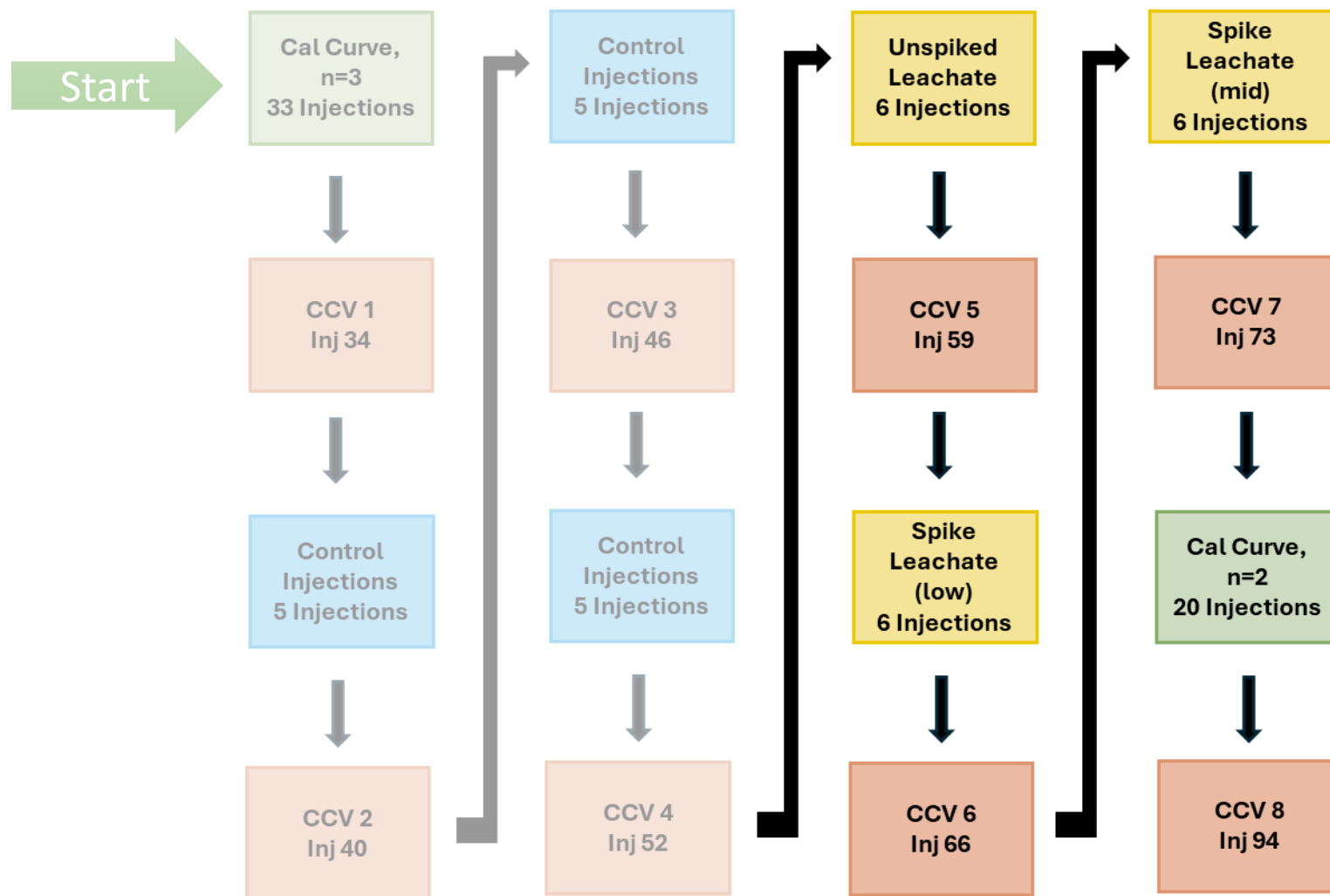
Continuous Calibration Verification (CCV)

- Equivalent to Cal 4

Extracted Lab Controls

- Repeated 3x
- Consisted of
 - Method Blank
 - 2 LLC (Spiked 2x Cal 1)
 - 2 MLC (Spiked Cal 4)

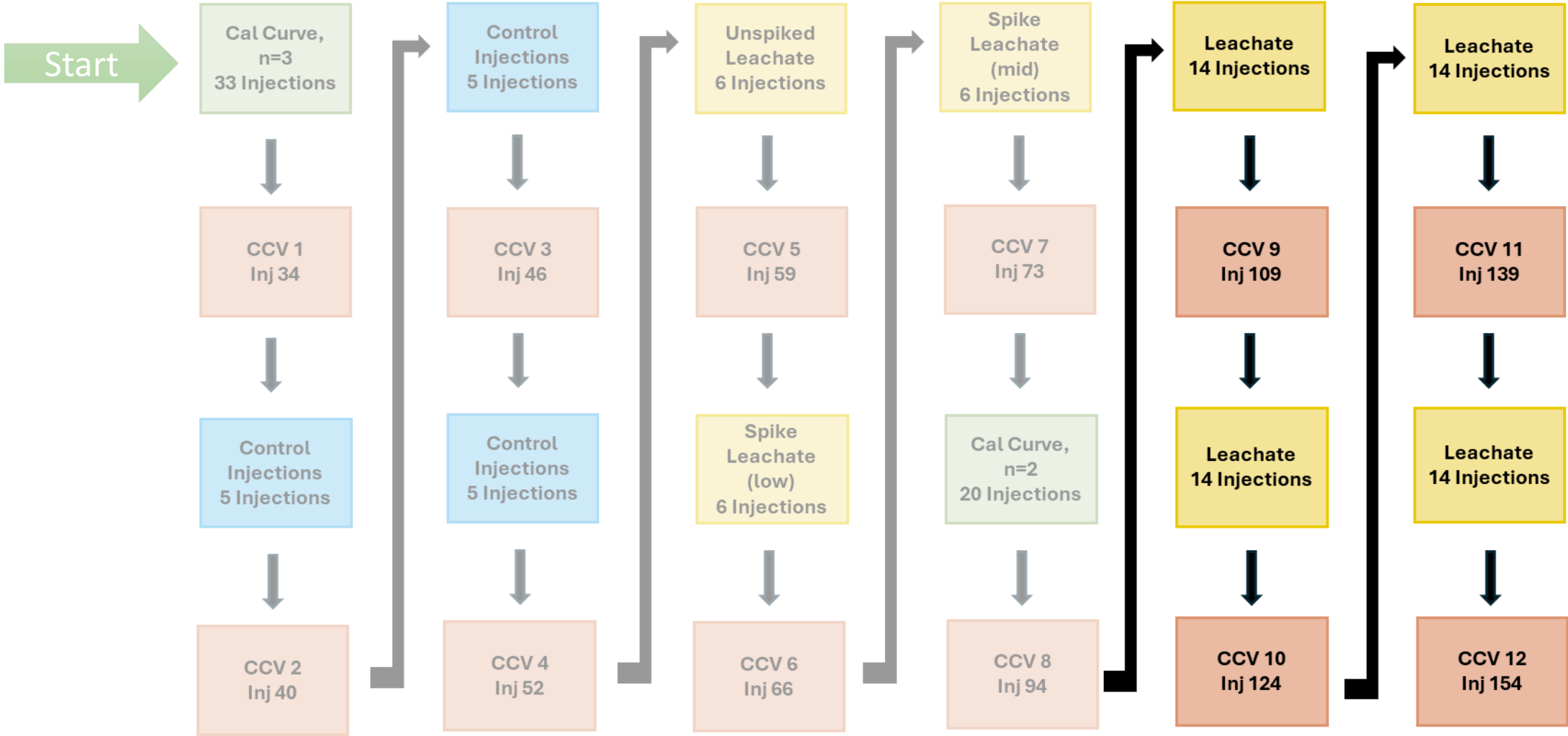
Injection Sequence



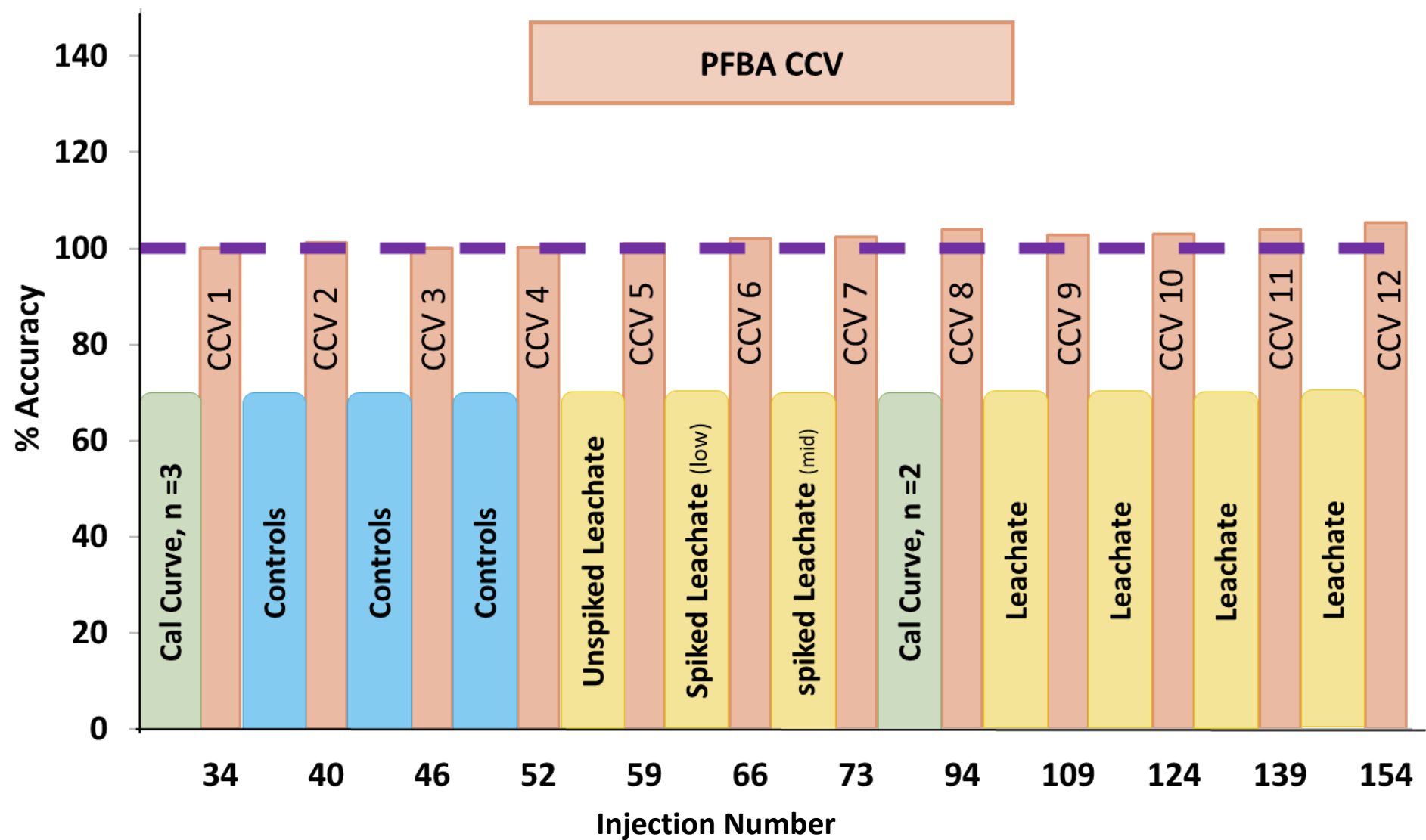
Leachate

- Consisted of
 - Unspiked
 - Spiked Low (Spiked 2x Cal 1)
 - Spiked Mid (Spiked Cal 4)

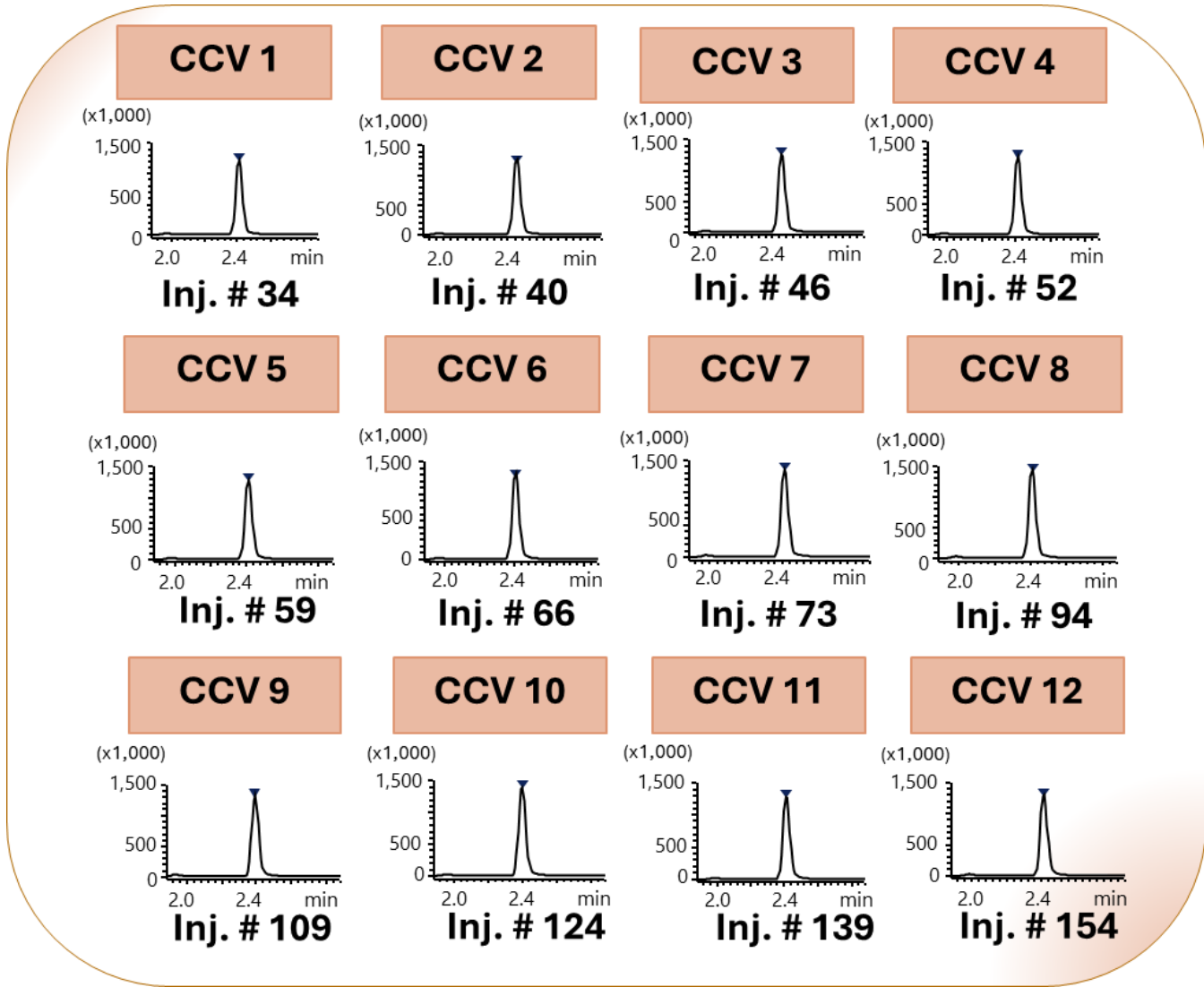
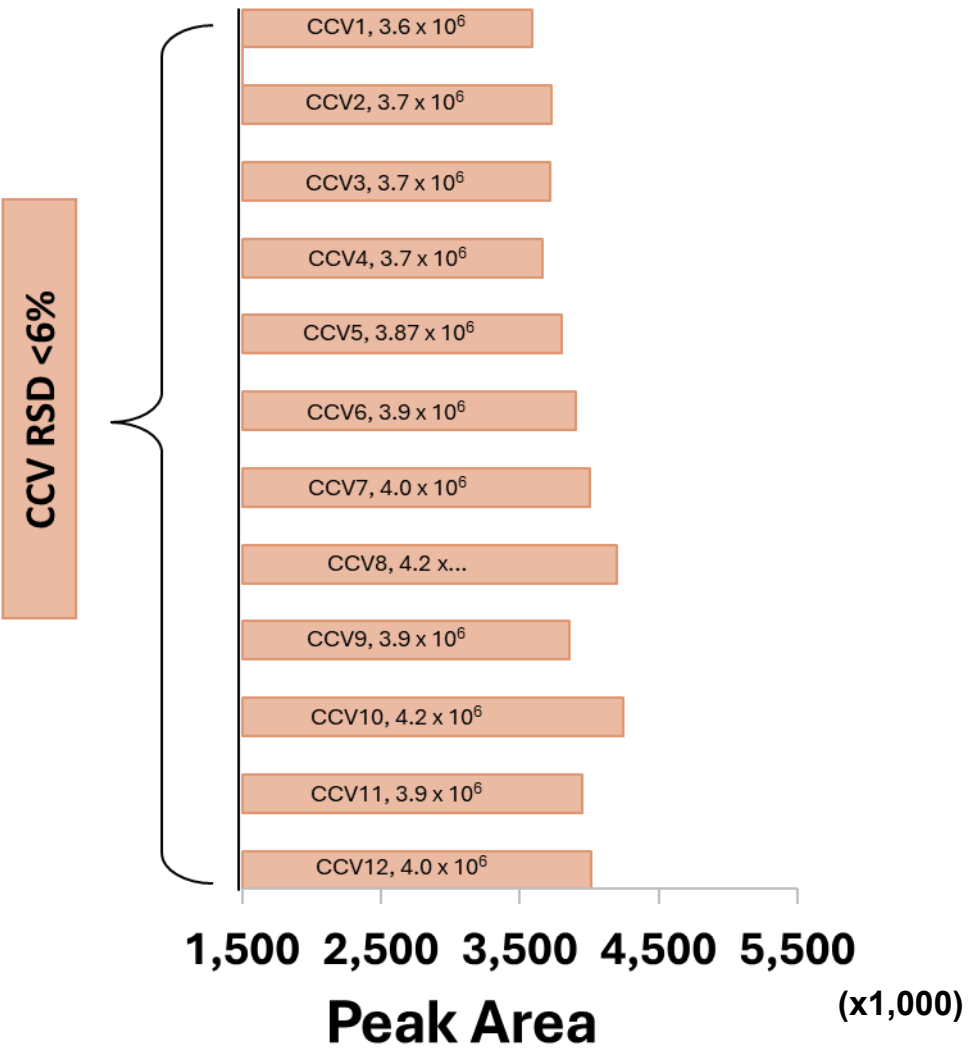
Injection Sequence



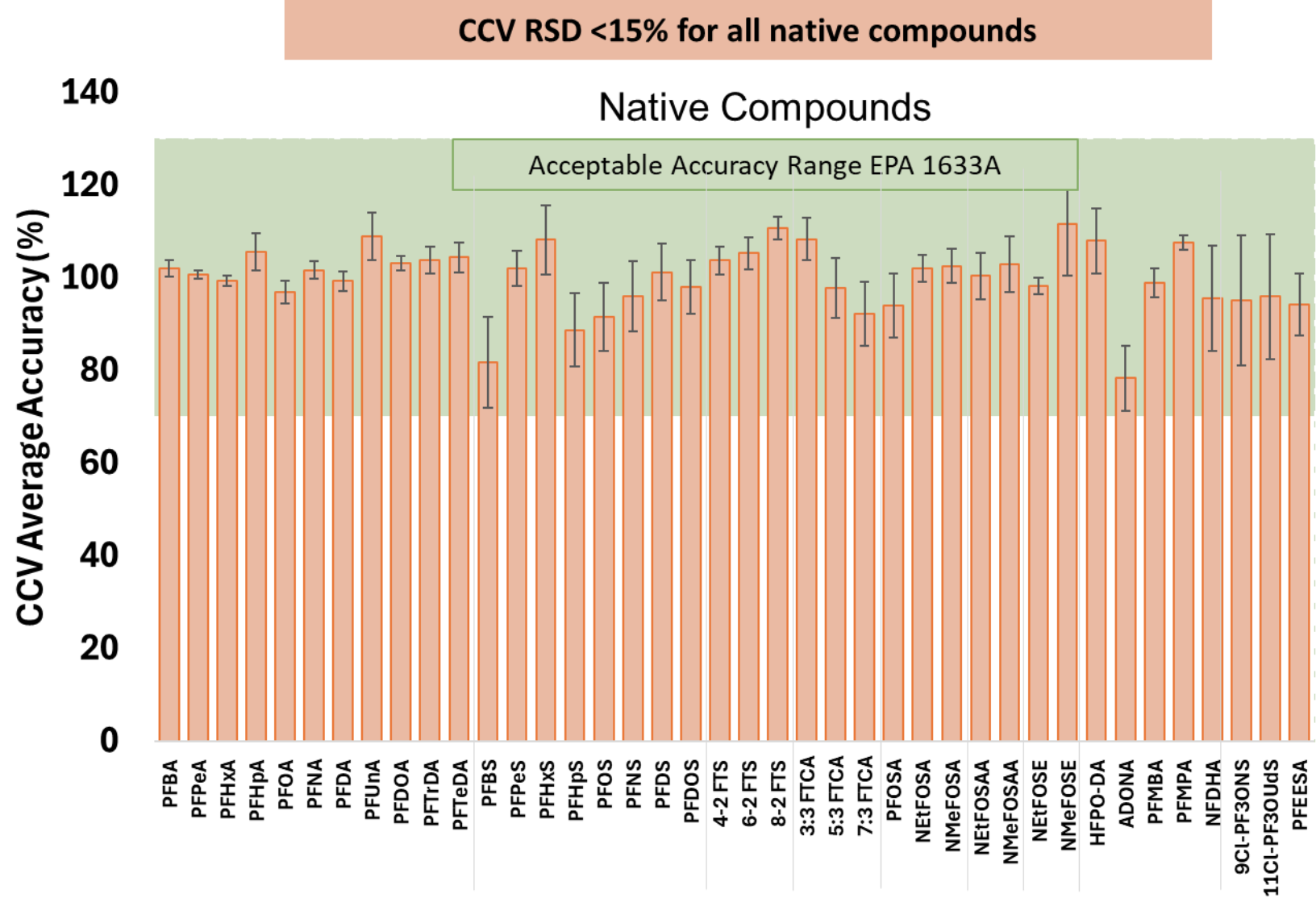
Accuracy-PFBA CCV



Reproducibility-PFBA



CCV Accuracy



EPA 1633A Acceptable Limits



Office of Water

www.epa.gov

December 2024

Method 1633, Revision A

Analysis of Per- and Polyfluoroalkyl Substances (PFAS) in Aqueous, Solid, Biosolids, and Tissue Samples by LC-MS/MS

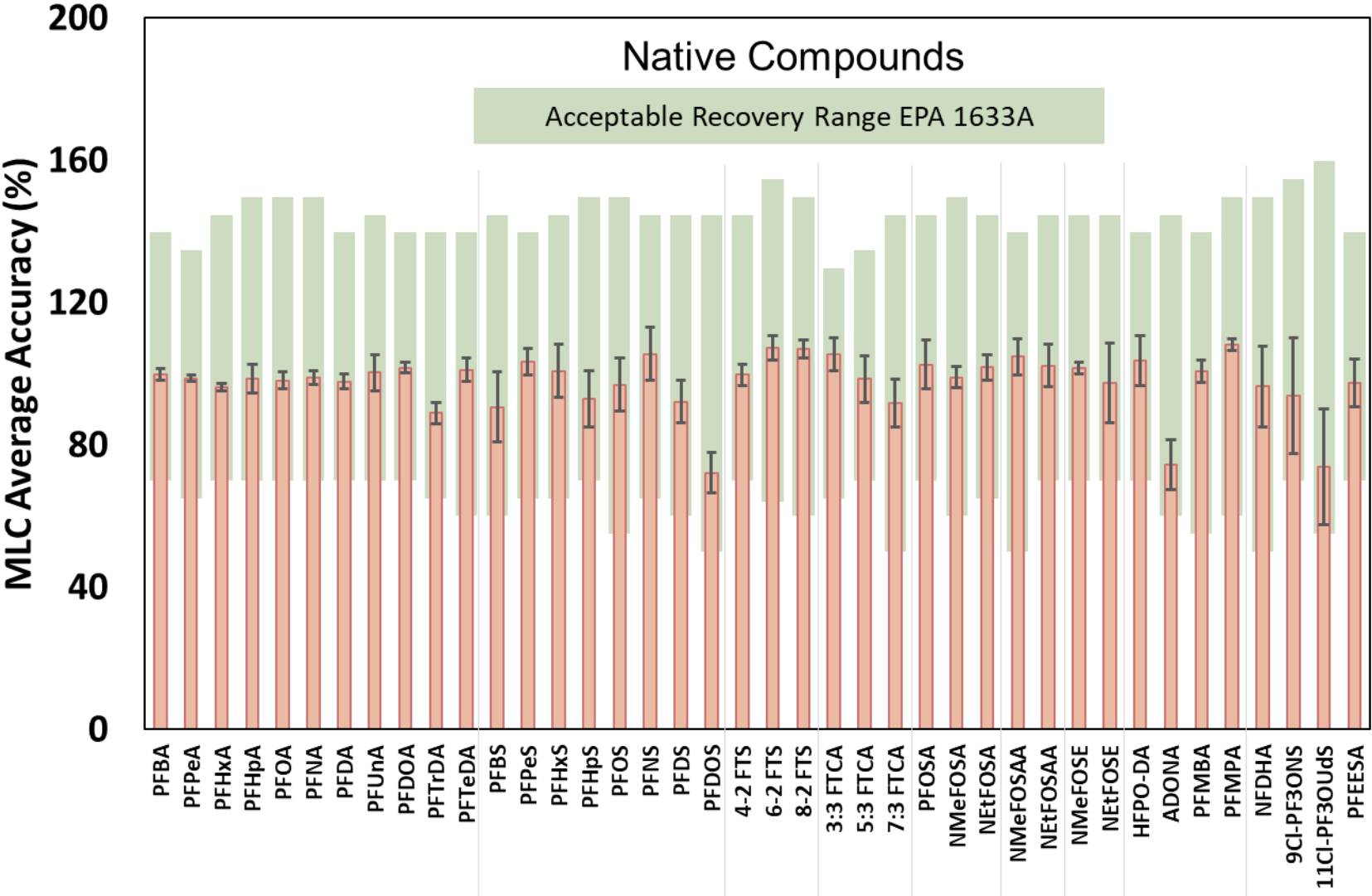
Table 5. IPR/OPR/LLOPR Acceptance Limits for Target Analytes in Aqueous Matrices¹

Target Analyte	IPR ²		OPR/LLOPR Recovery (%) ²
	Mean Recovery (%)	RSD (%)	
PFBA	70 - 135	21	70 - 140
PFPeA	70 - 135	23	65 - 135
PFHxA	70 - 135	24	70 - 145
PFHpA	70 - 135	28	70 - 150
PFOA	65 - 155		
PFNA	70 - 140		
PFDA	65 - 140		
PFUnA	70 - 135		
PFDoA	70 - 130		
PFTTrDA	60 - 145		

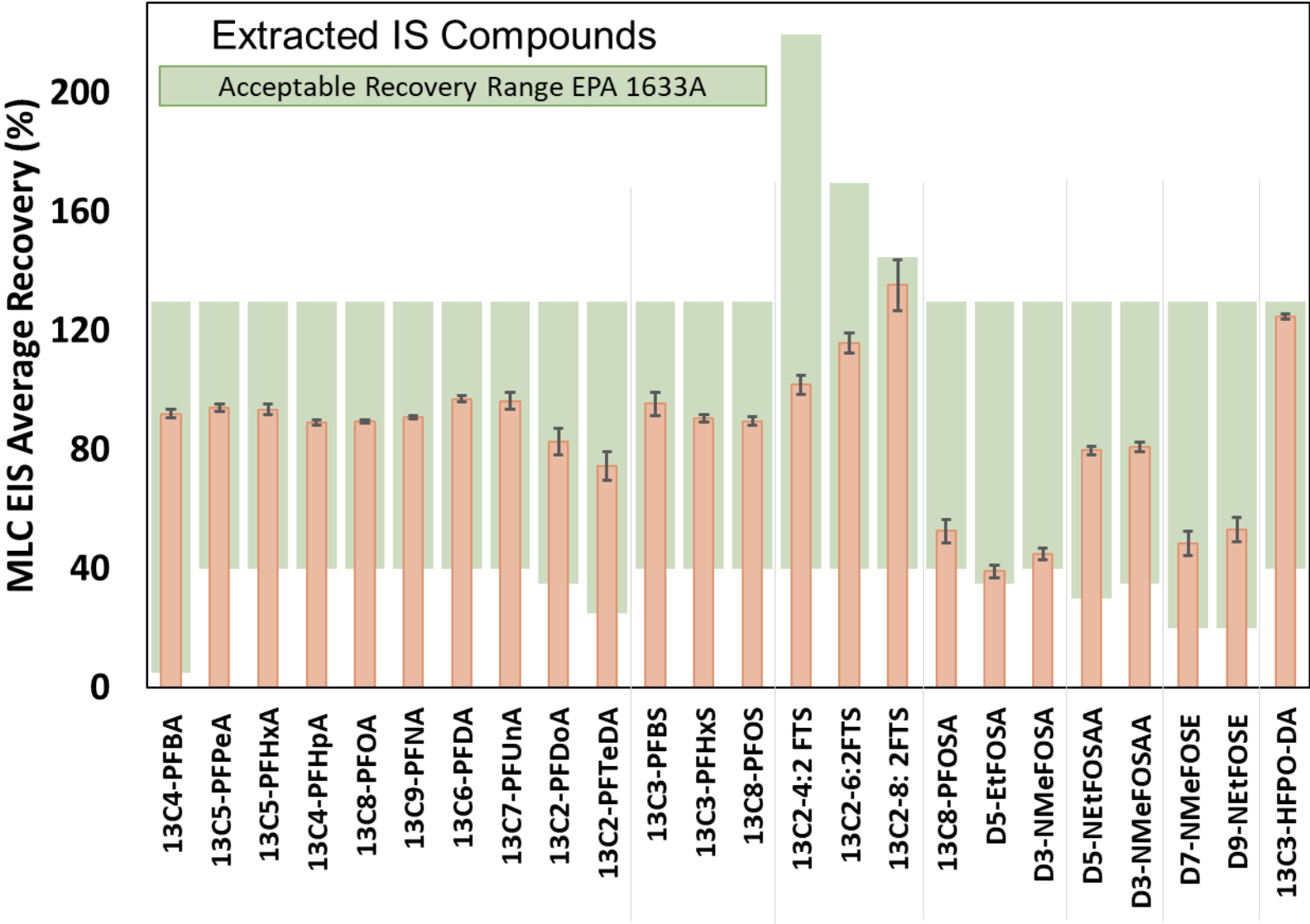
Table 6. Acceptance Limits for EIS and NIS Compounds in All Aqueous Matrices and Associated QC Samples

EIS Compound ¹	Aqueous Matrices Other Than Leachates ²	Landfill Leachates ³
	Recovery (%)	Recovery (%)
¹³ C ₄ -PFBA	5 ⁴ - 130	5 ⁴ - 130
¹³ C ₅ -PFPeA	40 - 130	40 - 130
¹³ C ₅ -PFHxA	40 - 130	40 - 130
¹³ C ₄ -PFHpA	40 - 130	40 - 130
¹³ C ₈ -PFOA	40 - 130	40 - 130
¹³ C ₉ -PFNA	40 - 130	40 - 130
¹³ C ₆ -PFDA	40 - 130	40 - 130
¹³ C ₇ -PFUnA	30 - 130	40 - 130
¹³ C ₇ -PFDoA	10 - 130	35 - 130

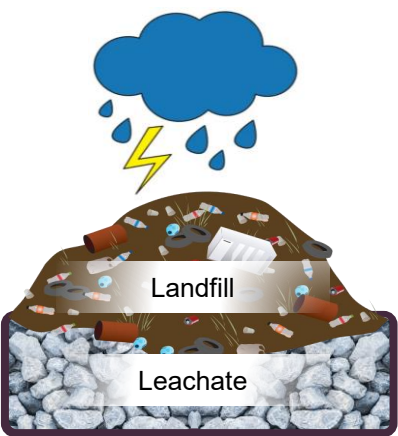
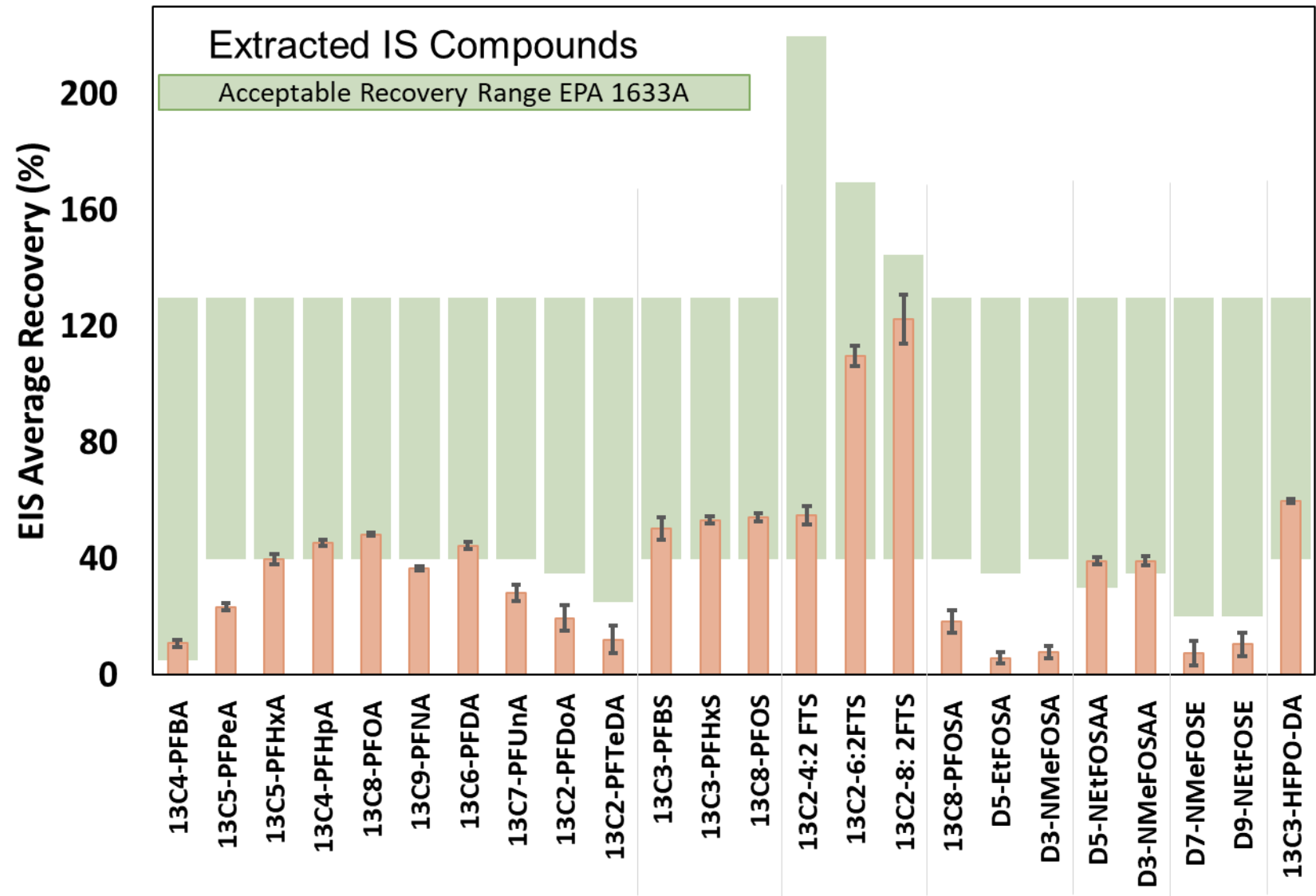
Extracted Lab Controls Native Recovery



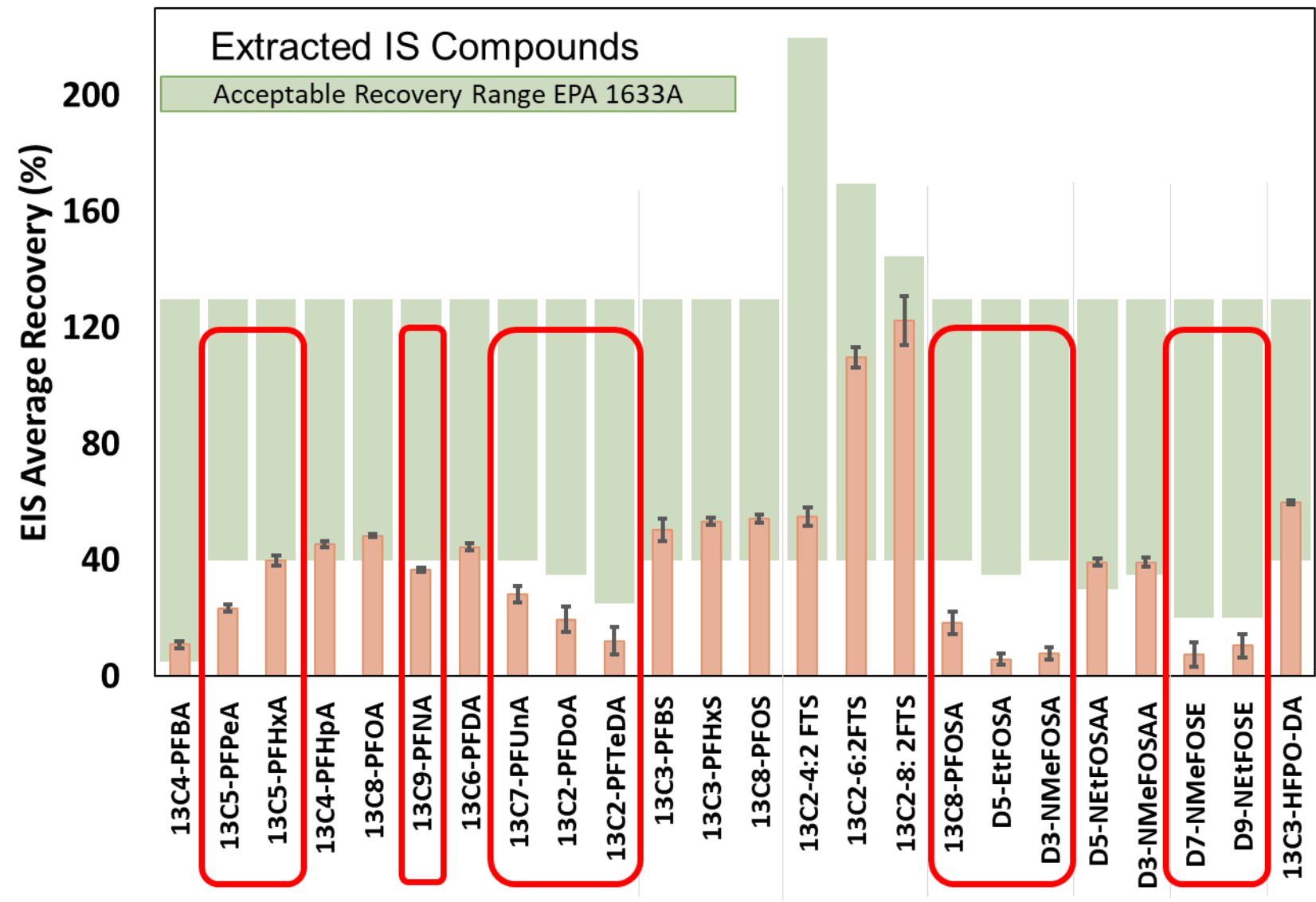
Extracted Lab Controls EIS Recovery



Leachate EIS Recovery

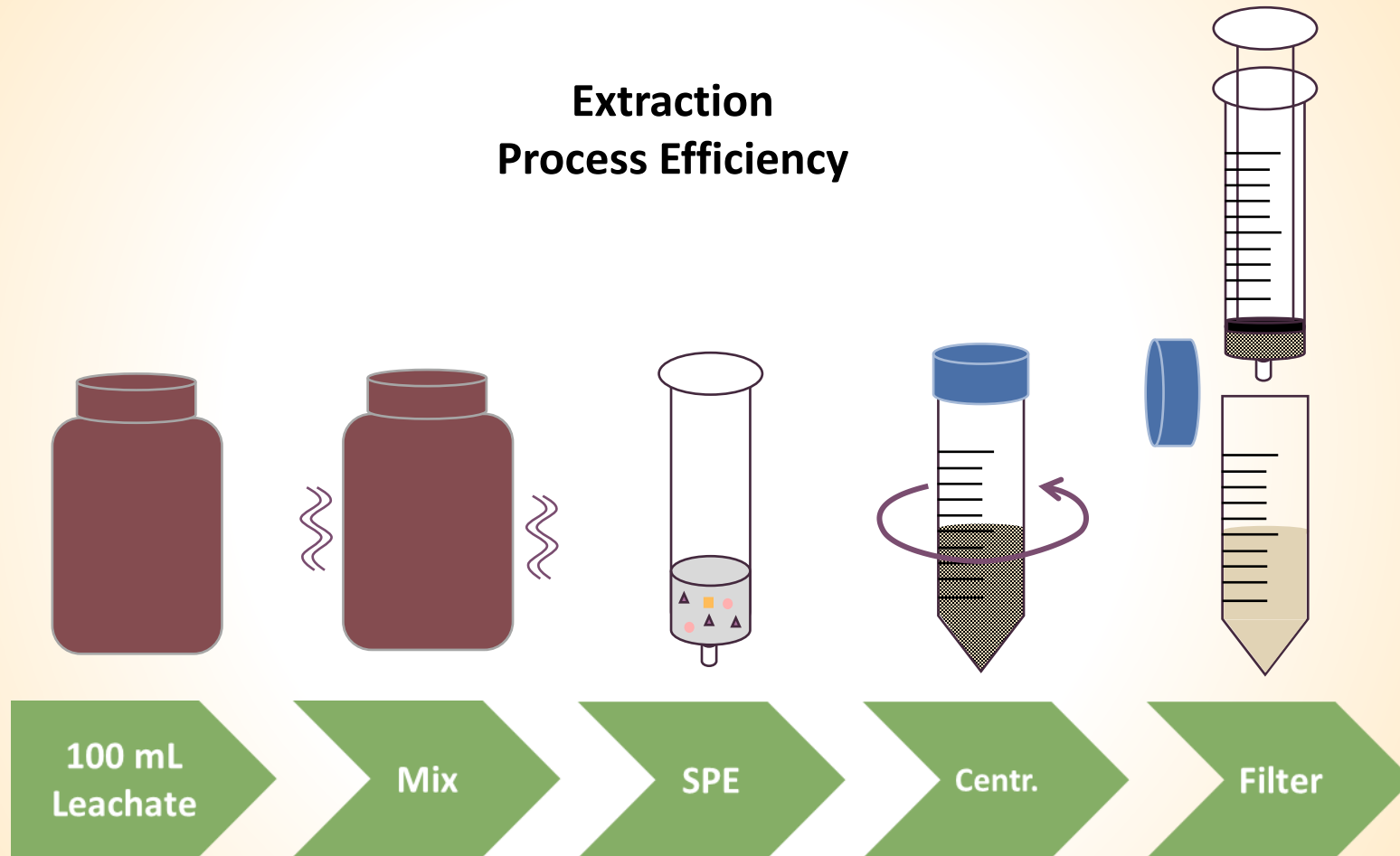


Leachate EIS Recovery



Complex Matrix Challenges

Extraction Process Efficiency

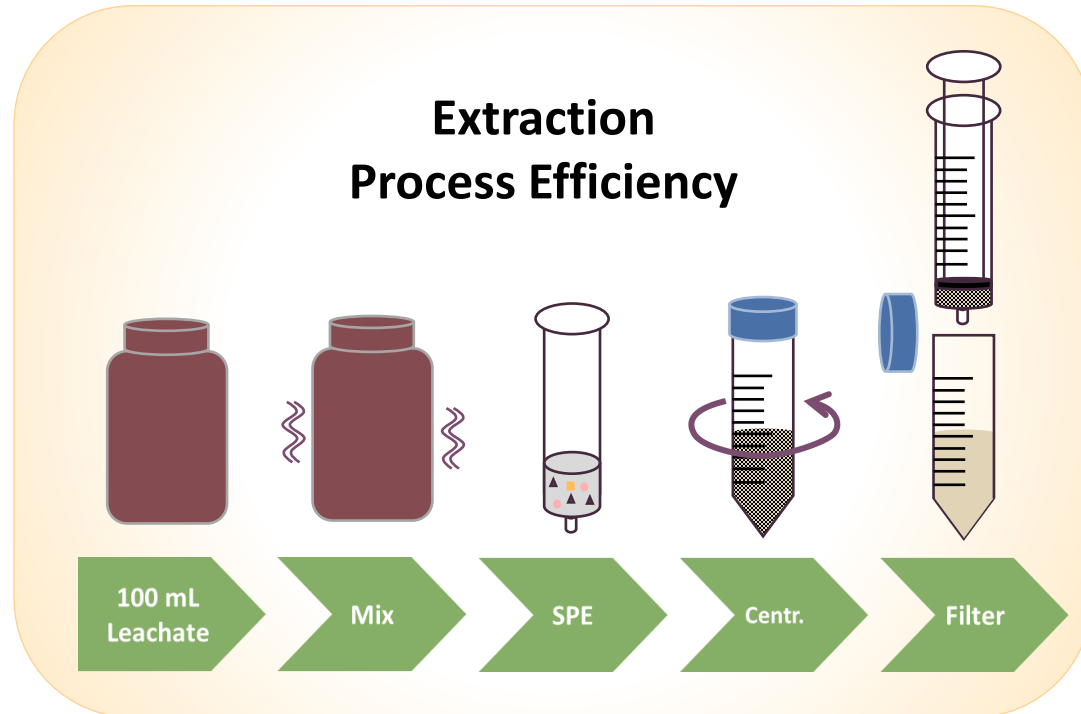


Instrument Matrix Tolerance



Analyze

Complex Matrix Challenges



Matrix Effects

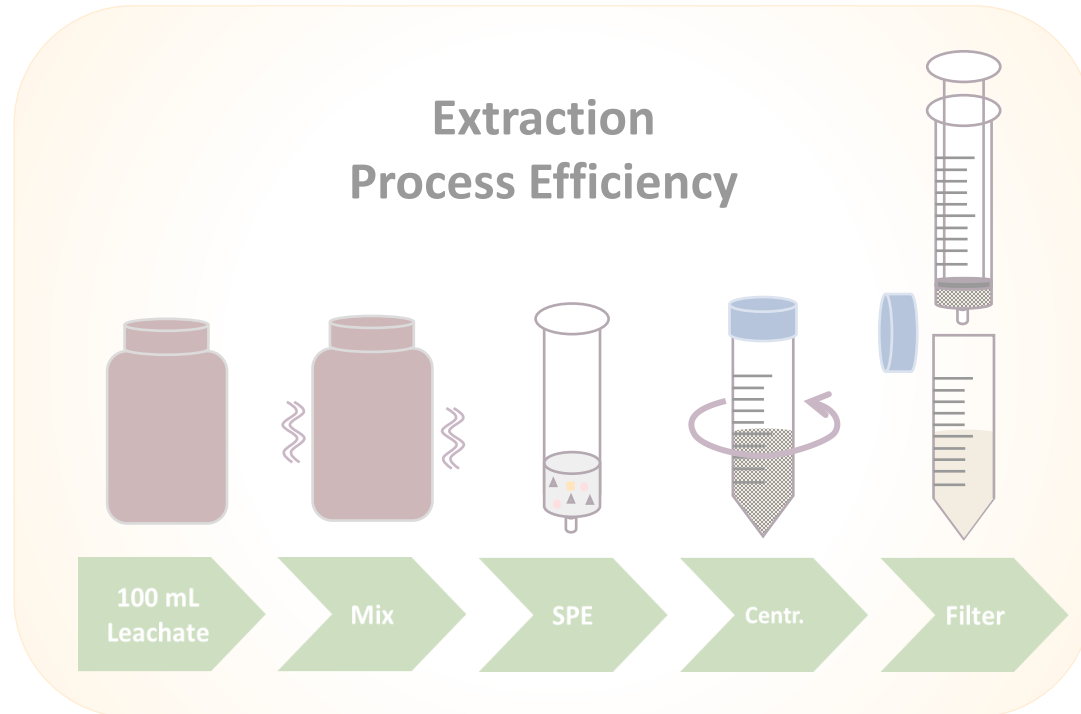
- Co-extraction of interfering compounds
- Sorbent overload or breakthrough
- Incomplete washing



Matrix Effects

- Matrix (Ionization) Robustness/Resilience
- Matrix Compatibility
- Source saturation
 - Ion suppression/enhancement

Complex Matrix Challenges



Matrix Effects

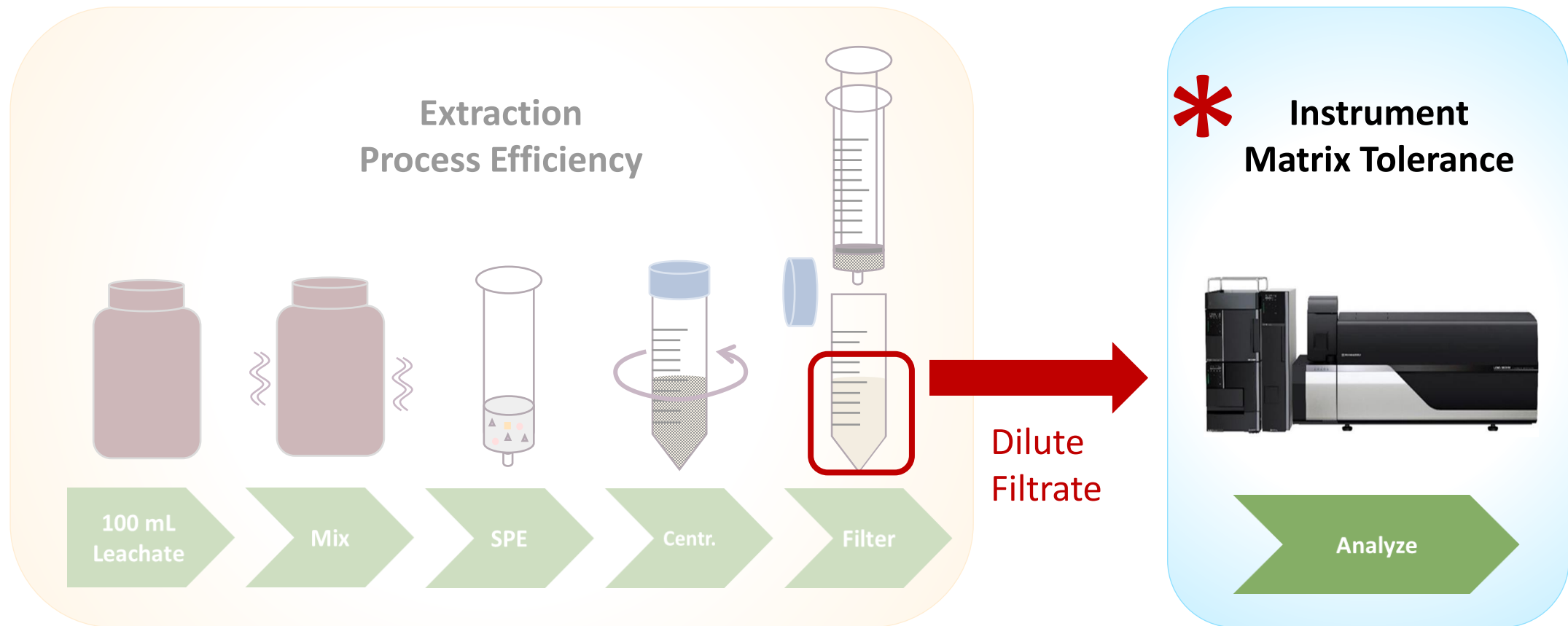
- Co-extraction of interfering compounds
- Sorbent overload or breakthrough
- Incomplete washing



Matrix Effects

- Matrix (Ionization) Robustness/Resilience
- Matrix Compatibility
- Source saturation
 - Ion suppression/enhancement

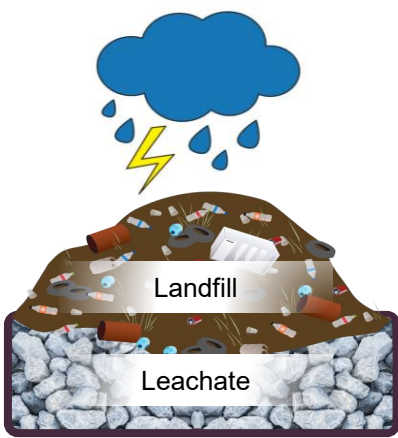
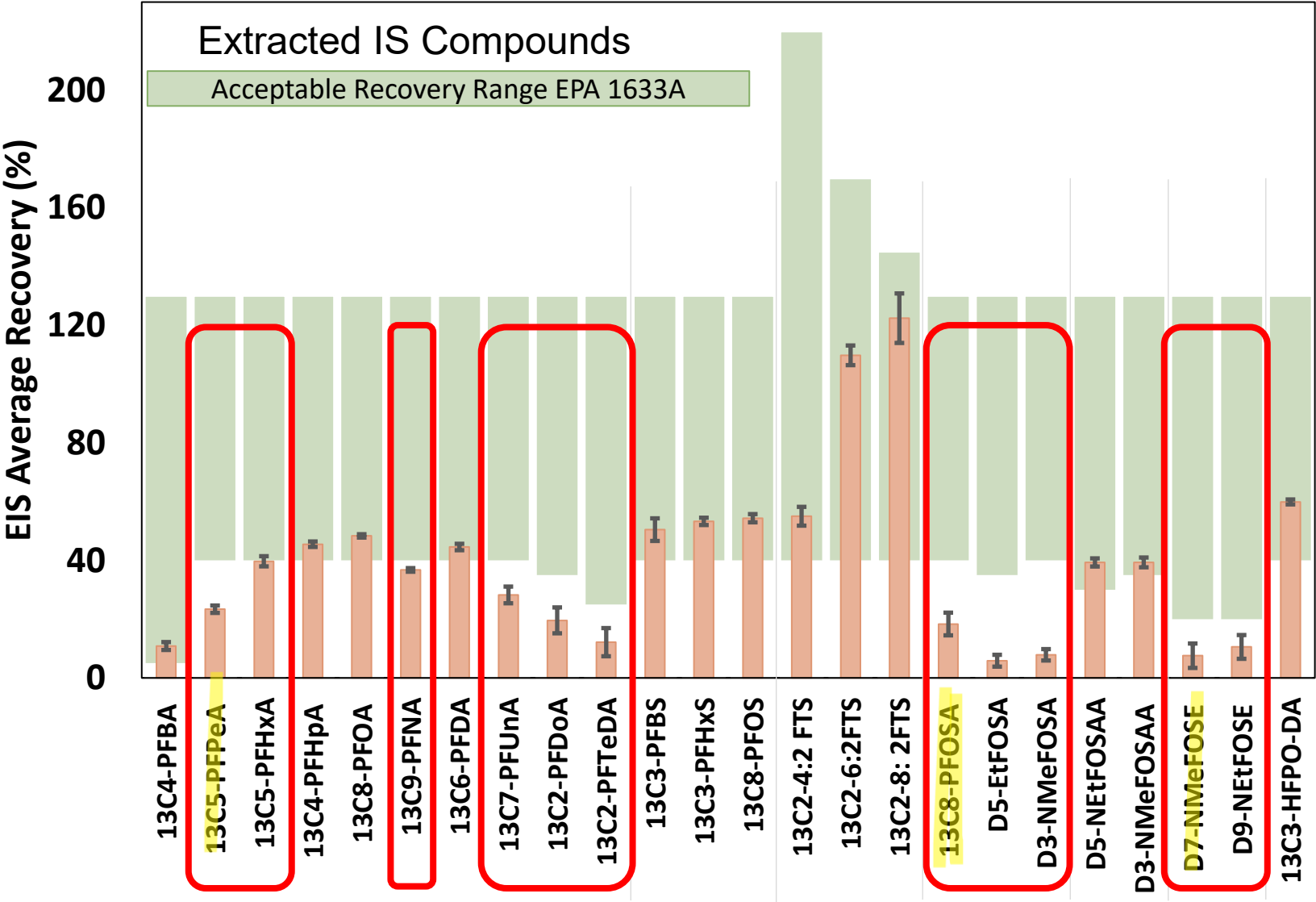
Post-Extraction Dilution Process



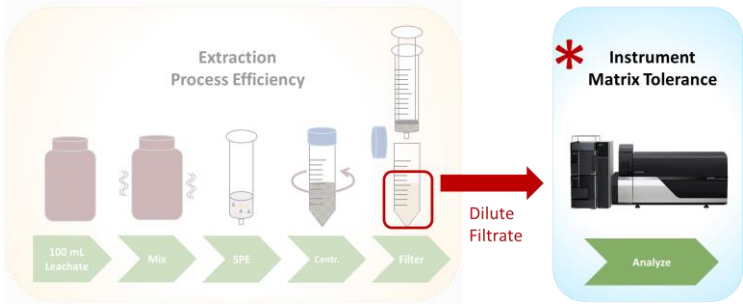
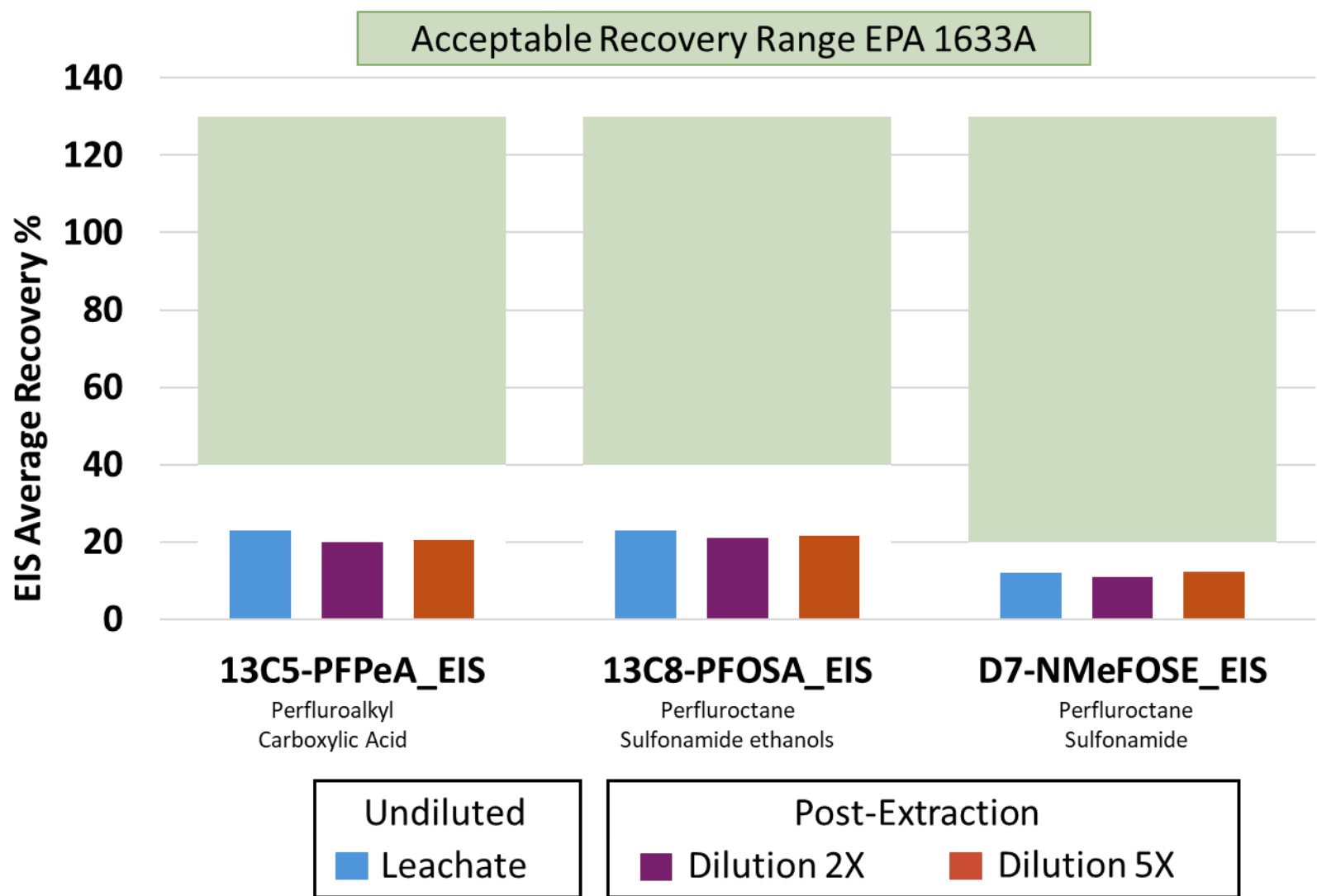
Instrument Matrix Tolerance: Evaluated by dilution of post-extraction filtrate.

- 2x & 5x dilution

Leachate EIS Recovery

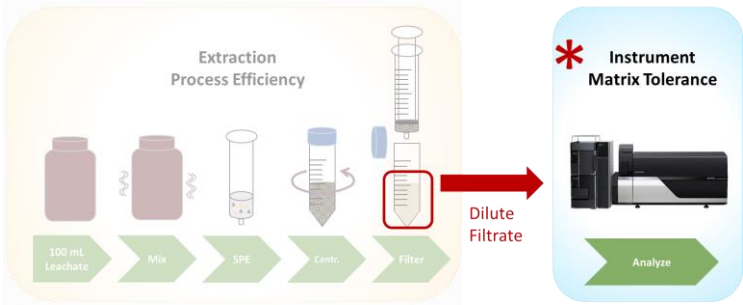
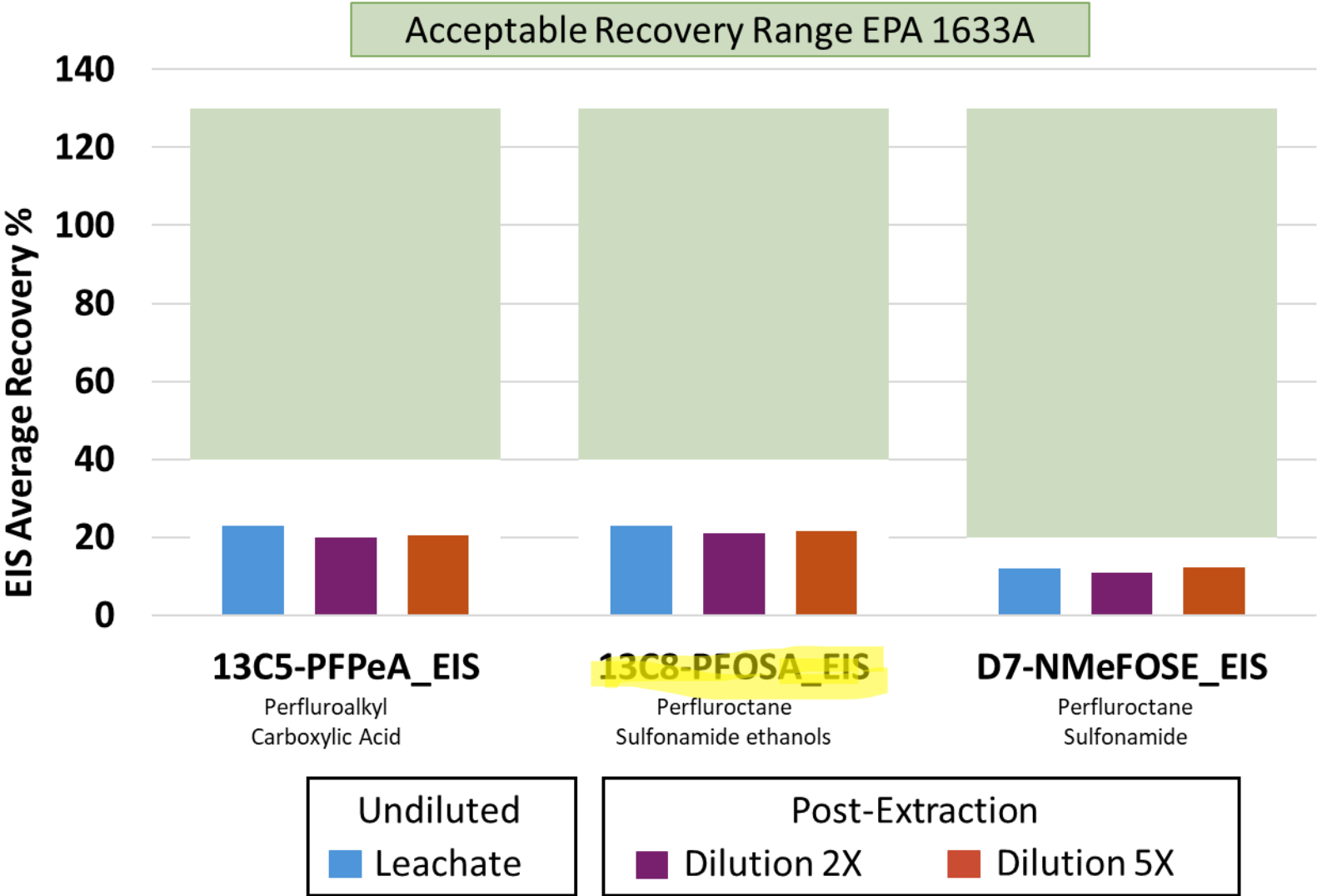


Post-Extraction Dilution EIS Impact



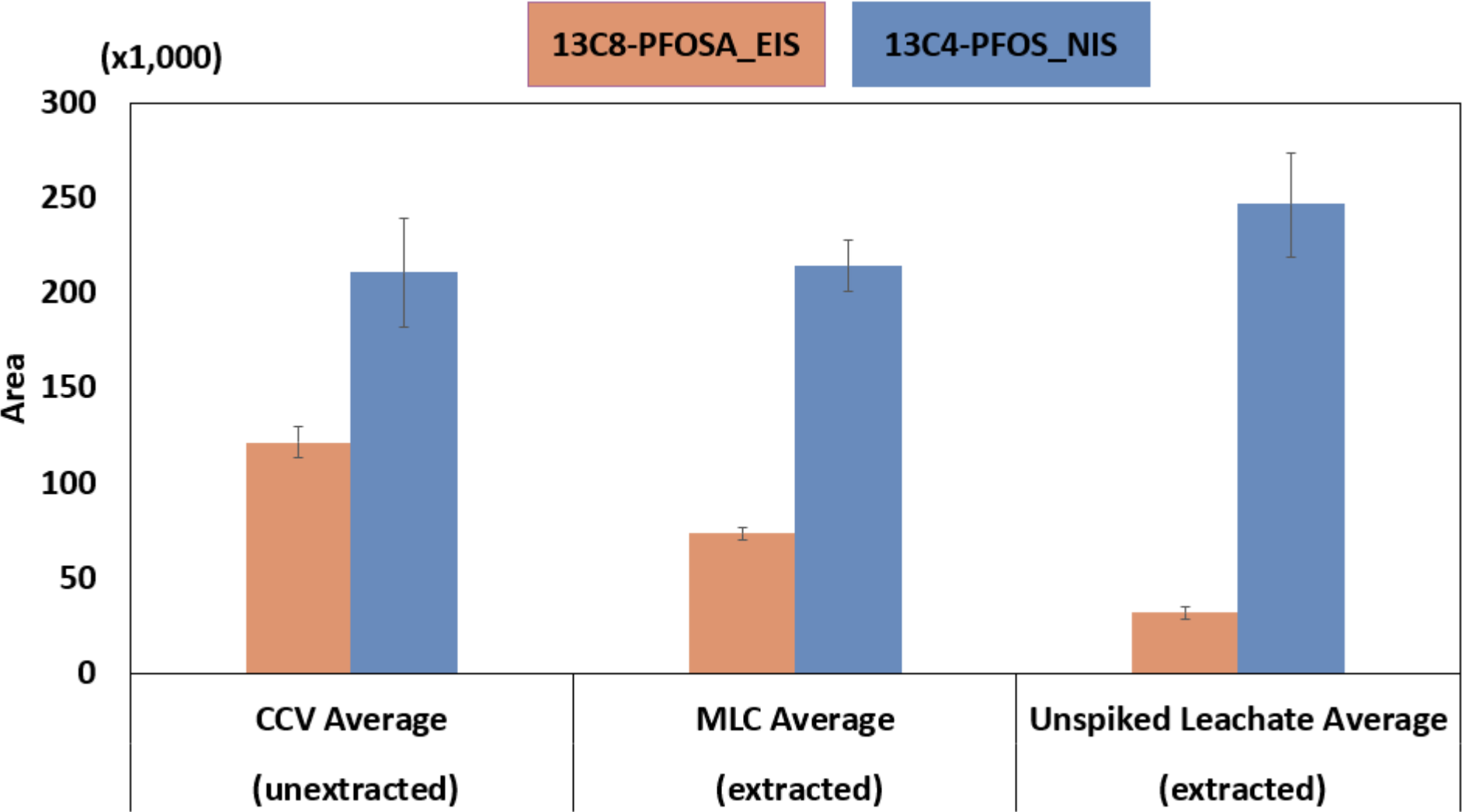
- Instrument Matrix Tolerance is **not** the issue.

Post-Extraction Dilution EIS Impact



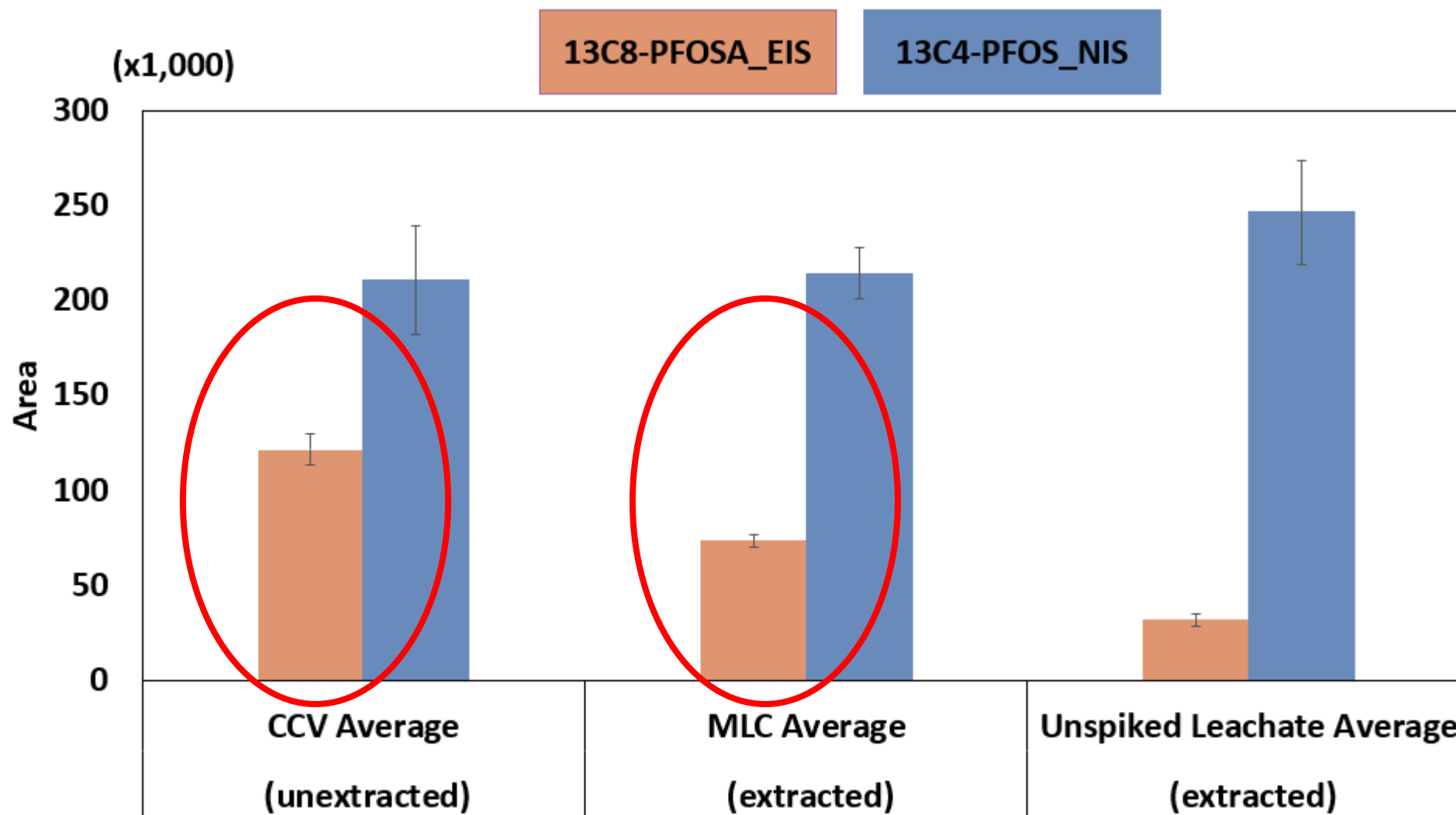
- Instrument Matrix Tolerance is **not** the issue.

Area EIS vs NIS



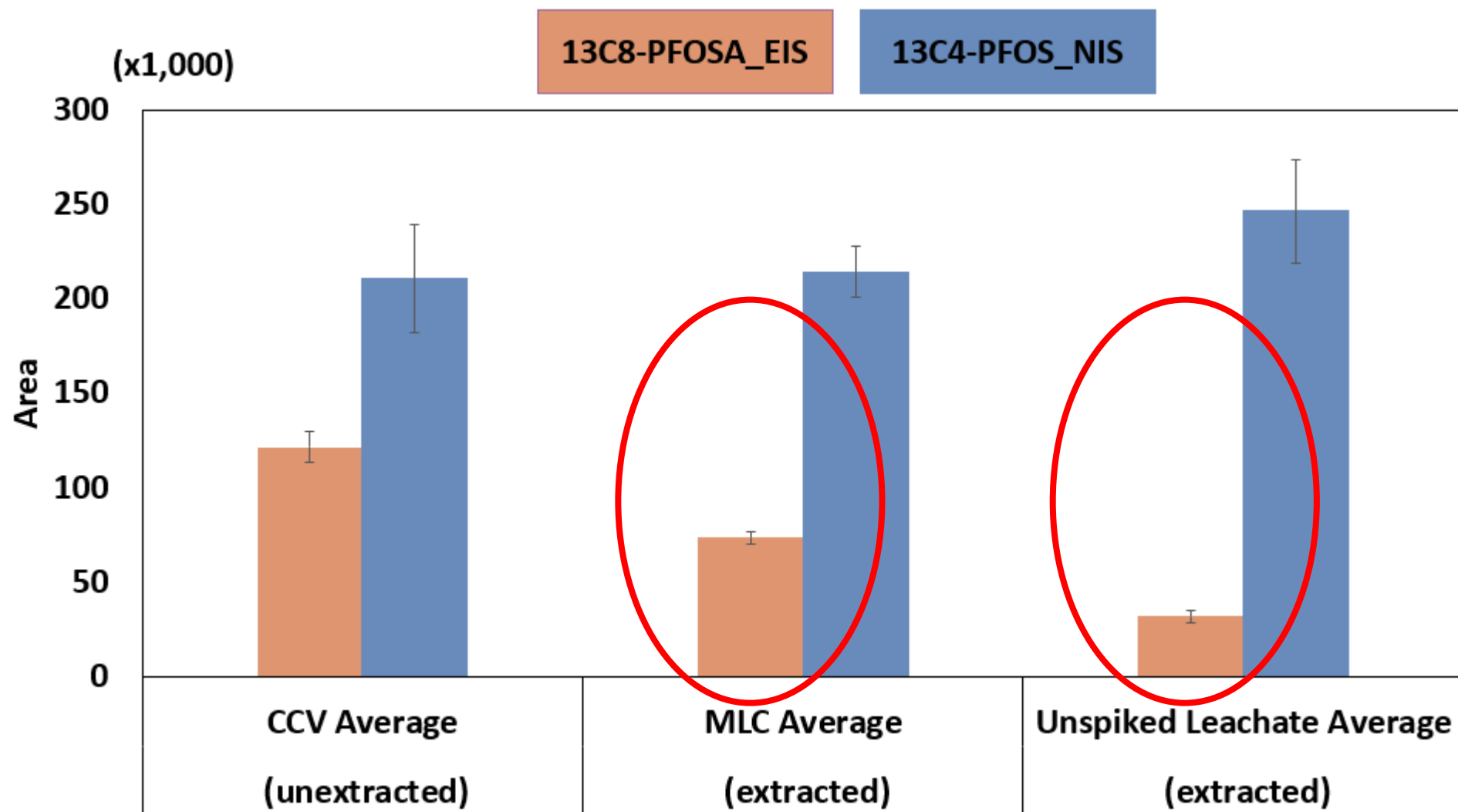
$$\frac{A_{EIS}}{A_{NIS}} = Recovery$$

Area EIS vs NIS



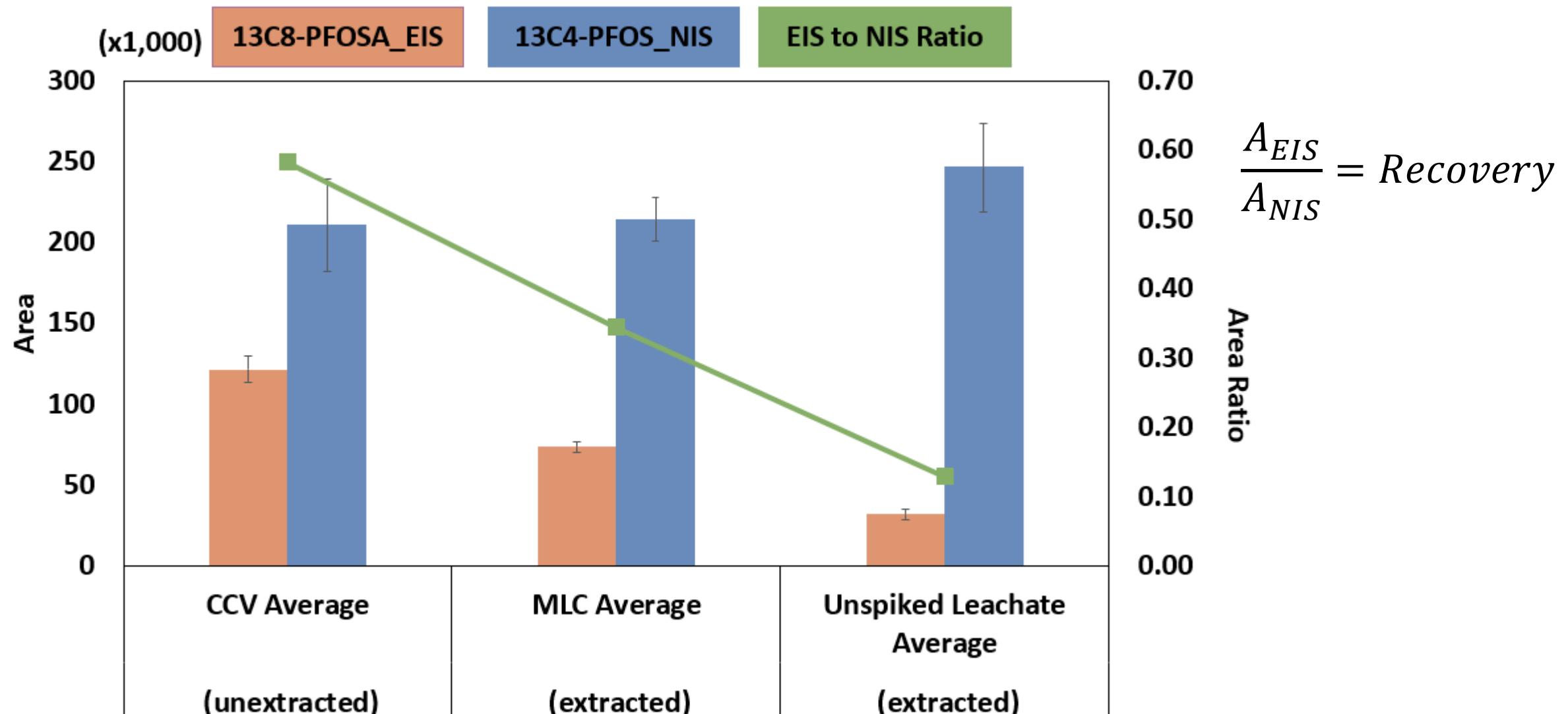
- EIS extraction efficiency < 100% for Lab Control

Area EIS vs NIS

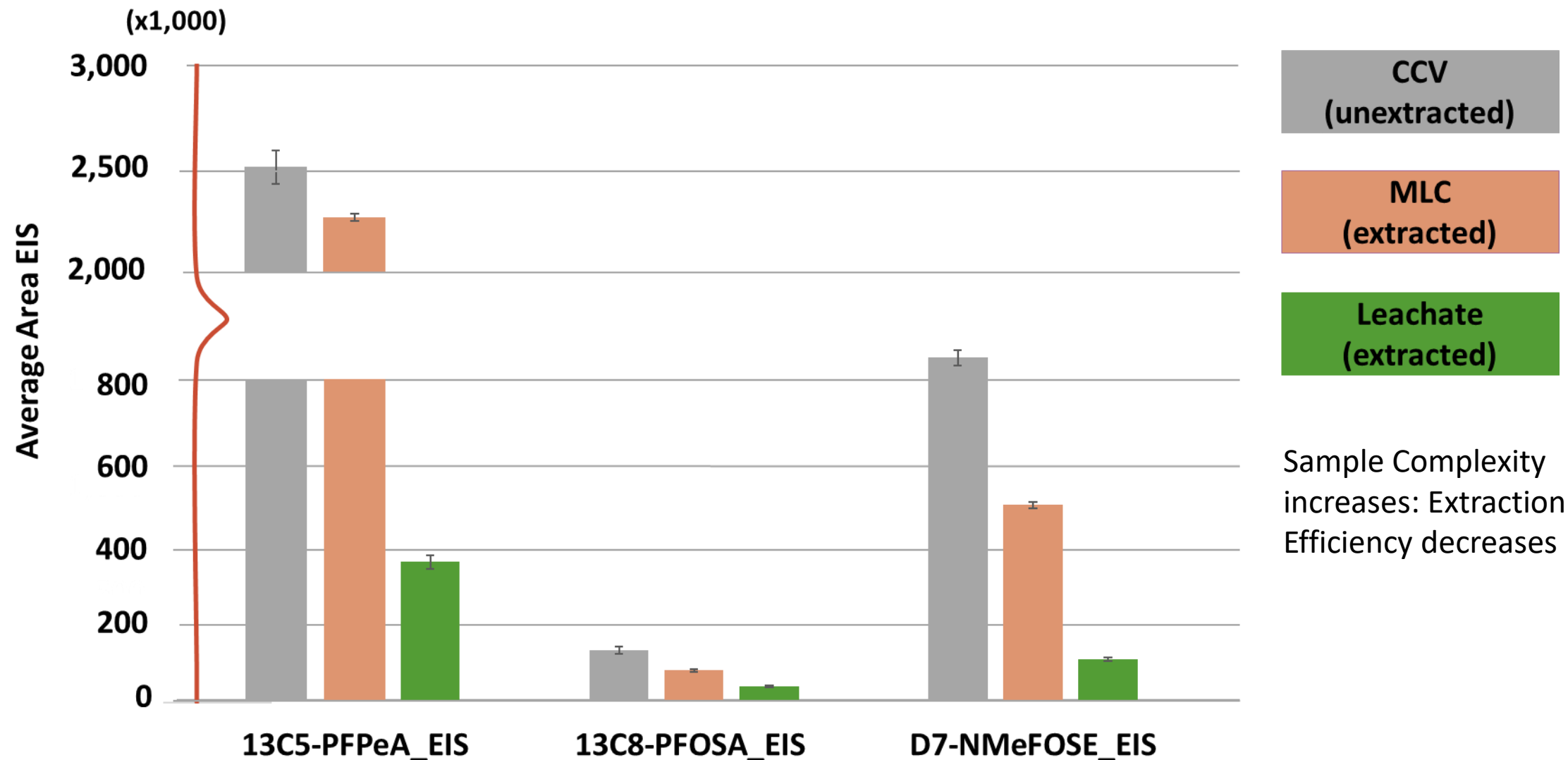


- EIS extraction efficiency < 100% for Lab Control
- EIS extraction efficiency is reduced further in leachate
- NIS is stable

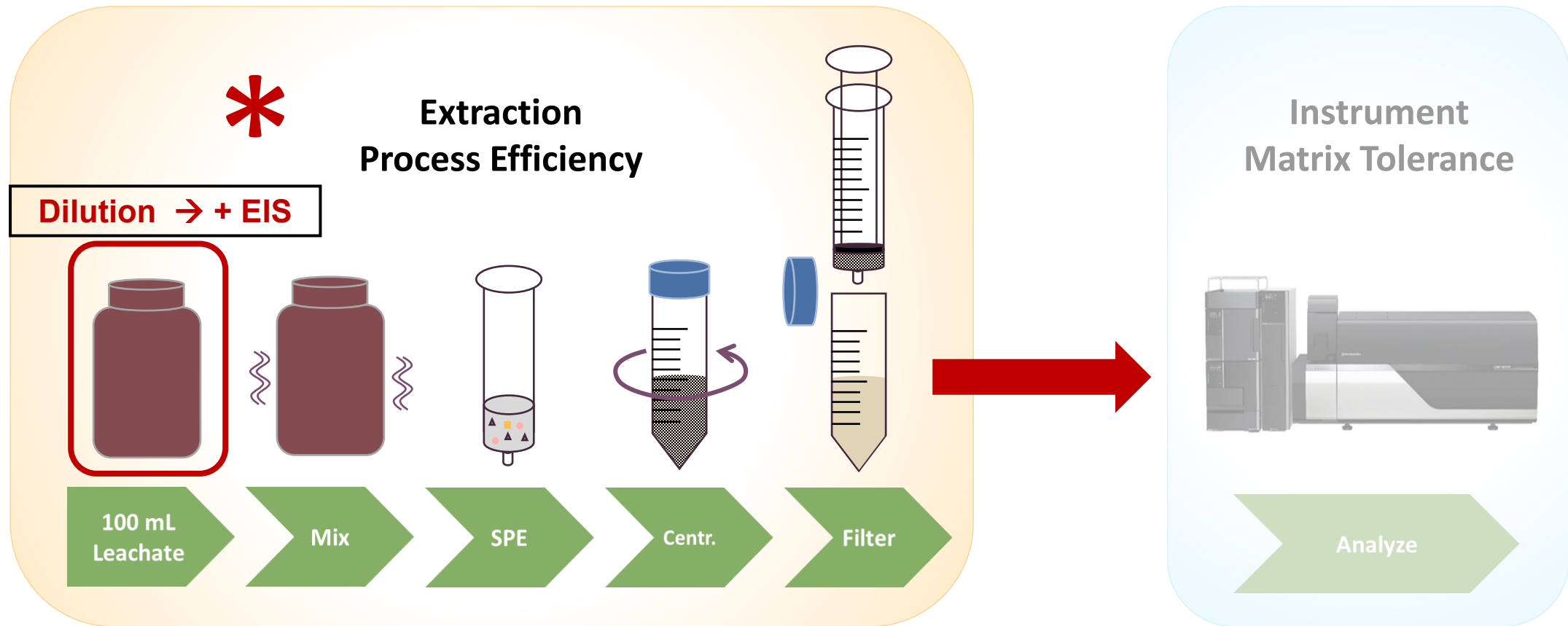
Area EIS & NIS Ratio



Area of Extracted Internal Standards



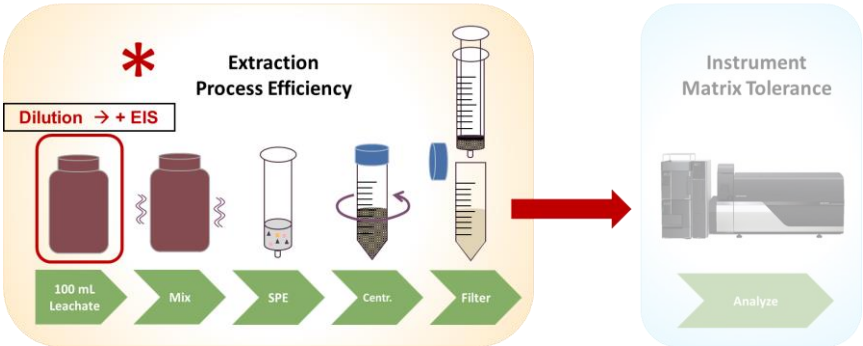
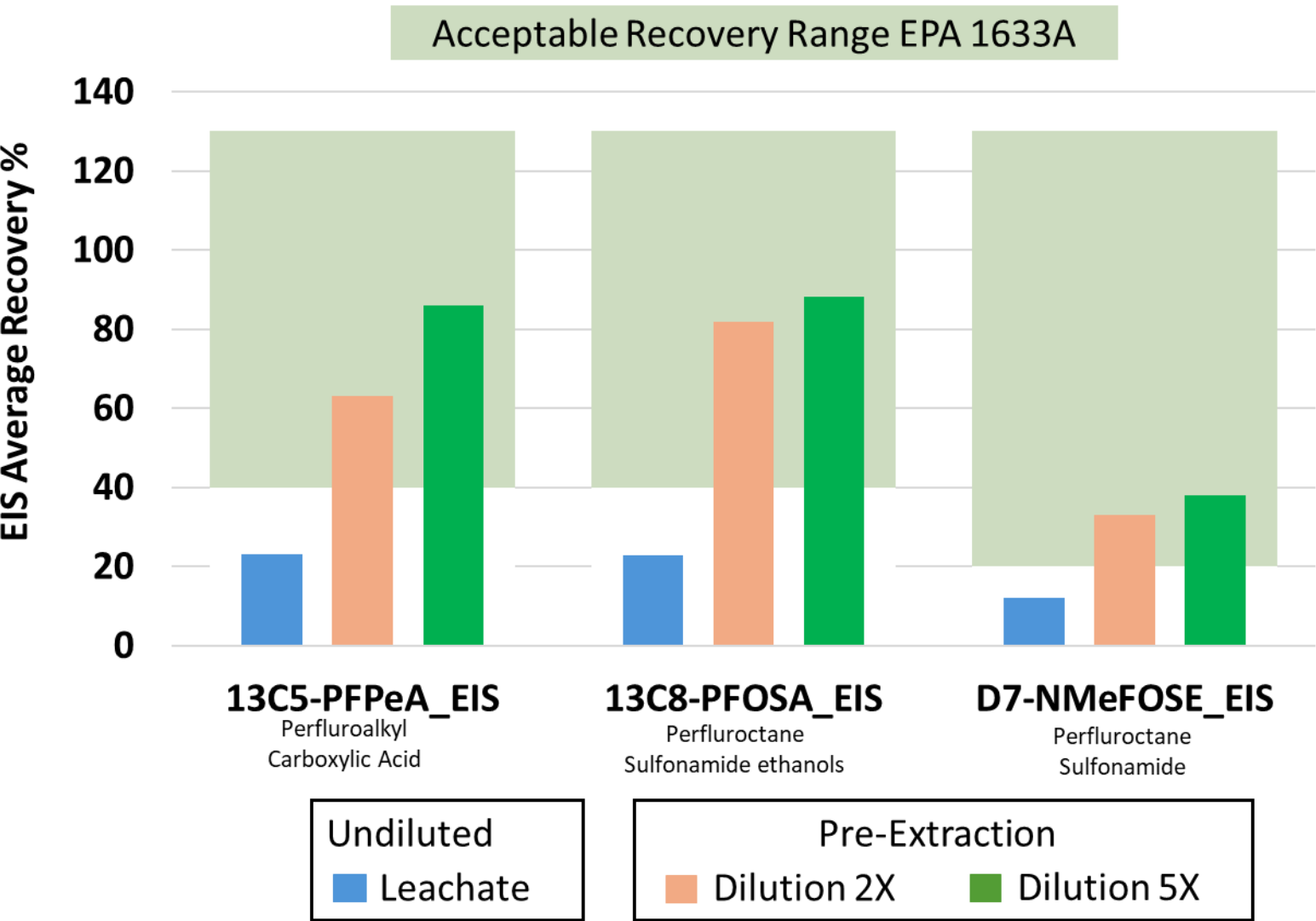
Pre-Extraction Dilution Process



Extraction Process Efficiency: Evaluated by dilution of leachate pre-extraction.

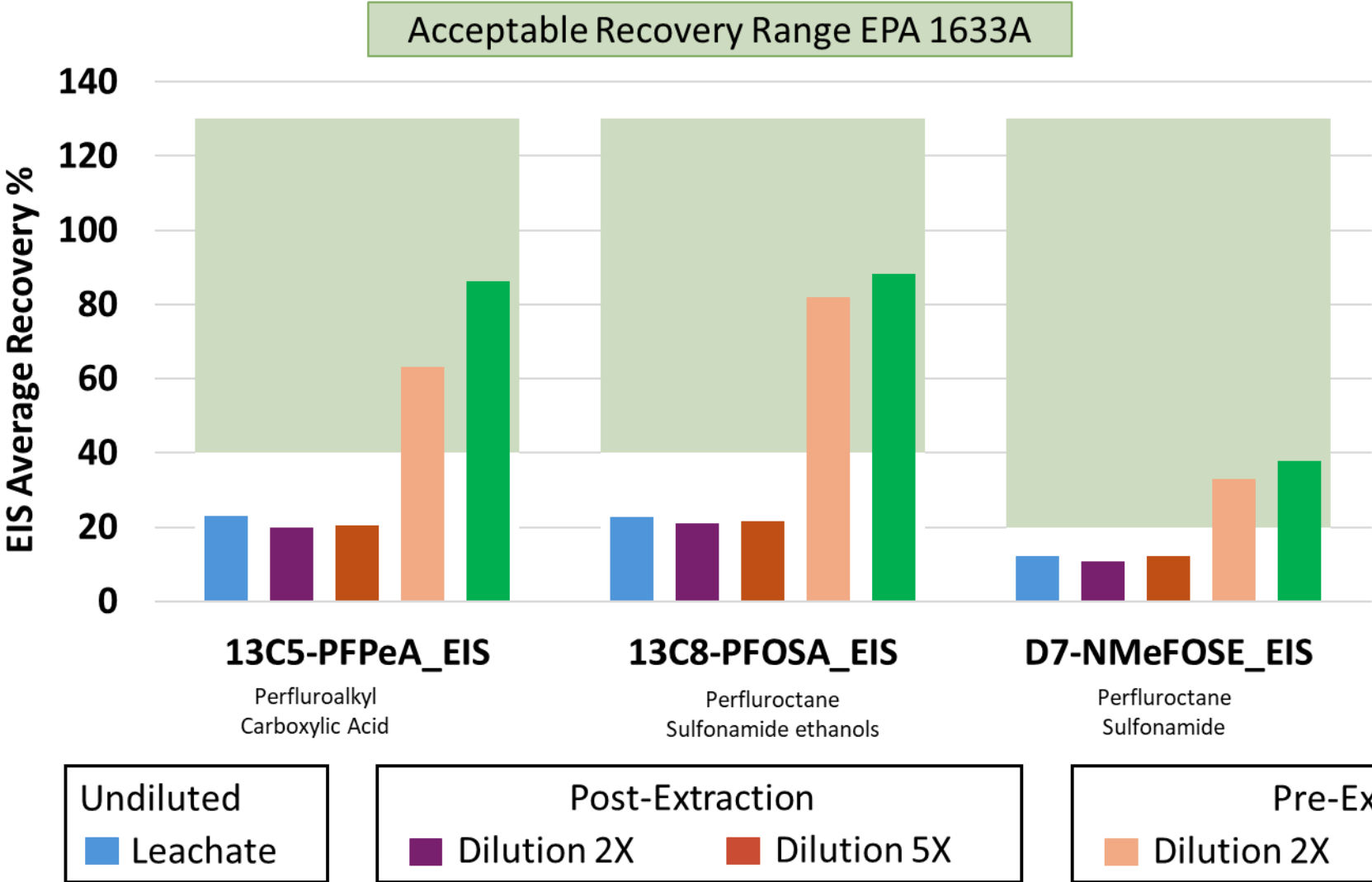
- 2x & 5x dilution

Pre-Extraction Dilutions EIS Impact



- Extraction efficiency is an issue.

Dilutions Comparison: EIS Impact



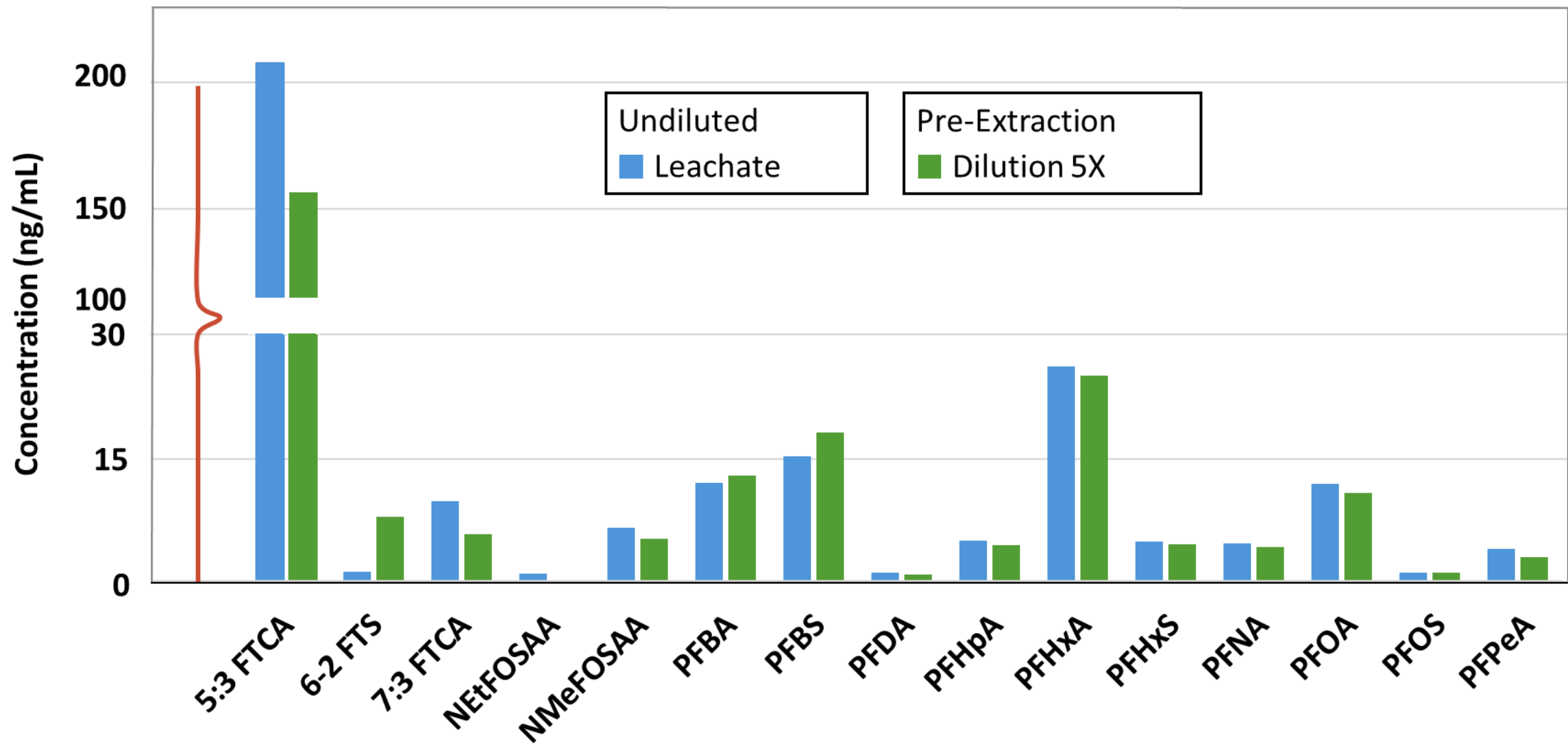
Post-Sample Prep Extraction

- Evaluates Instrument Matrix Tolerance
- not** the issue

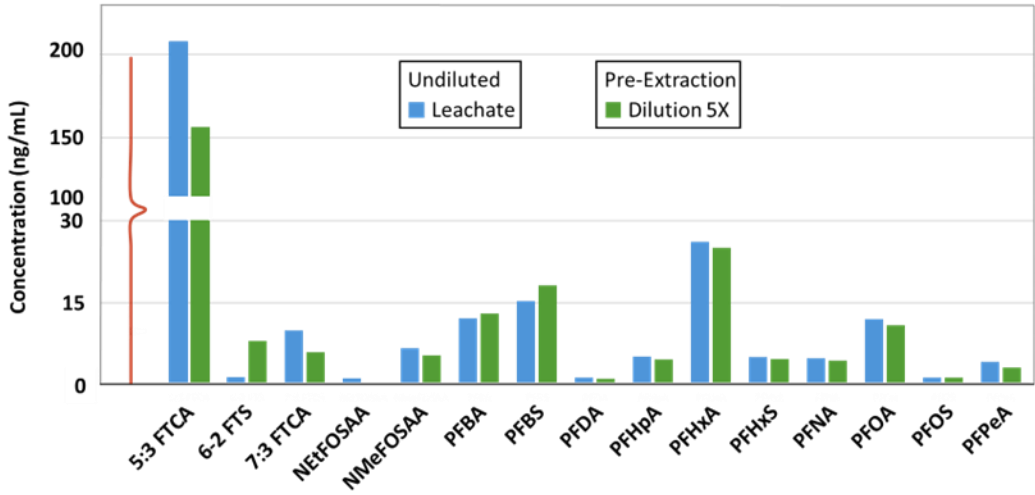
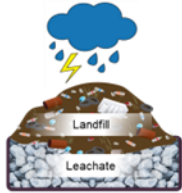
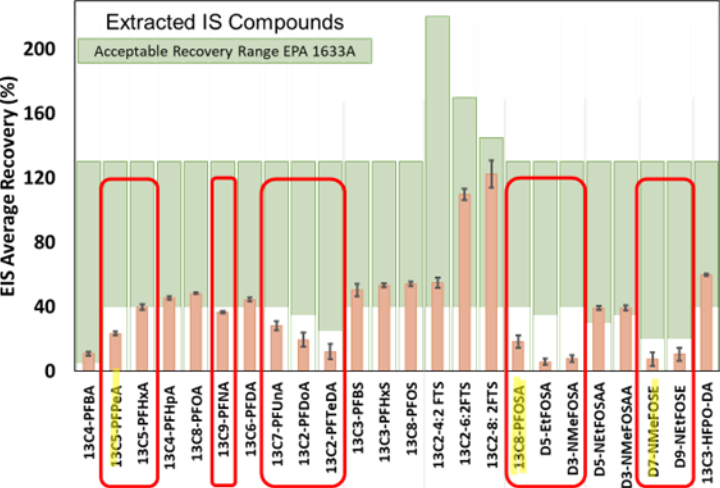
Pre-Sample Prep Extraction

- Evaluates the Efficiency of the Extraction Process
- is a concern

Effects of Dilution on Quantitation of Native PFAS in Leachate Sample

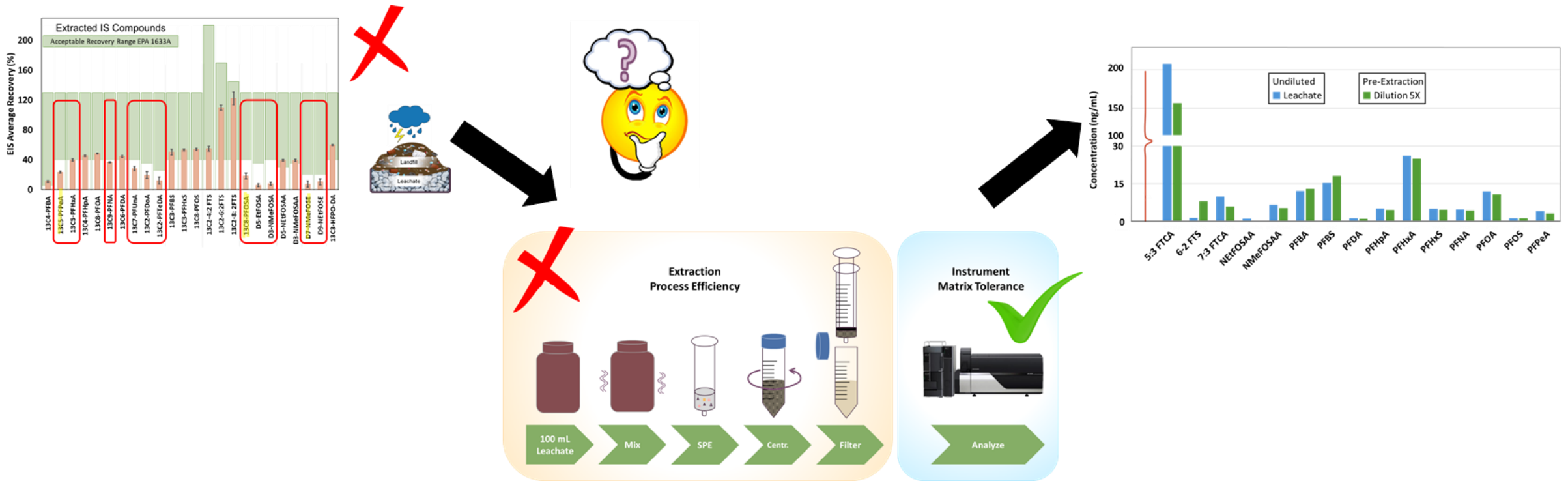


Conclusions



- Crucial to isolate the reason for bad recovery to ensure accurate results.

Conclusions



- **Crucial to isolate the reason for bad recovery to ensure accurate results.**
 - Instrument Matrix Tolerance: not an issue.
 - Extraction Process Efficiency: decreases with increased sample matrix complexity

THANK YOU!

Made possible with the support of

- Ruth Marfil-Vega, PhD
- Evelyn Wang, PhD
- Toshiya Matsubara, PhD
- Robert English, PhD
- Kevin Schug, PhD (UTA)
- Om Shrestha

Tiffany Liden, PhD
tmliden@shimadzu.com



Shimadzu Center for
Advanced Analytical Chemistry