Accurate Mass Library for PFAS Analysis in Environmental Samples Using High Resolution GC/Q-TOF

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Background

Per- and polyfluoroalkyl substances (PFAS) are emerging contaminants that are found in water, air, soil as well as wildlife

There are currently thought to be over 6,000 PFAS that have been commercially produced

A subset of PFAS that have been detected in the environment can be volatile or semi-volatile

A variety of analytical techniques are necessary for their detection

GC/MS is typically used for detecting volatile and non-polar PFAS compounds



PFAS Family & Classes

(ITRC; https://pfas-1.itrcweb.org/2-2-chemistry-terminology-and-acronyms/)





Accurate Mass GC/MS Libraries

Allow for flexibility of workflows (targeted and non-targeted)

Improve confidence in identification due to accurate mass

- Time-efficient: less review required
- Decrease false positive rate

Flurotyl





Workflows for the Accurate Mass PCDL









Non-Targeted Screening Workflow





Target/Suspect Screening and Quantitation





ChemVista Library Manager

New software for managing spectra, compounds, and RTs

- Provides access to multiple public databases (import/export, export without duplicates)
- Organize, manage, edit or create spectra
- Cheminformatics underpinning for structure generation and organizational focus
- Compound centric structure allows multiple spectra, RT and RI representing different analytical conditions
- Easily generated lists cater to multiple application workflows
- Facilitate identification workflows within MassHunter data analysis applications and beyond





Workflows Center Around Lists

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Using ChemVista to Enhance Identification Workflows

Import GCMS spectra from MassBank of North America (MoNA)

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Sampling and Extractions

Soil:

- Sampled from 2 fields in California
- Extracted with methylene chloride

Drinking water samples:

- Collected from two different water source categories:
 - a small surface water (Weaverville)
 - a mixed surface and ground water (Irvine)



- Extracted on a multi-mode SPE (HLB, WAX, WCS, Isoelut ENV)
- Eluted with 5% MTBE in MeOH, DCM, 0.5% NH4OH in 1:1 EtAC:MeOH, and 1.7% formic acid in 1:1 EtAC:MeOH
- The combined extracts were concentrated, solvent exchanged to EtAc and diluted 10x



Experimental Conditions

GC and MS Conditions	DB-5MS	DB-624
MS	7250	Q-TOF
GC	7	890
Inlet	MMI, 4-mm UI liner	single taper with wool
Inlet temperature	70 °C for 0.01 min;	300 °C/min to 250 °C
Injection volume	1	LμL
Columns	DB-5MS UI, 30 m x 0.25 mm x 0.25 μm	DB-624 UI, 30 m x 0.25 mm x 1.4 μm
Oven temperature program	35 °C for 2 min; 7 °C/min to 210 °C, 20 °C/min to 300 °C, 4 min hold	30 °C for 2 min; 3 °C/min to 7 °C, 2 °C/min to 110 °C, 10 °C/min to 210 °C, 20 °C/min to 240 °C, 2 min hold
Column flow	1.2 mL/min constant flow	1 mL/min constant flow
Carrier gas	Не	lium
Transfer line temperature	25	50 °C
Quadrupole temperature	15	50 °C
Source temperature	20	0°C
Electron energy	7() eV
Emission current	Variable by time se	egment, 0.01 to 5 μA
Spectral acquisition rate	5	5 Hz
Mass range (Tune)	50 to 2	1200 m/z



7250 GC/Q-TOF



Accurate Mass PFAS Library for GC/Q-TOF

- Contains high resolution accurate mass spectra of over a hundred PFAS compounds (~140)
- Expert curated and converted to theoretical m/z
- Retention Time
- Molecular structures
- CAS and other identifiers

	Name	Formula	Mass	Retention Time	CAS	
	2,2-Difluoroethyl triflate	C3H3F5O3S	213.97231	12.25	<u>74427-22-8</u>	
	Flurotyl	C4H4F6O	182.01663	3.74	<u>333-36-8</u>	
	4H-Perfluorobutanoic acid	C4H2F6O2	195.9959	14.354	<u>679-12-9</u>	
	PFUnDA / Perfluoroundecanoic acid (PFUnA)	C11HF21O2	563.96412	6.124	<u>2058-94-8</u>	
	Perfluoropentanamide	C5H3F8NO	245.00869	37.26	<u>355-81-7</u>	
•	6:2 Fluorotelomer methacrylate	C12H9F13O2	432.0395	36.783	<u>2144-53-8</u>	
	8:2 Fluorotelomer sulfon	010115517000	F07 00070	07.44	20100.24.4	
	N-Methyl-N-(2-hydroxyel 100 + 69:0834	9				
	11:1 Fluorotelomer alcol 월 50- 100.00	100/			432.0	.97
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Creation of the Accurate Mass PCDL

- Automated fragment formula annotation based on accurate mass and isotope ratios
- Conversion to theoretical m/z
- Noise ions are automatically removed



Name	Formula	Mass	Retention Time	CAS	IUPAC	SMILES	InChI Key	InChl	NumSpectra
Perfluoro-3,6,9-trioxatridecanoic acid	C10HF19O5	561.95206	18.85	<u>330562-41-9</u>	Difluoro{1,1,2,2.t	OC(=O)C(F)(F)OC(GDQL STEWIOEA		1
1-(Perfluorofluorooctyl)propane-2,3-diol 2	C11H7F17O2	494.01746	18.89	<u>94159-84-9</u>	4,4,5,5,6,6,7,7,8,	OCC(0)CC(F)(F)C(CGRI	nt	
3H-Perfluoro-2,2,4,4-tetrahydroxypentane	C5H5F7O4	262.00761	19.54	<u>77953-71-0</u>	1,1,1,3,5,5,5-Hep	OC(O)(C(F)C(O)(O	NZCX		
3-(Perfluorohexyl)-1,2-epoxypropane	C9H5F13O	376.01328	19.71	<u>38565-52-5</u>	2-(2,2,3,3,4,4,5,5,	FC(F)(F)C(F)(F)C(F	KGYU		
3-(Perfluoropropyl)propanol	C6H7F7O	228.03851	20.206	<u>679-02-7</u>	4,4,5,5,6,6,6-Hep	OCCCC(F)(F)C(F)(VACK		
6:1 Fluorotelomer alcohol	C7H3F13O	349.99763	20.56	<u>375-82-6</u>	2,2,3,3,4,4,5,5,6,	OCC(F)(F)C(F)(F)C	STLN		F F
1-lodo-1H,1H,2H,2H-perfluoroheptane	C7H4F11I	423.9182	21.35	<u>1682-31-1</u>	1,1,1,2,2,3,3,4,4,	FC(F)(F)C(F)(F)C(F	KEHJ	~	ХХ



🔆 Agilent

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PFAS Classes in the PCDL





1,1,1,2,2,3,3,4,4-Nonafluoro-6-iodohexane (6:2 FTI)







PFAS Classes in the PCDL



2,2,3,3,4,4,4-Heptafluorobutyl methacrylate





Methyl perfluorohexanoate





SureMass Deconvolution and Library Search



SureMass signal processing



Examples of PFAS Identified in Soil and Water Using the PFAS PCDL

Sample: soil extracted from Field 1







Examples of PFAS Identified in Soil and Water Using the PFAS PCDL

Sample: drinking water (Irvine)









Disinfection by-products:

*Bromodichloromethane *Chloral *Dichloroacetonitrile *Bromoacetonitrile *Dibromochloromethane 1,1-Dimethyl-3-chloropropanol *Bromochloroacetonitrile Dichloroacetic acid methyl ester *Tribromomethane Methyl bromo(chloro)acetate *Dibromoacetonitrile 2,2-Dichloroacetamide

NIST20

Library Match score >75

* - known or probable carcinogens

Industrial processes:

*Tetrachloroethylene 1,2-Dichlorbenzene Caprolactam Phthalic anhydride Biphenyl Dimethyl phthalate 2,4-Di-tert-butylphenol Dibenzofuran **Diethyl Phthalate** 2-(Methylmercapto)benzothiazole Tributyl phosphate *Tris(2-chloroisopropyl)phosphate Di-sec-butyl phthalate

Dibutyl phthalate Octachlorostyrene Drometrizole Pyrene *Bisphenol A Bis(4-chlorophenyl) sulfone 2,2'-Methylene-bis-(4-methyl-6-tbutylphenol) Phthalic acid, di(2-propylpentyl) ester Bis[3,4-dichlorophenyl]sulfone *Bumetrizole *Bis(2-ethylhexyl) isophthalate *Decachlorobiphenyl

*2-Picoline *Methanesulfonate-methyl *o-Toluidine Phenol *2,4,5-Trimethylaniline 2-Nitrophenol Diphenylamine

Pesticides and environmental contaminants PCDL

Library Match score >75

* - known or probable carcinogens

NIST20

PAHs and their derivatives:

*Naphthalene 2-Methylnaphthalene Acenaphthene Fluorene 9H-Fluoren-9-one 9H-Fluoren-9-ol *Anthracene 9-Methylene-fluorene3-Methylphenanthrene2-MethylanthraceneCyclopenta(def)phenanthrenoneFluoranthenePyrene1-Azapyrene

NIST20

Library Match score >75

* - known or probable carcinogens

Personal care products:

Dimethyl phthalate Diethyl Phthalate *Benzophenone Di-sec-butyl phthalate

p-Phenylenediamine Acetophenone

NIST20

Library Match score >75

* - known or probable carcinogens

Pesticides and environmental contaminants PCDL



Pesticides:

Diethyltoluamide (DEET) *Hexachlorobenzene Cyclic octaatomic sulfur

Thanite Benzaldehyde Isoxadifen

NIST20

Library Match score >75

* - known or probable carcinogens

Pesticides and environmental contaminants PCDL



High Abundance Contaminants in Drinking Water Samples







Comparison of Water Sourced in Irvine vs Weaverville



Irvine vs Weaverville Water

Log of fold change (cut-off 2) vs. log of p-Value (cut-off 0.05)



Conclusions

- PFAS accurate mass library containing over 100 EI spectra has been created for high resolution GC/Q-TOF including several emerging volatile PFAS
- PFAS compounds have been identified in soil and water extracts using PFAS PCDL
- Additional contaminants have been identified in drinking water from two different source categories and included disinfection byproducts, chemicals from personal care products, drugs, pesticides and other industrial contaminants without re-injecting the sample



Acknowledgements & Reference



Tom Young's team



Matthew Curtis Andrew McEachran

ASMS 2023 Poster: TP-242

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Thank you!