

Scaling Liquid-Liquid Sample Preparation: Moving from Separatory Funnels to a Vial-Based Extraction for Semivolatile Organics Analysis by GC-MS/MS

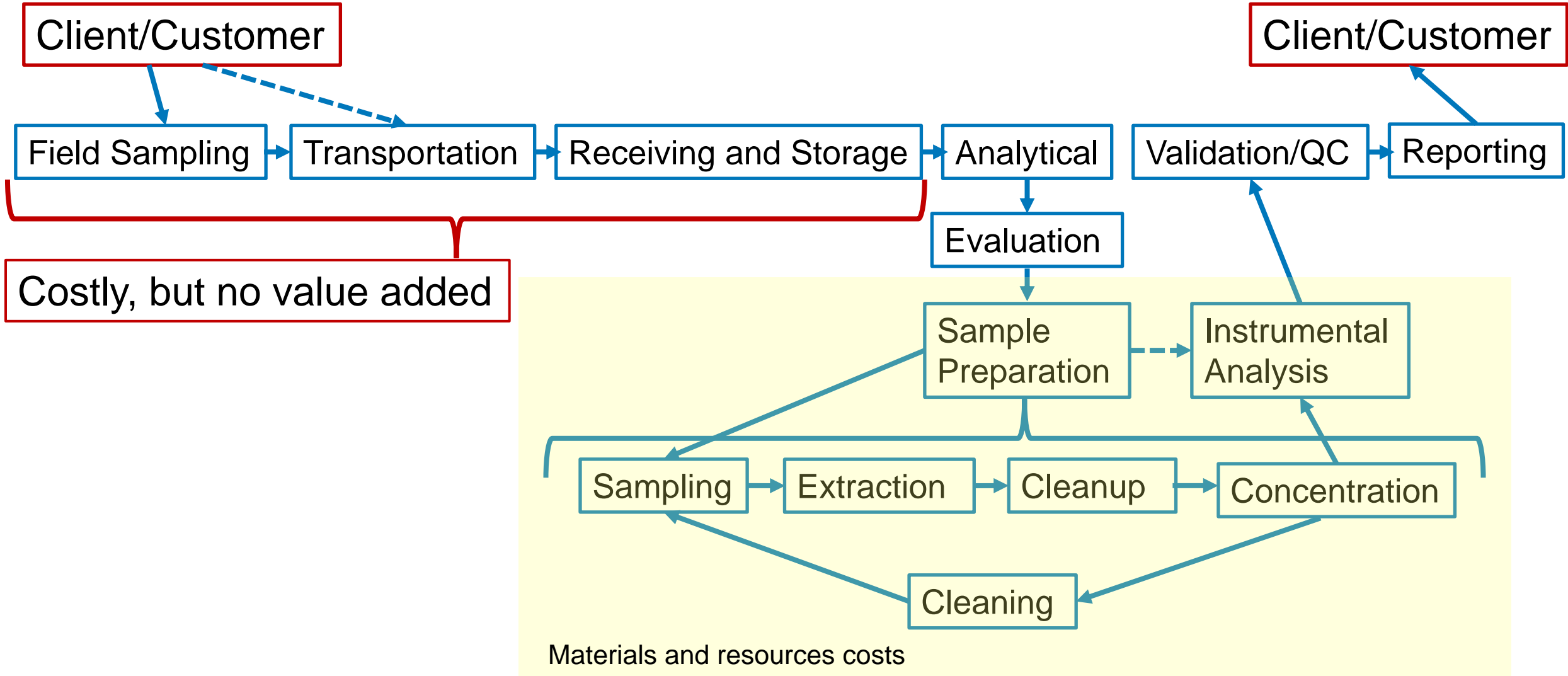
Ken Rosnack

and

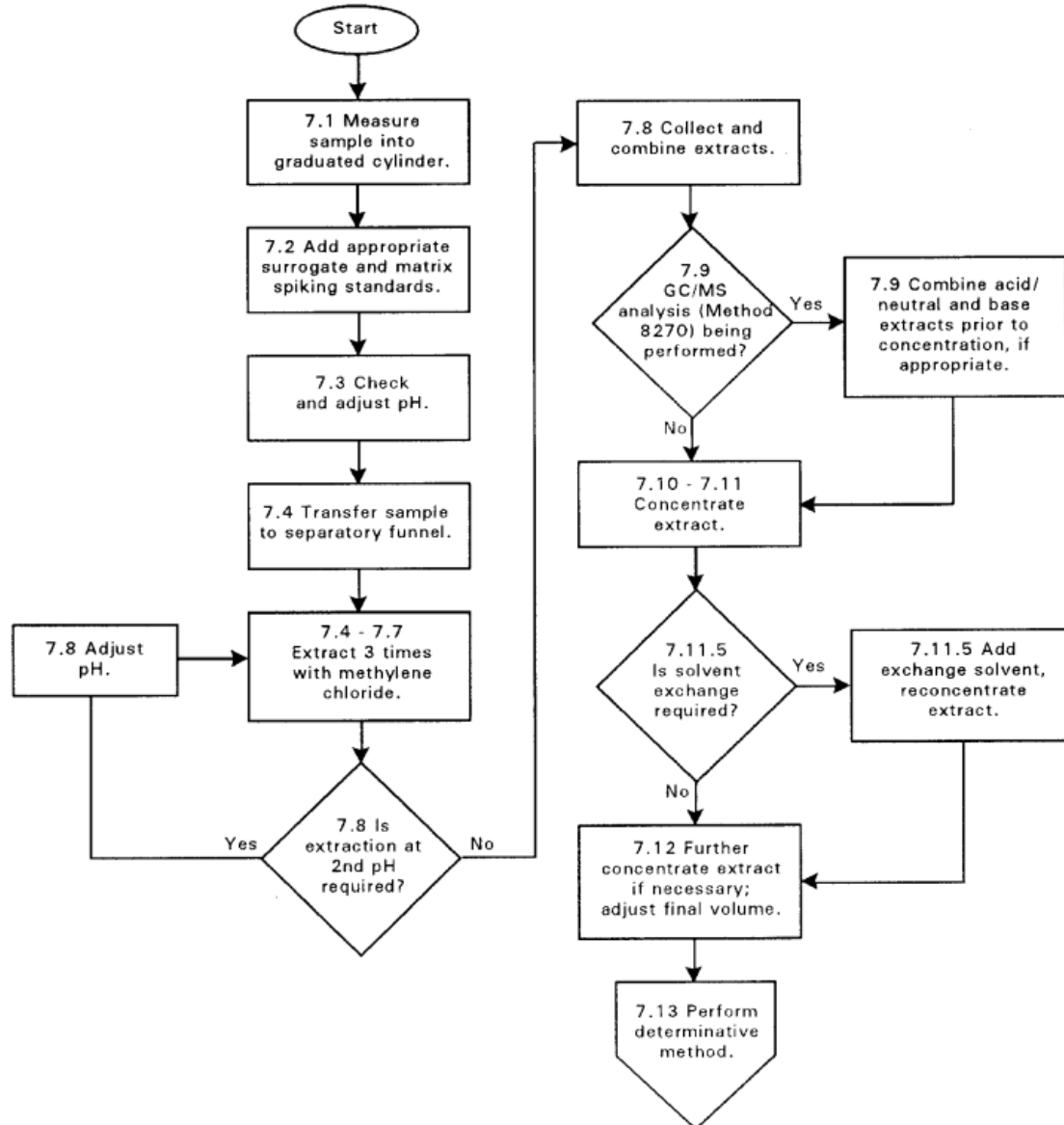
Frank Dorman, Sarah Dowd & Douglas Stevens

NEMC: Monday July 31st 11:00AM

The Environmental Laboratory (Extractable Organics)



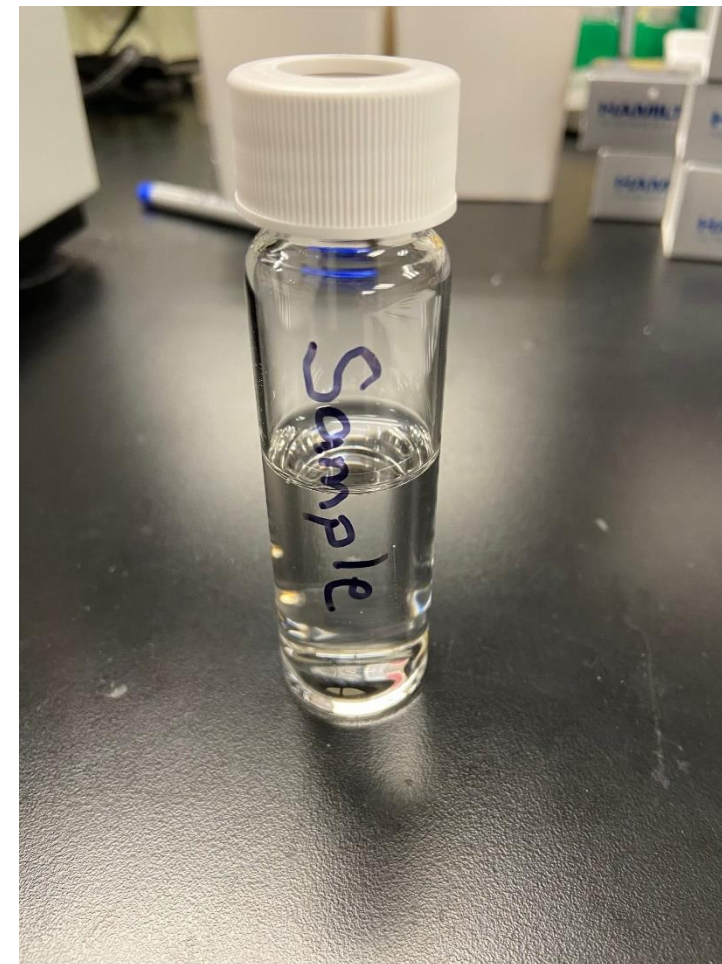
Original Sample Prep



- Based on EPA 3510C Separatory Funnel Liquid-Liquid Extraction
<https://www.epa.gov/sites/default/files/2015-12/documents/3510c.pdf>
- Starting quantity: 1 L of water
- Extraction Solvent: Methylene Chloride
 - 3 x 60 mL of methylene chloride extracts at pH 11
 - 3 x 60 mL of methylene chloride extracts at pH 2
- Extract is concentrated from ~360 mL to 1.0 mL for injection
- Uses a lot of consumables and reusable glassware!

Scaled Sample Prep

- Starting quantity: 20 mL of water
- Extraction Solvent: Methylene Chloride
 - 3 x 2 mL of methylene chloride extracts at pH 11
 - 3 x 2 mL of methylene chloride extracts at pH 2
- No concentration
- Disposable glassware – 40 mL VOA vials and glass pipettes



Scaled Sample Prep Procedure – proof of concept (This retains a similar volume ratio of aqueous to organic)

Load 20 mL of Water Sample to 40 mL VOA vial

Adjust the pH to 10 with Sodium Hydroxide

Add 2 mL of Methylene Chloride for Extraction

Sample and Methylene Chloride Mixed and Vortexed for ~2 min

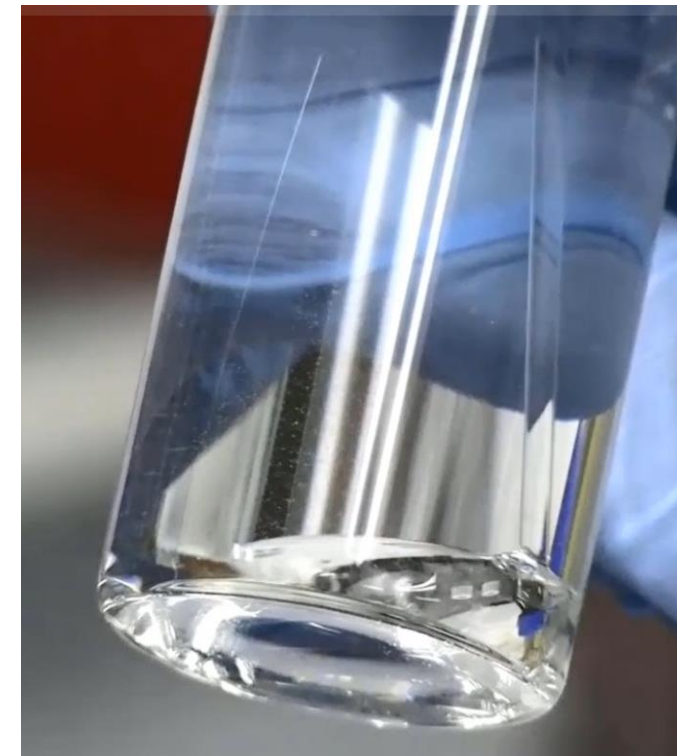
Methylene Chloride Extract Removed from the Sample Vial

Repeat Methylene Chloride Extraction 2 x for 6 mL total at pH 10

Adjust the pH to 2 with Sulfuric Acid

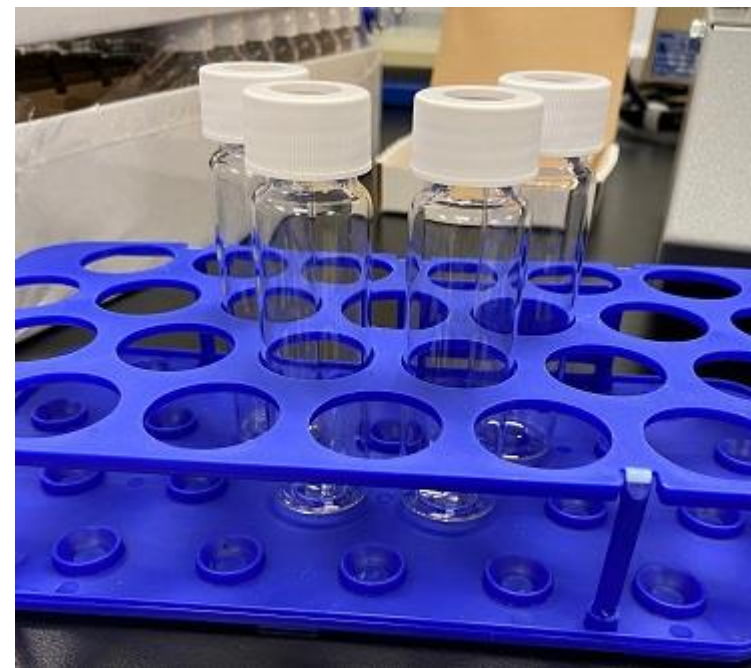
Extract with Methylene Chloride Extraction 3 x for 6 mL total at pH 2

Combine Extracts and Dry with Sodium Sulfate



Scaled Sample Prep Test

- 20 mL of water loaded into VOA vials
- Spiked Samples were prepared at 50 pg/ μ L with the Restek 8270MegaMix Standard (Catalog No. 31850)
 - 8 replicate spiked samples were prepared to test recovery
 - The MegaMix includes 76 SVOC compounds of varying chemistry
 - Internal standards (Restek Catalog No 31885) were added prior to injection on the GC-MS/MS
- A calibration curve from 3- 500 pg/ μ L was prepared in methylene chloride for sample quantitation



Experimental: Xevo TQXS MS Parameters

- Ionization: API⁺
- Corona pin: 2.0 μA
- Source temp: 120 °C
- Cone gas flow: 250 l.h⁻¹
- Auxiliary gas: 200 l.h⁻¹



Experimental: GC Method (N₂ Carrier Gas)

- Column: Rxi-SVOCms 20m x 0.15mm ID, 0.15 μm film thickness
- Transfer line temperature: 320°C
- Make-up gas: 300 ml.min⁻¹ (nitrogen)
- Injection Port Temperature: 280°C
- Injection Volume: 1 μL (10:1 split)
- Pressure Program:

Rate (psi/min)	Value (psi)	Hold (min)
Init.	10.182	0
90	30	0.1
99	10.29	0
1.4	28	0

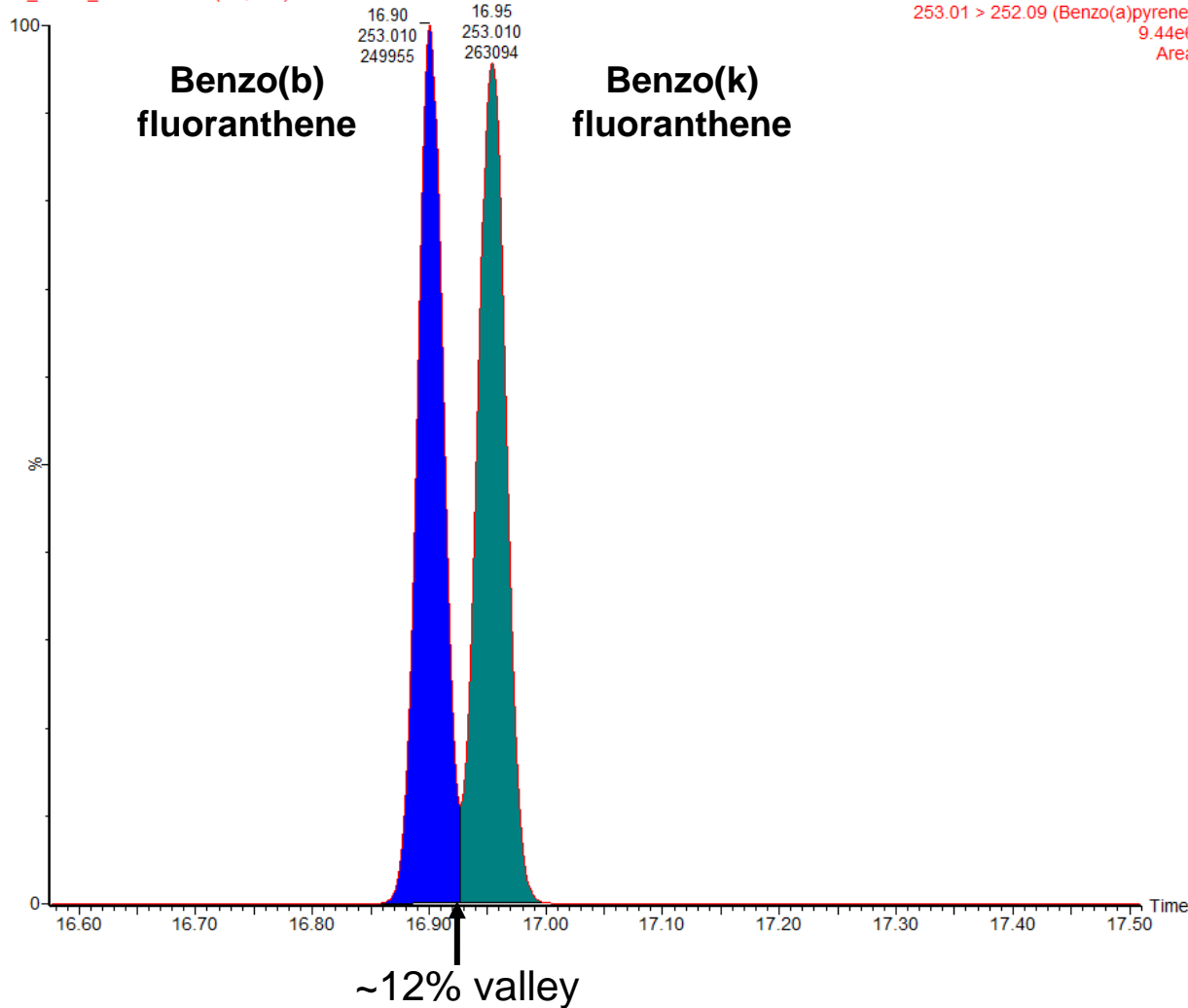
- Temperature Program:

Ramp (°C/min)	Temp (°C)	Hold (min)	Total Time (min)
Initial	45	0.6	
16.8	100	0	
20.9	270	0	
8.3	310	1.75	
20.8	320	3	22.058

Chromatographic Performance: N₂ Carrier Gas, 20m Column

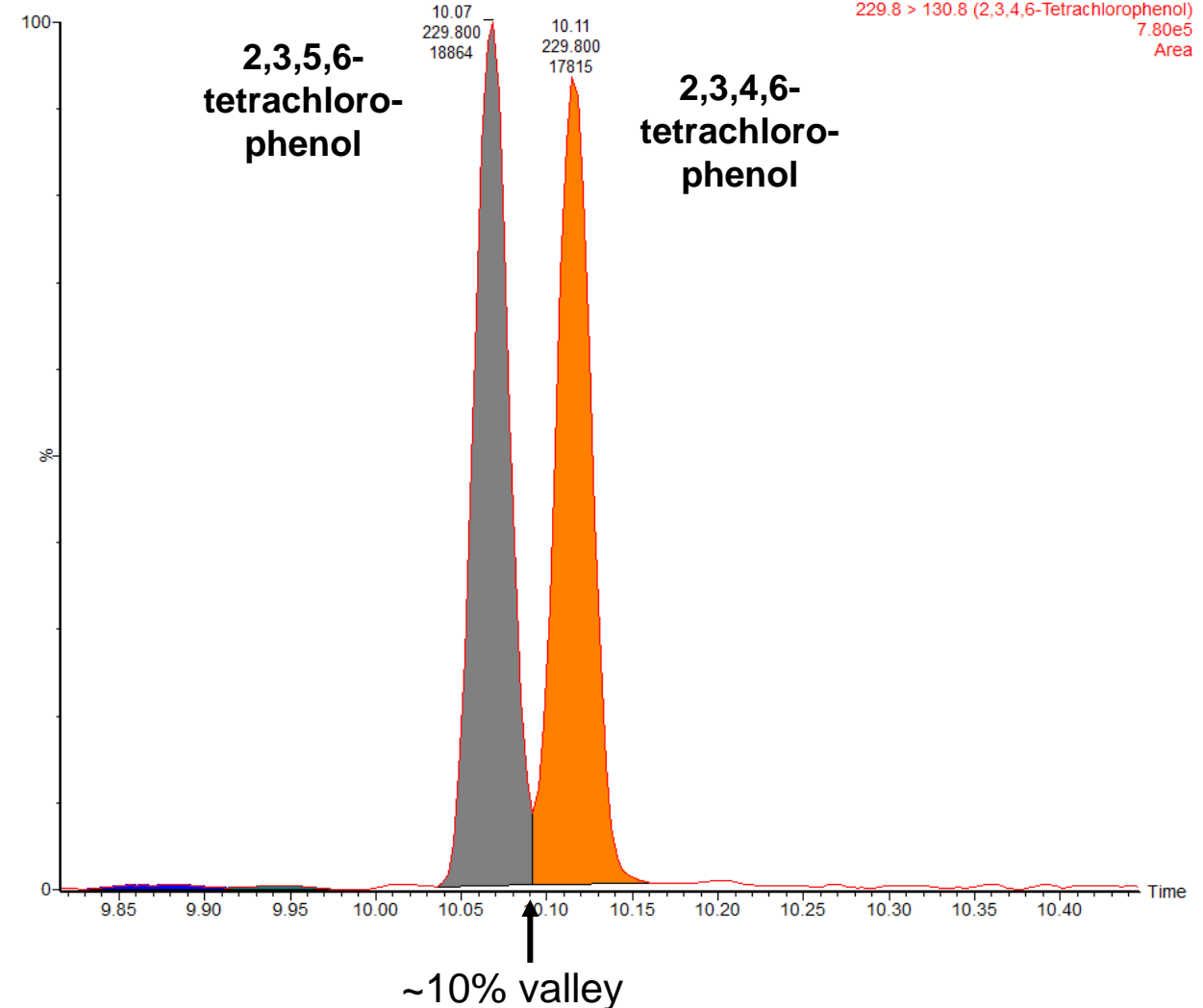
Benzo(b,k)fluoranthene

N2_SVOC_Curve105 Sm (Mn, 2x2)



Tetrachlorophenols

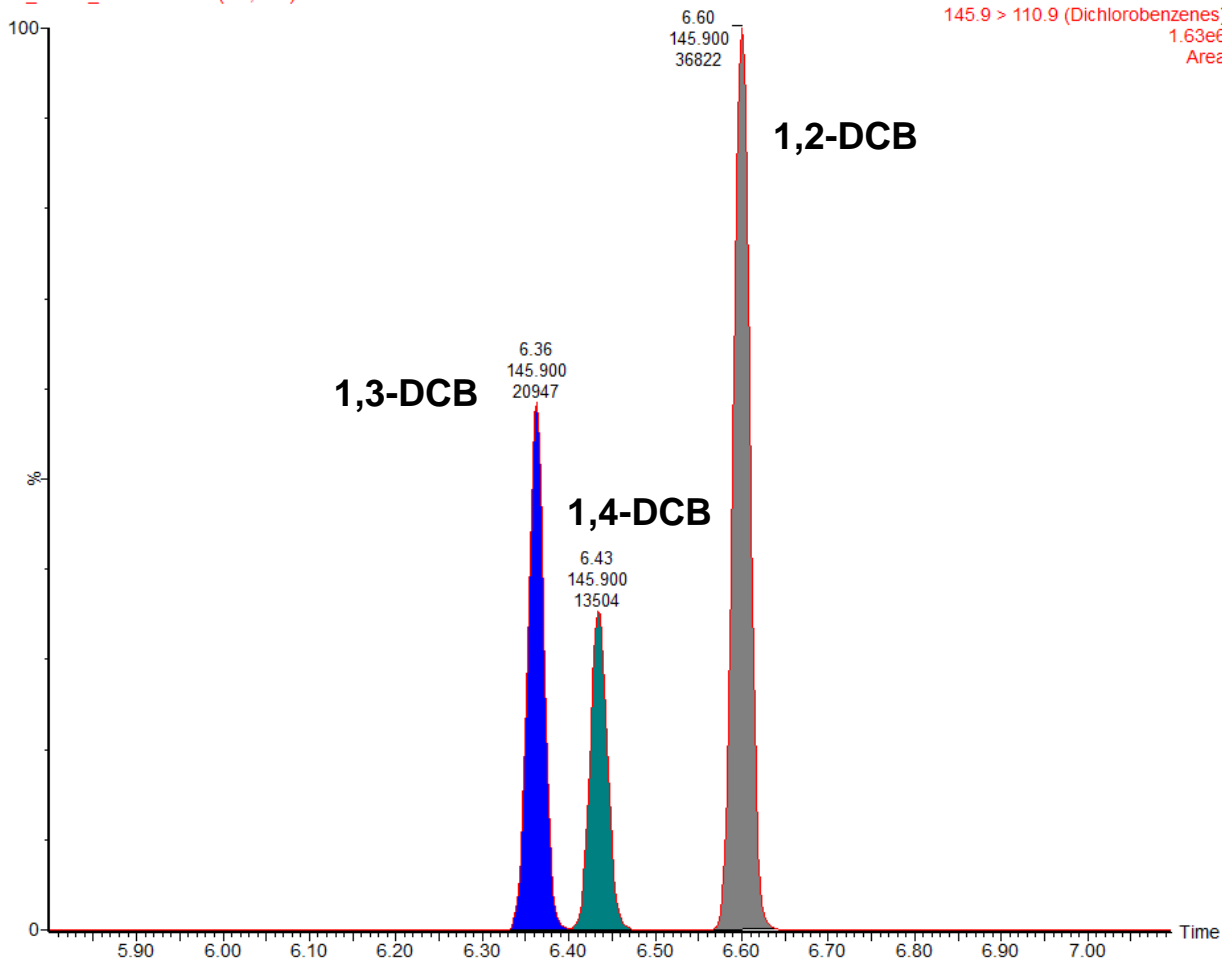
N2_SVOC_Curve105 Sm (Mn, 2x2)



Chromatographic Performance: N₂ Carrier Gas, 20 m Column

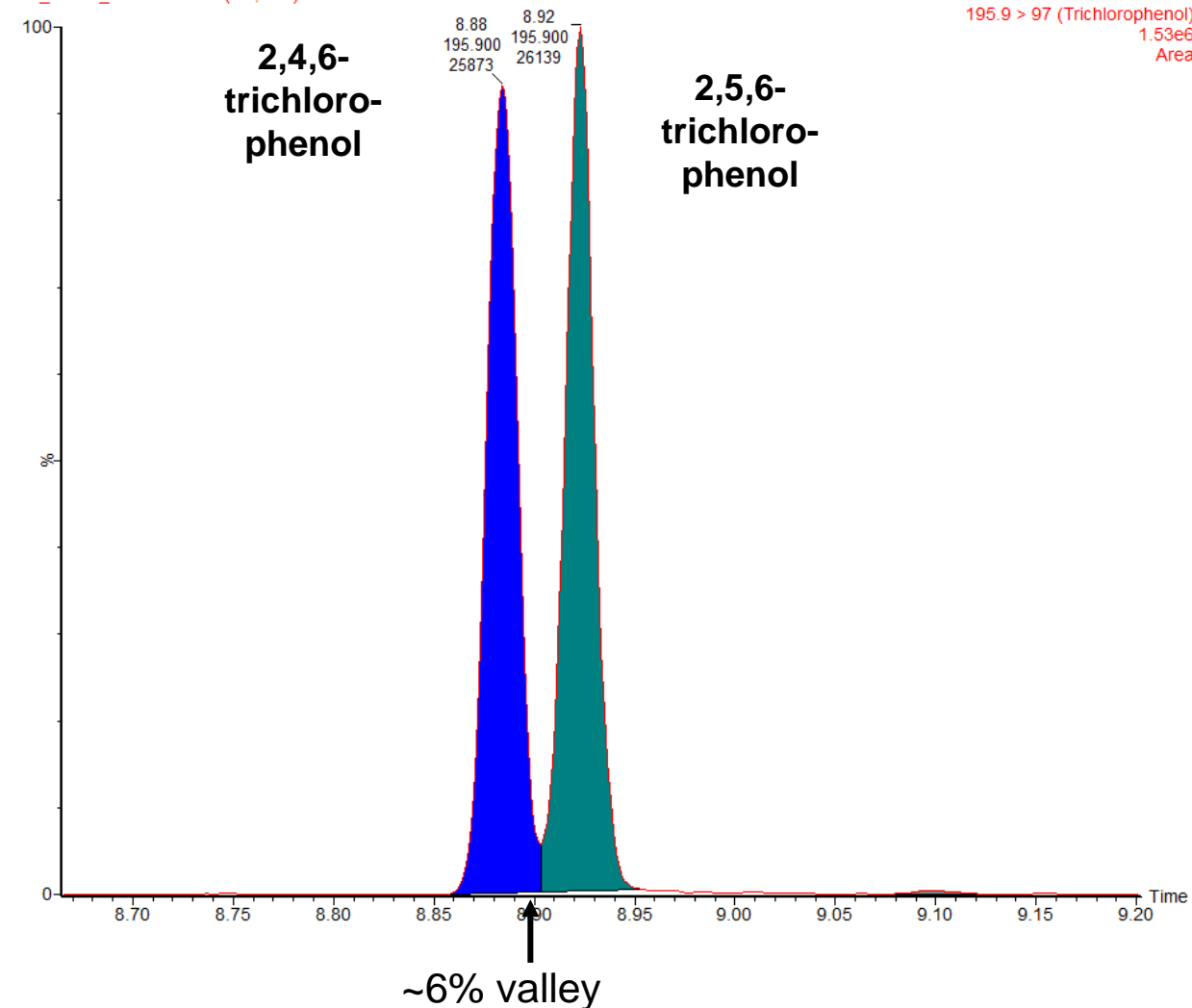
Dichlorobenzenes

N2_SVOC_Curve105 Sm (Mn, 2x2)



Trichlorophenols

N2_SVOC_Curve105 Sm (Mn, 2x2)

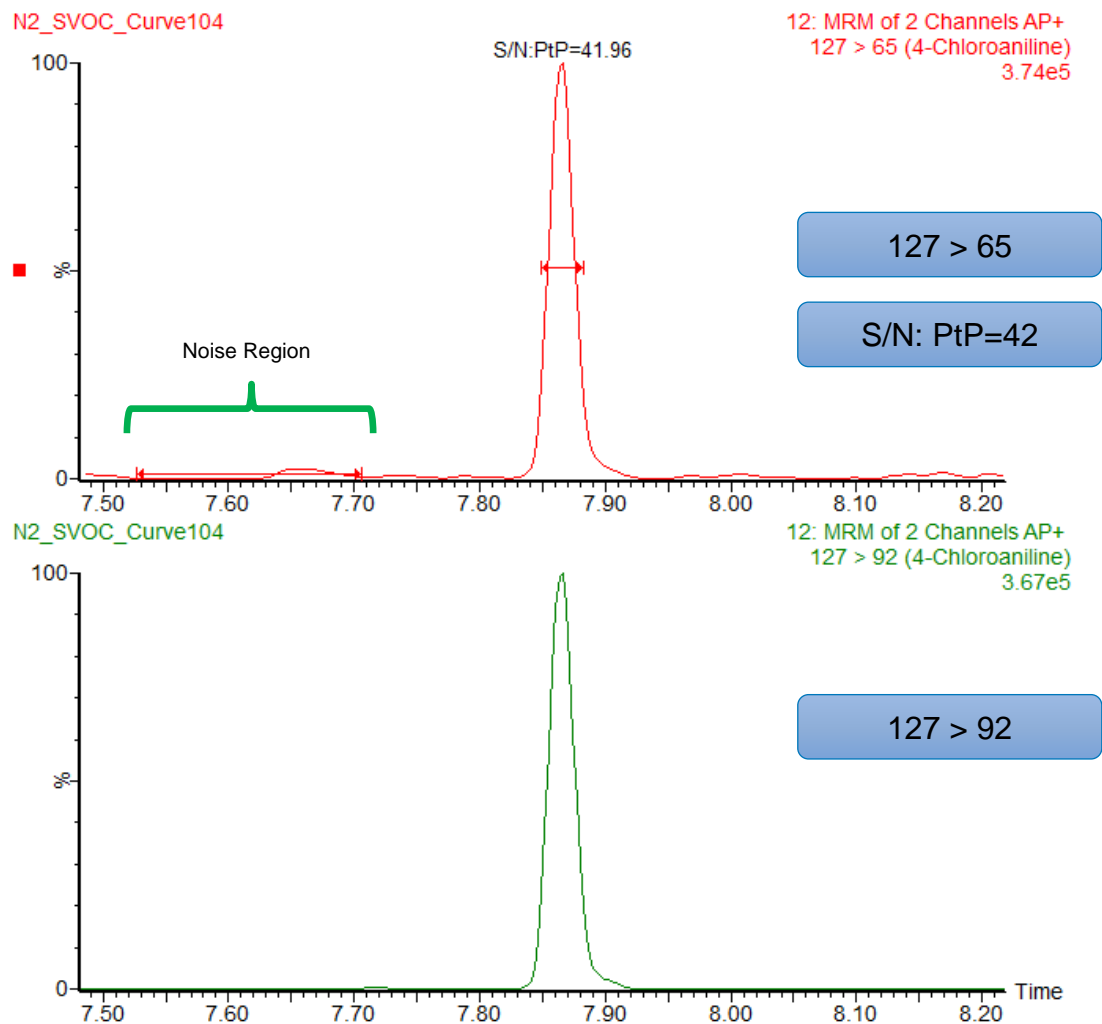


25: MRM of 2 Channels AP+
145.9 > 110.9 (Dichlorobenzenes)
1.63e6
Area

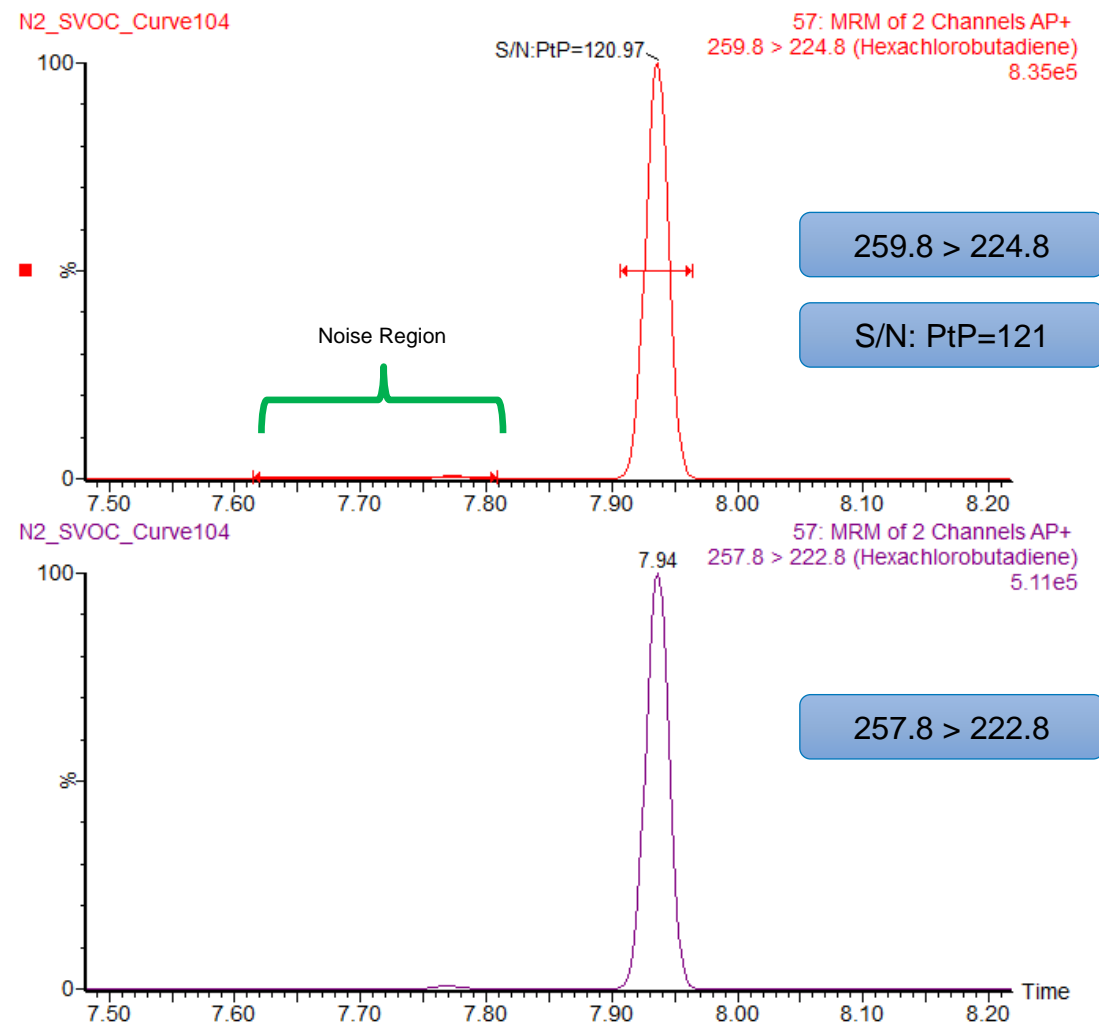
45: MRM of 2 Channels AP+
195.9 > 97 (Trichlorophenol)
1.53e6
Area

Sensitivity of the APGC-MS/MS Method

4-Chloroaniline at 5 pg/μL



Hexachlorobutadiene at 5 pg/μL



Analysed the Restek 8270 MegaMix Standard (cat.# 31850), containing 76 components, spiked with the SV Internal Standard Mix (cat.# 31206). Compounds fall into three categories:

Charge Transfer Only

1,2,4-Trichlorobenzene, 1,2-dichlorobenzene, 1,3-dichlorobenzene, 1,4-dichlorobenzene, **1,4-Dichlorobenzene-d4**, 1-Methylnaphthalene, 2,3,4,6-Tetrachlorophenol, 2,3,5,6-Tetrachlorophenol, 2,4-Dichlorophenol, 2,4-Dimethylphenol, 2,4-Dinitrophenol, 2,4-Dinitrotoluene, 2-Chloronaphthalene, 2-Chlorophenol, 2-Methylnaphthalene, 2-Methylphenol, 2-Nitroaniline, 2-Nitrophenol, 3-Methylphenol, 3-Nitroaniline, 4-Bromophenyl phenyl ether, 4-Chloro-3-methylphenol, 4-Chloroaniline, 4-Chlorophenyl phenyl ether, 4-Methylphenol, 4-Nitroaniline, 4-Nitrophenol, Acenaphthylene, Aniline, Benzyl alcohol, Bis(2-chloroethyl)ether, Carbazole, Dibenzofuran, Dimethyl phthalate, Hexachlorobenzene, Hexachlorobutadiene, Hexachlorocyclopentadiene, Hexachloroethane, Pentachlorophenol, Phenol, Pyridine, Trichlorophenol

Both Modes

Acenaphthene
Acenaphthene-d10
Anthracene
Benzo(a)anthracene
Benzo(a)pyrene
Benzo(b)fluoranthene
Benzo(k)fluoranthene
Benzo[g,h,i]perylene
Chrysene, **Chrysene-d12**
Dibenz(a,h)anthracene
Fluoranthene, Fluorene
Indeno[1,2,3-cd]pyrene
Naphthalene, **Naphthalene-d8**
Perylene-d12
Phenanthrene
Phenanthrene-d10
Pyrene

Protonation Only

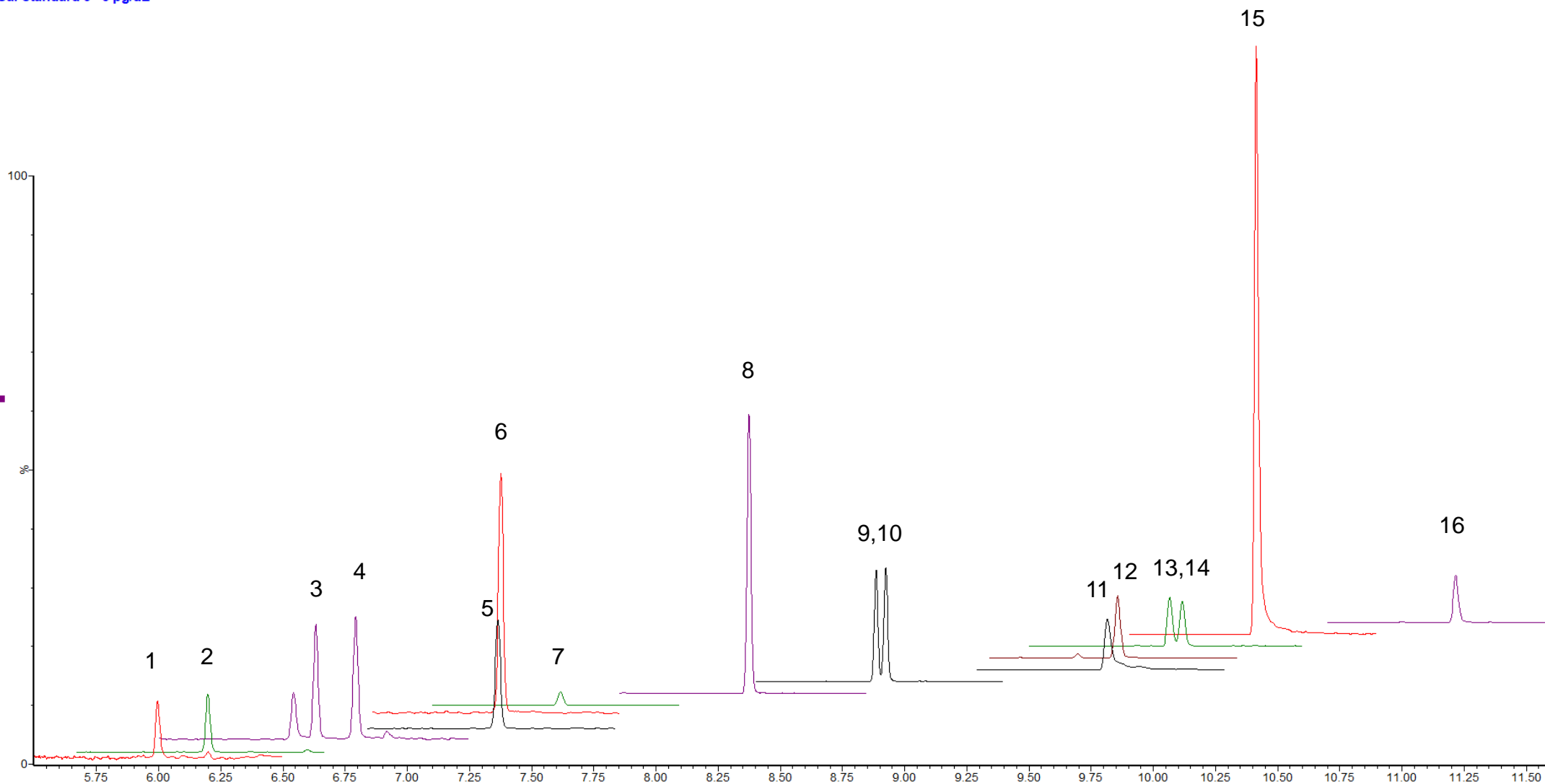
1,3-Dinitrobenzene
1,4-Dinitrobenzene
2,2'-oxybis(1-chloropropane)
2,6-Dinitrotoluene
4,6-Dinitro-2-methylphenol
Bis(2-chloroethoxy)methane
Bis(2-ethylhexyl) phthalate
Bis(2-ethylhexyl)adipate
Diethyl phthalate
Di-n-butyl phthalate
Di-n-octyl phthalate
Isophorone
Nitrobenzene
N-Nitroso-di-n-propylamine

Low Level Detection of Phenols

1 μ l of 5 pg/ μ L standard = 500 fg on column

Cal Standard 3 - 5 pg/ μ L

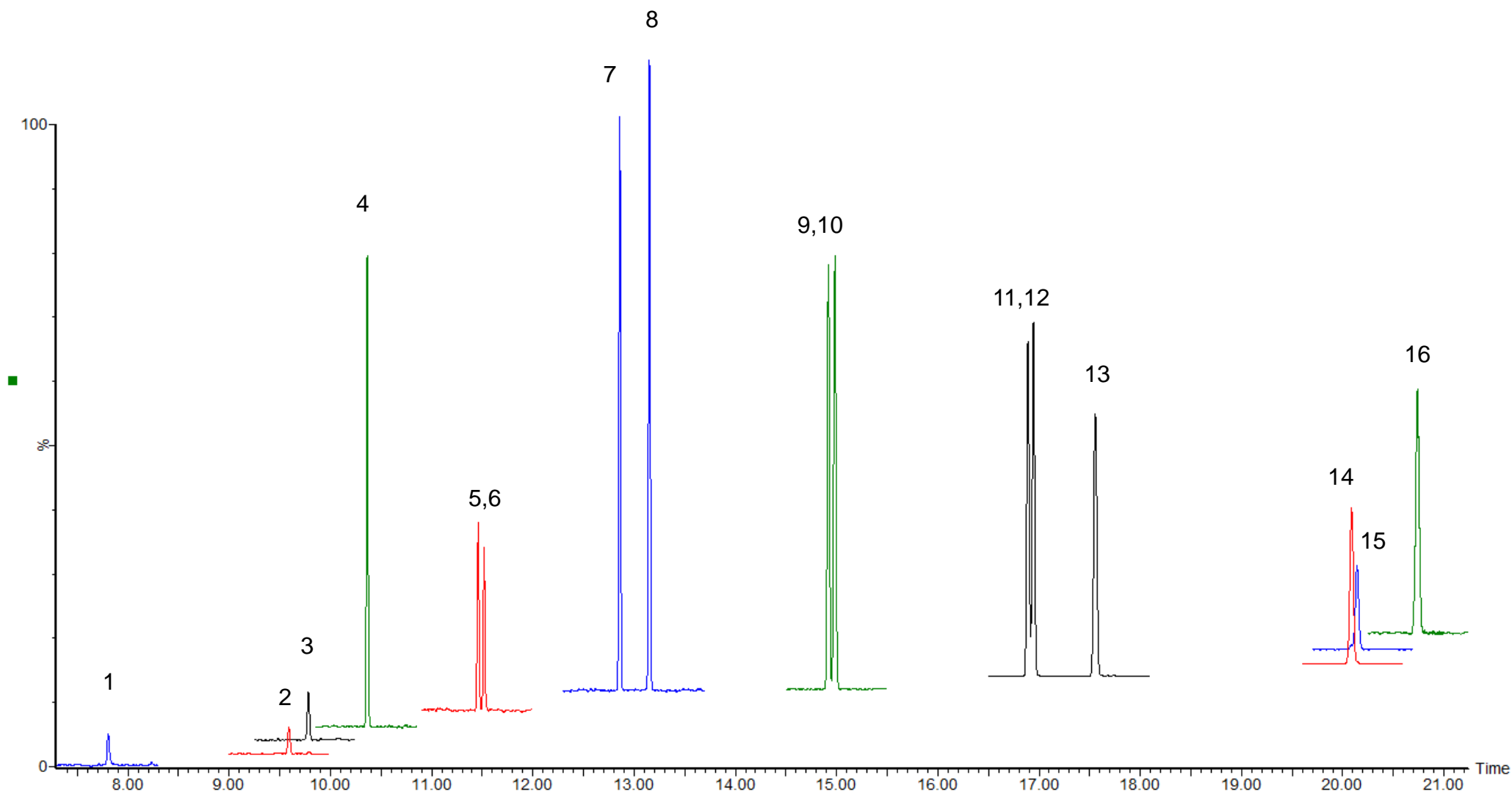
- 1 Phenol
- 2 2-Chlorophenol
- 3 2-Methylphenol
- 4 3-Methylphenol & 4-Methylphenol
- 5 2-Nitrophenol
- 6 2,4-Dimethylphenol
- 7 2,4-Dichlorophenol
- 8 4-Chloro-3-methylphenol
- 9 2,4,6-Trichlorophenol
- 10 2,4,5-Trichlorophenol
- 11 2,4-Dinitrophenol
- 12 4-Nitrophenol
- 13 2,3,5,6-Tetrachlorophenol
- 14 2,3,4,6-Tetrachlorophenol
- 15 4,6-Dinitro-2-methylphenol
- 16 Pentachlorophenol



Low Level Detection of PAHs

1 µl of 1 pg/µL standard = 100 fg on column

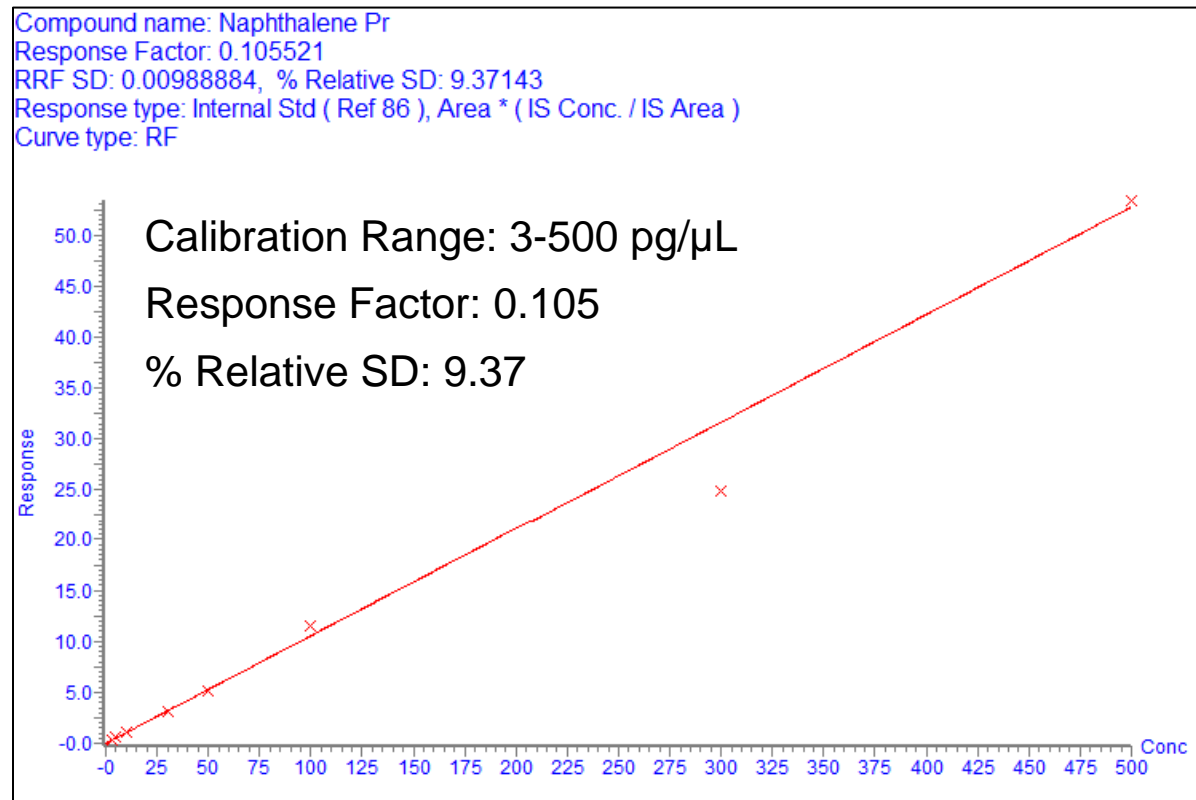
N2_SVOC_Curve102 Sm (Mn, 3x3)



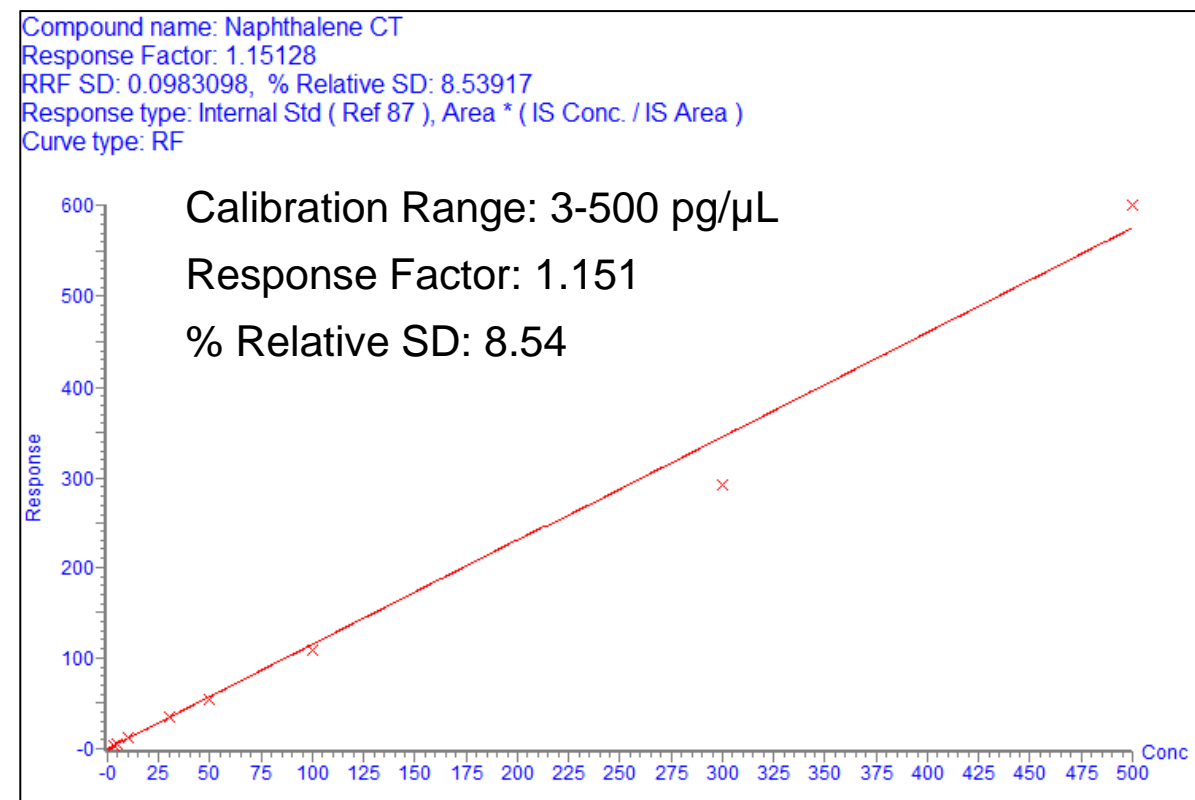
- 1 Naphthalene
- 2 Acenaphthylene
- 3 Acenaphthene
- 4 Fluorene
- 5 Phenanthrene
- 6 Anthracene
- 7 Fluoranthene
- 8 Pyrene
- 9 Benzo(a)anthracene
- 10 Chrysene
- 11 Benzo(b)fluoranthene
- 12 Benzo(k)fluoranthene
- 13 Benzo(a)pyrene
- 14 Indeno[1,2,3-cd] pyrene
- 15 Dibenz(a,h)anthracene
- 16 Benzo[g,h,i]perylene

Example Calibration Curves for Naphthalene

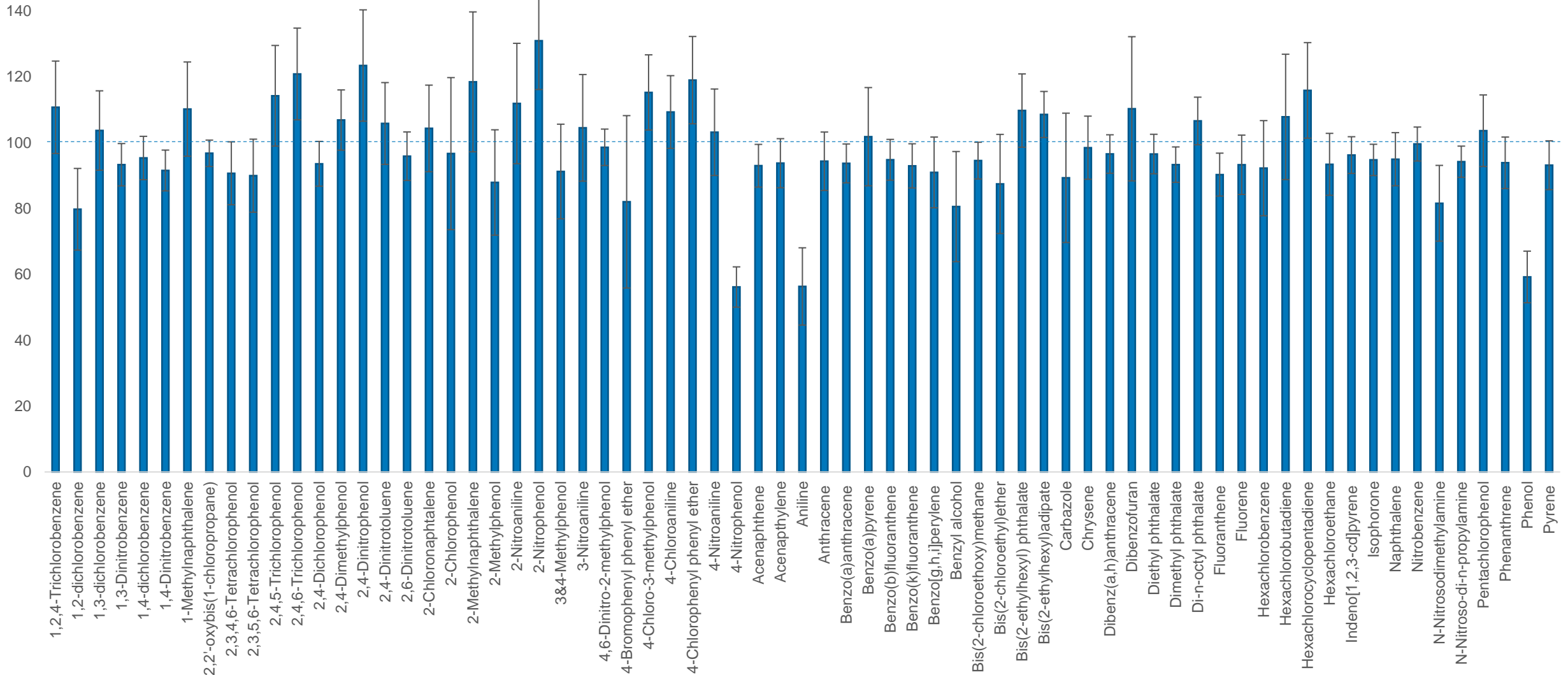
Naphthalene Protonation



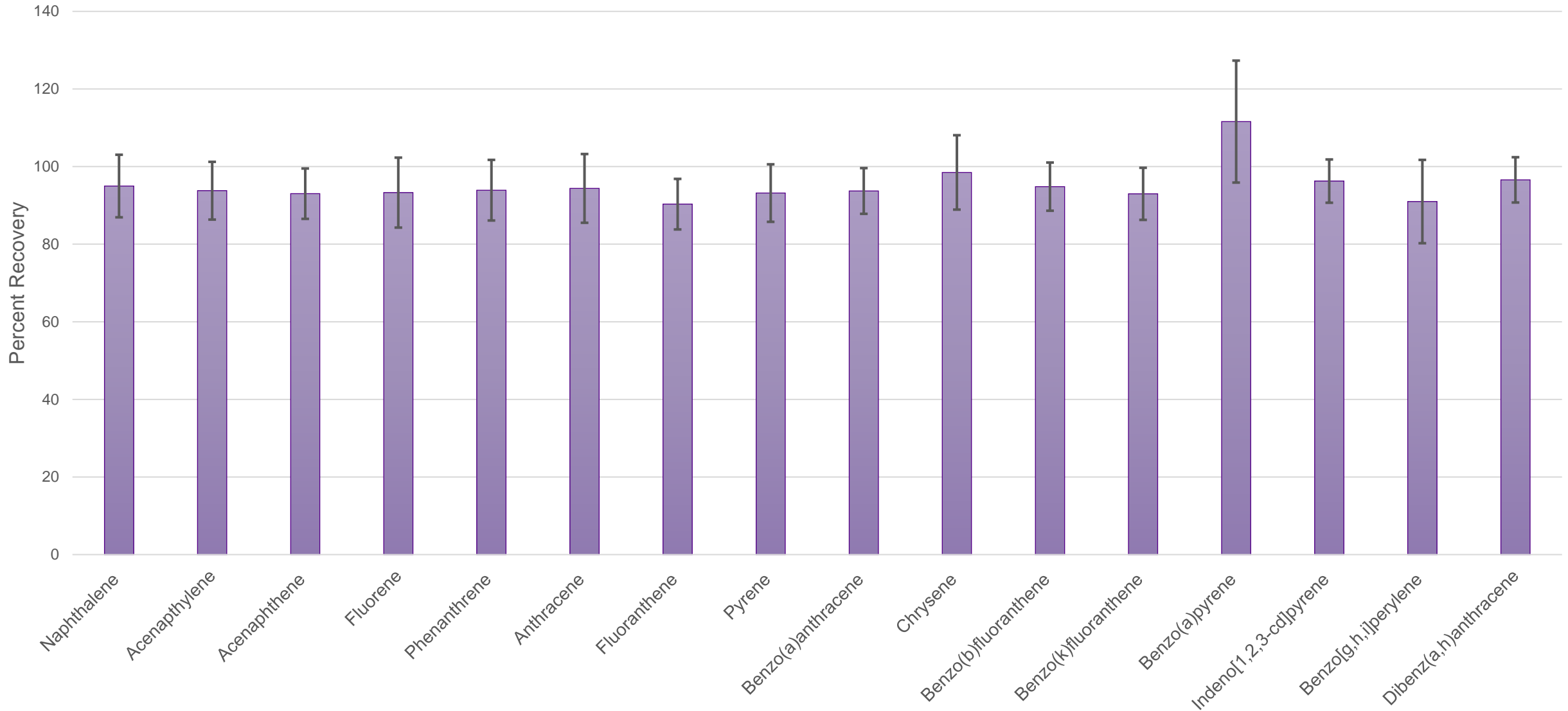
Naphthalene Charge Transfer



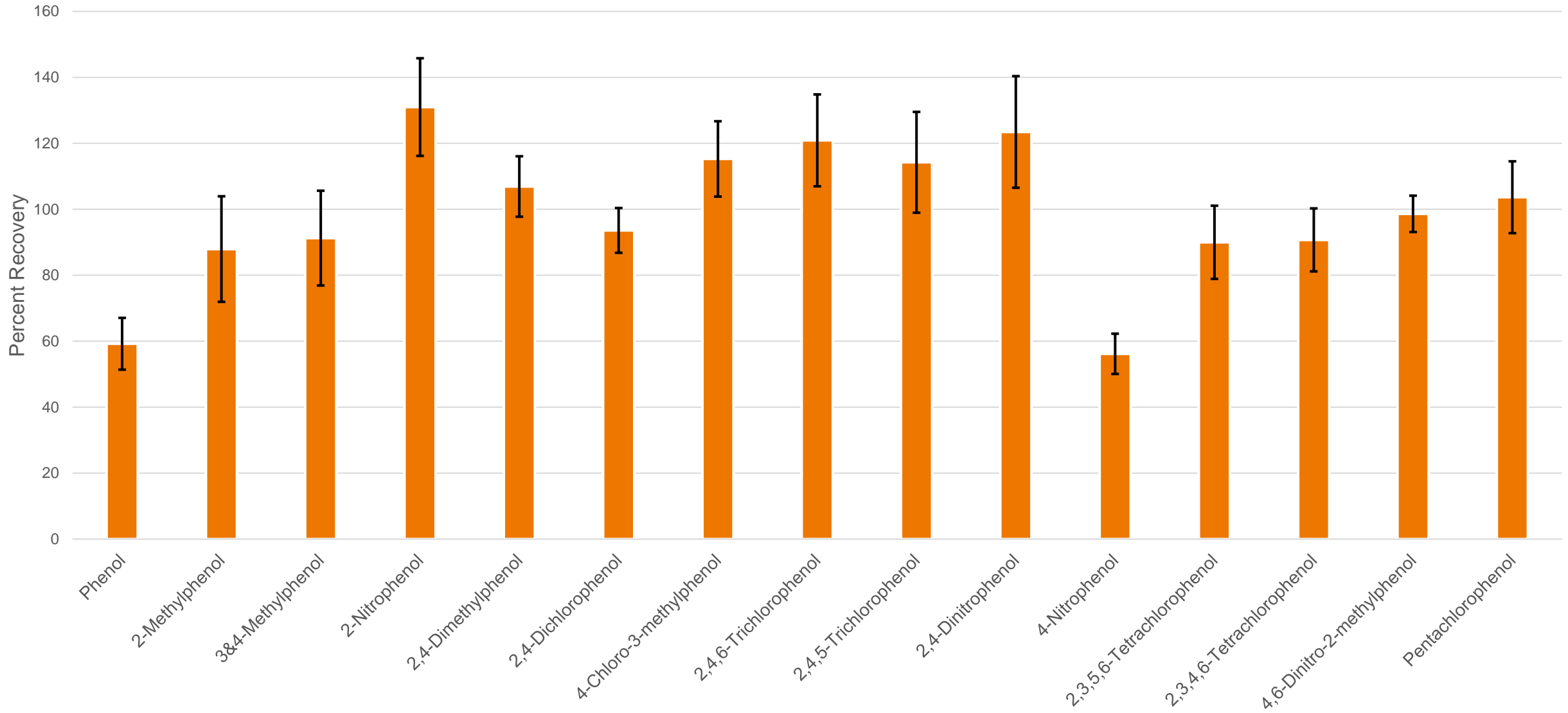
Recoveries – All but dibutyl phthalate (laboratory contaminant)



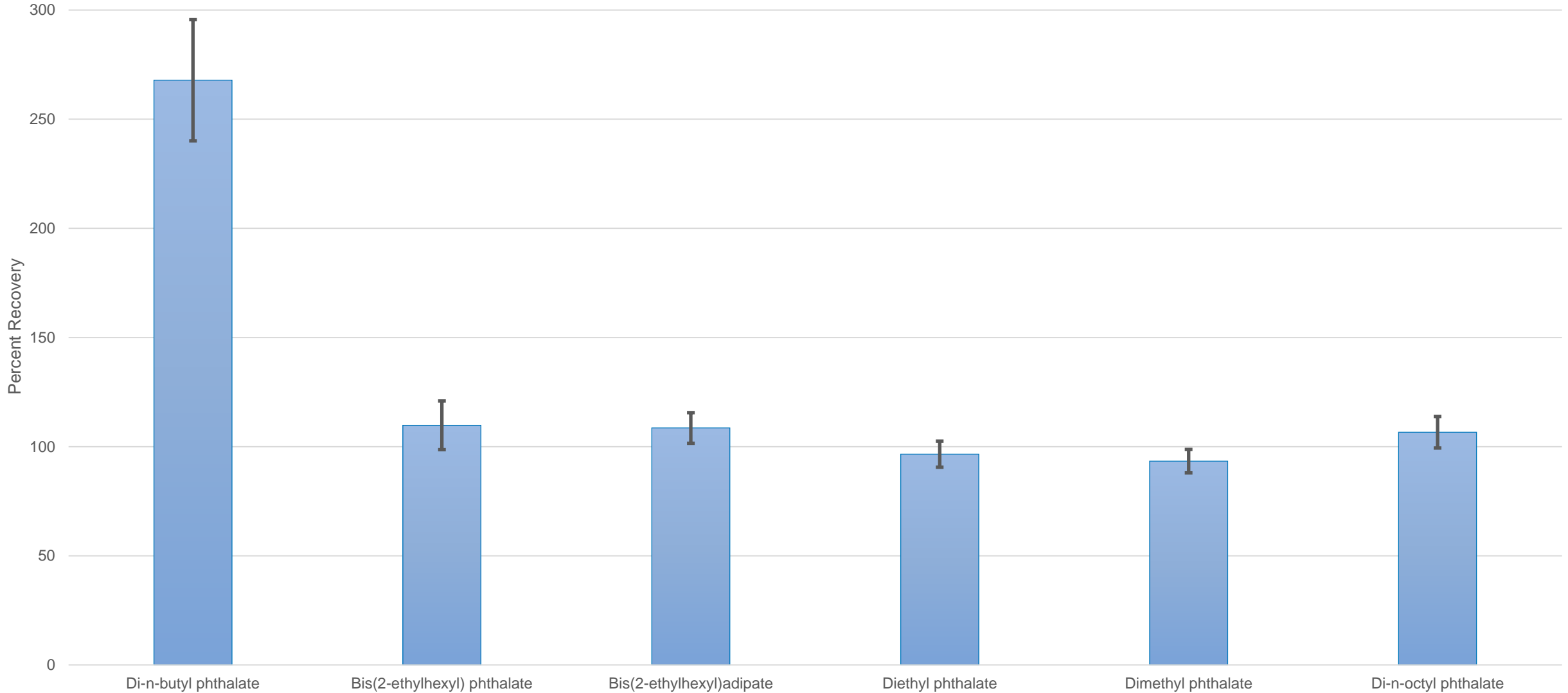
Recoveries: PAHs



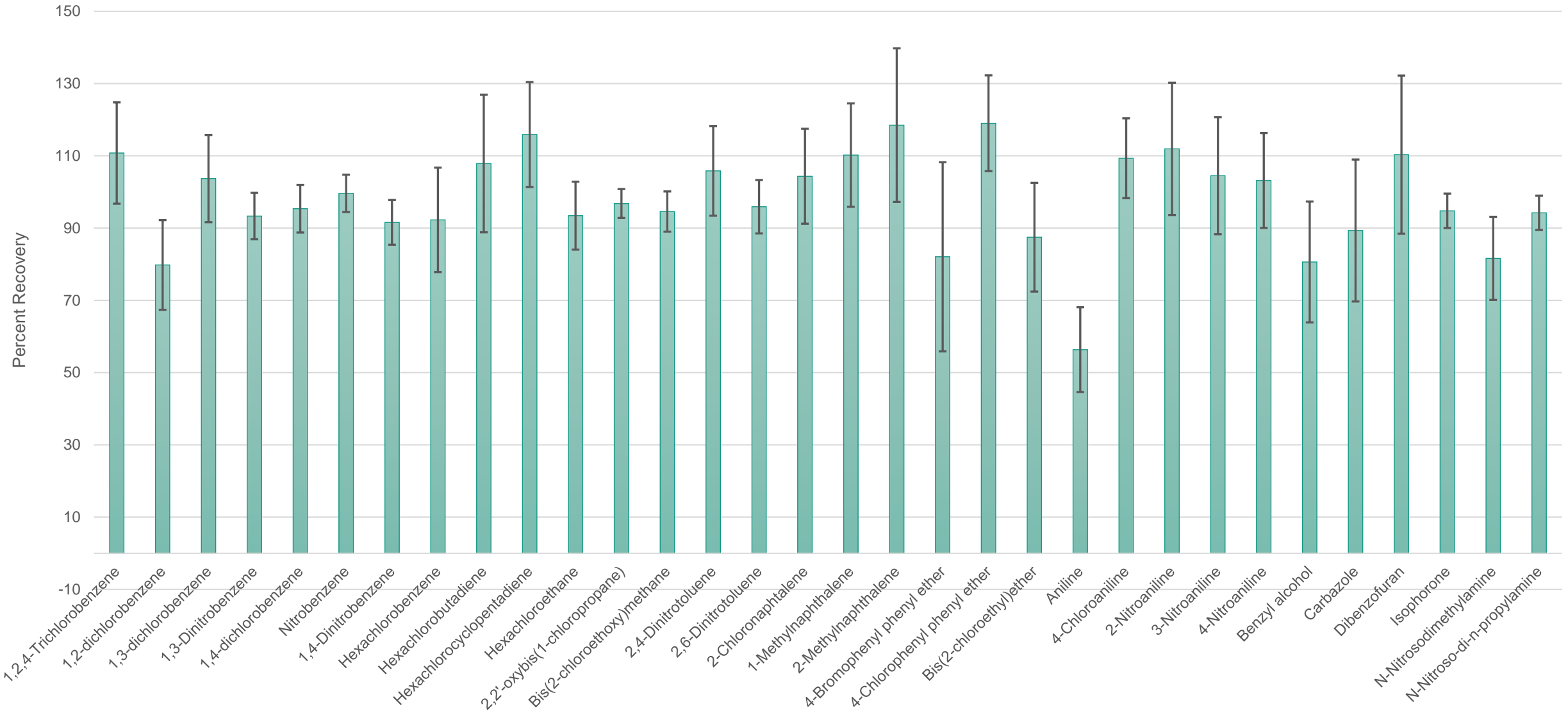
Recoveries: Phenols



Recoveries: Phthalates



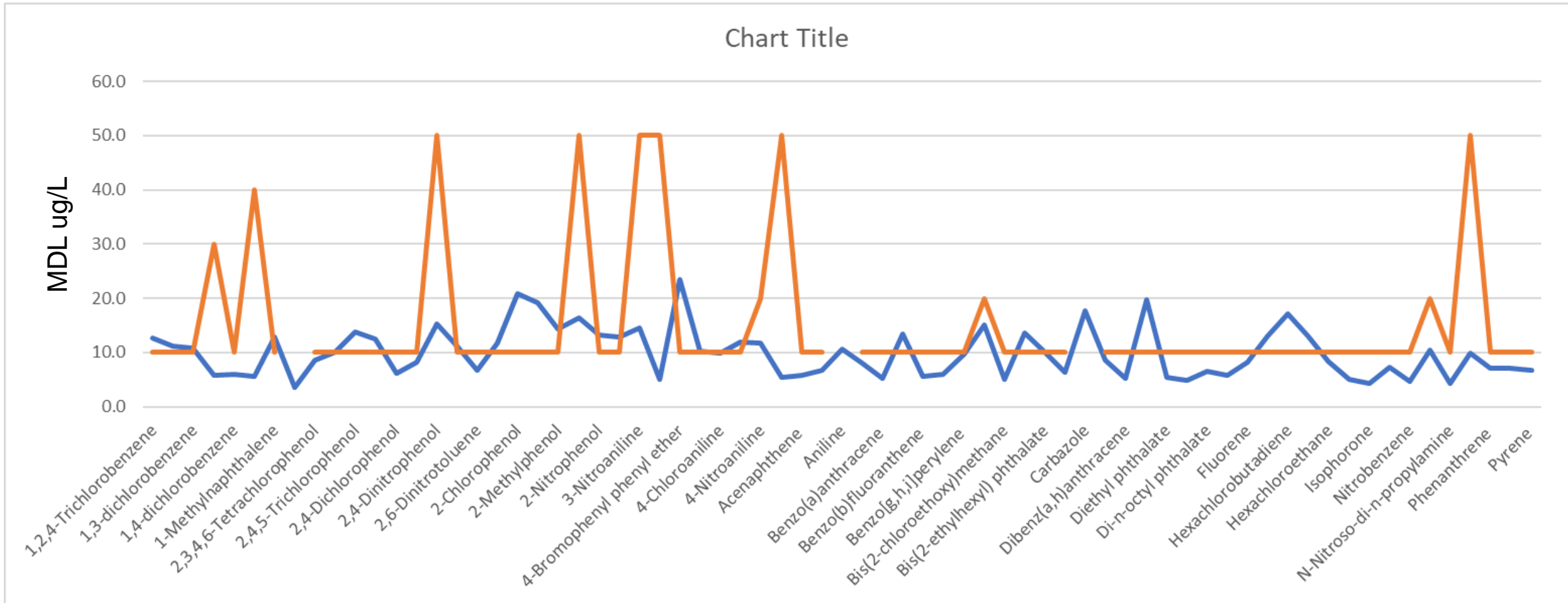
Recoveries: All Other Compounds



Compounds with Average % Recovery <80% or >120%

Compound Name	RRF %RSD	Average % Recovery	% Recovery Std Dev
Aniline	9.94	56.4	11.7
Phenol	17.59	59.2	7.8
1,2-Dichlorobenzene	23.47	79.8	12.4
2,4,6-Trichlorophenol	19.84	120.9	13.9
2,4-Dinitrophenol	18.15	123.4	16.9
2-Nitrophenol	19.3	131.0	14.8
Di-n-butyl phthalate	11.64	267.9	27.7

Method Detection Limit (MDL) Calculated



- Blue is Calculated Value
- Orange is USEPA Method 8270 Reference Value

Summary and Future Work

- The scaled sample prep showed a promising initial proof of performance
 - All compounds were detected in the n=8 replicate extracts
 - Only 7 of 73 compounds were outside the range of 80-120% recovery
- APGC can use N₂ as a carrier gas without seeing a loss in performance.
 - Scaled column dimensions allowed for a 22 min chromatographic method
 - The Rxi-SVOCms column had nice peak shape for phenolic compounds
 - Chromatographic performance was good enough to separate closely eluting isomers
- Additional testing will see if the sample preparation method can be improved:
 - Fewer extraction steps
 - Test with a single pH for extraction

Thank You for Your Attention

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