

Environment Testing America



Progress on the Use of Accurate Mass qTOF for PFAS Investigations

Charles Neslund Scientific Officer and PFAS Practice Leader Eurofins Environment Testing America





The tools for **PFAS** forensics are a developing area of applications. We currently have several tools already in use that can be applied towards forensic investigations;

 Chemical Fingerprinting

- Isomer comparison
- Applications of TOP Assay

🛟 eurofins



Additional techniques that are gaining in use and application

Total Organic Fluorine Analysis
Non-Target Analysis

🛟 eurofins





Source: ITRC Naming Conventions and Physical Chemical Properties fact sheet



Chemical Fingerprinting – PFAS by Isotope Dilution



Matrices

- · Potable water
- Nonpotable water
- Soil/sediment
- Tissue/biota
- Dust wipes
- Landfill leachate
- AFFF Formulations

70 Compounds

Solid Phase Extraction/Cleanup using weak anion exchange

Isotope Dilution quantitation

• 25+ isotopically labeled internal standards

Injection Standards for monitoring instrument vs extraction performance

Advantages

- · Isotope Dilution offers the highest degree of quantitative accuracy and precision
- · Broadest list of compounds and widest range of matrices
- Lowest reporting limits across matrices

🛟 eurofins



🛟 eurofins

| Perfluorobutanoic acid (PFBA) |
|---------------------------------------|
| Perfluoropentanoic acid (PFPeA) |
| Perfluorohexanoic acid (PFHxA) |
| Perfluoroheptanoic acid (PFHpA) |
| Perfluorooctanoic acid (PFOA) |
| Perfluorononanoic acid (PFNA) |
| Perfluorodecanoic acid (PFDA) |
| Perfluoroundecanoic acid (PFUnA) |
| Perfluorododecanoic acid (PFDoA) |
| Perfluorotridecanoic Acid (PFTriA) |
| Perfluorotetradecanoic acid (PFTeA) |
| Perfluorobutanesulfonic acid (PFBS) |
| Perfluorohexanesulfonic acid (PFHxS) |
| Perfluoroheptanesulfonic Acid (PFHpS) |
| Perfluorooctanesulfonic acid (PFOS) |
| Perfluorodecanesulfonic acid (PFDS) |
| Perfluorooctane Sulfonamide (FOSA) |
| Perfluoro-1-pentanesulfonate (PFPeS) |
| PFPrA |

g

| Perfluoro-1-nonanesulfonate (PFNS) | NFDHA | PFO4DA |
|--|------------|--------------------|
| Perfluorododecanesulfonic acid (PFDoS) | PFEESA | PFO3OA |
| Perfluoro-n-hexadecanoic acid (PFHxDA) | PFMPA | PFO2HxA |
| Perfluoro-n-octadecanoic acid (PFODA) | PFMBA | PFO5DA |
| IMeFOSAA | 3:3 FTCA | R-EVE |
| IEtFOSAA | 5:3 FTCA | NVHOS |
| IEtFOSA | 7:3 FTCA | Hydro-EVE Acid |
| IMeFOSA | 6:2 FTCA | EVE Acid |
| IMeFOSE | 8:2 FTCA | R-PSDA |
| IEtFOSE | 10:2 FTCA | Hydrolyzed PSDA |
| :2FTS | 6:2 FTUCA | R-PSDCA |
| :2FTS | 8:2 FTUCA | PS Acid |
| :2FTS | 10:2 FTUCA | Hydro-PS Acid |
| 0:2FTS | PFECHS | 4:2 FTOH |
| DONA | PFPrS | 6:2 FTOH |
| IFPO-DA (GenX) | PFMOAA | 7:2S FTOH |
| 1CI-PF3OUdS | PFECA G | 8:2 FTOH |
| CI-PF3ONS | MTP | 10:2 FTOH |
| PMPA | PEPA | |



- Fluorotelomer Alcohols
 - GCMSMS method
 - Water and solids
 - Instrumental set-up like 8270E and extractions like 3510 and 3540/50
 - Current compound list
 - 4:2 Fluorotelomer alcohol
 - 6:2 Fluorotelomer alcohol
 - 7:2S Fluorotelomer alcohol
- 8:2 Fluorotelomer alcohol
- 10:2 Fluorotelomer alcohol



Chemical Fingerprinting





Herzke, et al., 2012, Chemosphere, 88, 980-987

🛟 eurofins



$$F_3C-CF_2-CF_2-CF_2-CF_2-CF_2-CF_2-CF_2-SO_3$$

CF₃

Linear Perfluorooctane sulfonate (PFOS)

Branched Perfluorooctane sulfonate (PFOS)

Figure 4-1. Linear and one branched isomer of PFOS

ITRC PFAS Fact Sheet Naming Conventions April 2020



Isomer Comparison





Chromatogram of PFOS Standard of Linear Isomer

Chromatogram of PFOS Standard of Branched/Linear Mix Typical Ratio



🛟 eurofins

Isomer Comparison





Chromatogram of PFOS Sample with Branched/Linear Mix High Bias Ratio

Chromatogram of PFOS Sample with Branched/Linear Mix Low Bias Ratio









What is the Total Oxidizable Precursor (TOP) Assay?

A PFAS sample preparation technique

- Indicates presence of unidentified precursors
- Used in conjunction with standard analysis
- Contrasts pre and post oxidation results



What the TOP Assay is NOT

- A risk assessment tool
- Total PFAS methodology
- Identify Unknown PFAS
- > Mass balance PFAS
- Non-target identification

https://pubs.acs.org/doi/10.1021/es302274g

🛟 eurofins

Environment Testing America 12



| Compound | Pre-Ox | Post-Ox | Difference |
|----------|----------|---------|------------|
| PFBA | ND | 98 ng/l | 98 ng/l |
| PFPeA | ND | 87 ng/l | 87 ng/l |
| PFHxA | 5 ng/l | 61 ng/l | 56 ng/l |
| 6:2 FTS | 100 ng/l | ND | - 100 ng/l |
| PFHpA | 11 ng/l | 32 ng/l | 21 ng/l |
| PFOA | 7 ng/l | 26 ng/l | 19 ng/l |
| PFOS | 56 ng/l | 52 ng/l | - 4 ng/l |
| 8:2 FTS | 26 ng/l | ND | - 26 ng/l |
| PFNA | ND | 5 ng/l | 5 ng/l |



Total Organofluorine (TOF)



CASE STUDY

Total Organic Fluorine (TOF)410 mg F/kgExtractable Organic Fluorine (EOF)390 mg F/kgLC-MS/MS ΣPFAS (n=28)120 mg/kg



- Sample (or treated sample) is combusted in a furnace at 900°C – 1100°C
- Effluent collected in buffer and injected into ion chromatograph (IC)
- Quantify fluorine (as fluoride) content
- Compare ratio of total (or extractable) fluorine to total PFAS

CIC: Combustion Ion Chromatography





Environment Testing America 14





Quadrupole Time of Flight (qToF)

🛟 eurofins



Technique utilizes LC/MS-qTOF (quadrupole time of flight mass spectrometry)

- Technique allows for determination of accurate mass (0.0001 amu)
- Initial differentiation based on extraction of sample
- Then analysis of targeted compounds (knowns) to remove those from "background"
- Compare remaining peaks to limited mass spectral libraries to identify the known/unknowns
- Remaining peaks are unknowns and would rely on regression of accurate mass determinations for possible identification



Non-Target Analysis



Problems?

Accurate mass solves a variety of PFAS problems

No More Limitations

Precursors without TOP Assay No LIMS constraints Want to know all byproducts?

Byproducts?

SWATH uses a moving small mass window for nontarget MS/MS spectra; can capture all byproducts



QTOF exact mass analysis for > 40 PFAS analytes

Exact mass confirmation of 'suspect' positive results

Mitigation of matrix effects for short chain analytes

Application for PFAS lacking standards and unknowns (NTAs)

🛟 eurofins

Non-Target Results



| # | Applyte Peak Name | Precursor | Found At | Library Hit | Library | Formula Finder | Formula Finder | Combined |
|---------------------|---------------------------|-----------|----------|--|---------|------------------|----------------|----------|
| | Analyte Feak Malle | Mass | Mass | Library m | Score | Results | Score | Score |
| 47 | 207.1384 / 9.59 | 207.140 | 207.1386 | Ser-Cys (NIST) | 86.3 | C13H20O2 | 77.080 | 81.702 |
| 75 | 205.1582 / 10.62 | 205.159 | 205.1591 | Met-Gly (NIST) | 82.3 | C8H23N4P | 83.194 | 82.724 |
| 93 | 271.2263 / 11.21 | 271.227 | 271.2271 | DLbetaHydroxypalmitic acid (NIST) | 81.8 | C16H32O3 | 68.518 | 75.154 |
| 119 | 265.1468 / 12.04 | 265.148 | 265.1472 | Dodecyl sulfate (NIST) | 99.3 | C8H24N6P2 | 78.457 | 88.862 |
| 127 | 199.1699 / 12.08 | 199.171 | 199.1699 | Dodecanoic acid (NIST) | 93.5 | C12H24O2 | 81.919 | 87.725 |
| 128 | 297.1516 / 12.16 | 297.153 | 297.1520 | Ricinoleic acid (NIST) | 97.5 | C8H21F2N8P | 89.209 | 93.349 |
| 129 | 205.1591 / 12.22 | 205.160 | 205.1592 | 2,6-Di-tert-butylphenol (NIST) | 100.0 | C8H23N4P | 82.310 | 91.155 |
| 130 | 297.2424 / 12.22 | 297.243 | 297.2428 | Ricinoleic acid (NIST) | 97.5 | C18H34O3 | 71.444 | 84.466 |
| 146 | 514 9789 / 12 55 | 514 980 | 514 9792 | CI-PFOS (chloro-perfluorooctane sulfonate) | 89.8 | C8H13EN6O15S2 | 98.473 | 94.123 |
| 140 514.57057 12.55 | 514.51057 12.55 | 514.500 | 514.5152 | (neg) | 00.0 | C011311001332 | 50.415 | 04.120 |
| 152 | 309.1728 / 12.64 | 309.174 | 309.1733 | Ethylene glycol dodecyl ether sulfate (NIST) | 100.0 | C14H30O5S | 73.122 | 86.561 |
| 168 | 531.0069 / 12.94 M- H- | 531.008 | 531.0081 | CI-PFENS neg | 81.5 | Too many formula | 0.000 | 40.743 |
| 171 | 353.1999 / 12.94 | 353.201 | 353.1996 | Diethylene glycol dodecyl ether sulfate (NIST) | 99.6 | C15H29F3N4S | 91.220 | 95.397 |
| 176 | 241.2162 / 13.06 | 241.217 | 241.2165 | N2-Trifluoroacetyl-L-glutamine (NIST) | 89.3 | No formula found | 0.000 | 44.666 |
| 192 | 293.1788 / 13.45 | 293.180 | 293.1784 | Myristyl sulfate (NIST) | 97.8 | C14H30O4S | 73.162 | 85.459 |
| 216 | 253.2158 / 14.02 | 253.217 | 253.2168 | cis-7-Hexadecenoic acid (NIST) | 97.8 | C16H30O2 | 77.687 | 87.726 |
| 220 | 339.1986 / 14.08 | 339.200 | 339.1991 | Tridecylbenzenesulfonic acid (NIST) | 80.2 | C13H33N4O2PS | 89.239 | 84.697 |
| 260 | 281.2480 / 14.90 | 281.249 | 281.2479 | 1,4-D-Xylobiose (NIST) | 100.0 | C18H34O2 | 73.760 | 86.880 |
| 300 | 407.2938 / 15.97 | 407.295 | 407.2942 | .gammaMuricholic acid (NIST) | 96.5 | C21H37FN6O | 95.929 | 96.220 |
| 327 | 311.2943 / 17.36 | 311.295 | 311.2943 | Benzenesulfonic acid, 4-undecyl- (NIST) | 76.3 | C16H36N6 | 53.227 | 64.777 |
| 434 | 265.1465 / 26.87 | 265.148 | 265.1470 | Dodecyl sulfate (NIST) | 84.6 | C13H27FS2 | 84.840 | 84.720 |



Non-Target Results





🛟 eurofins

Non-Targeted Analysis





🛟 eurofins

Case Study





Note: the fragmentation pattern of the entire peak, although the 368.9 is present (Red Arrow) the primary fragmentation is 112.98.



Case Study







Note: The fragmentation of m/z 412.96 for the peak at 10.343 includes the pattern for PFOA, it also includes a predominant fragment at m/z 112.98; 134.98 and 184.95.

🛟 eurofins





Note: The fragmentation pattern at 10.50 (RT of peak in CAL standard) – Fragmentation of 412.96 shows the presence of 368.97 however not as the predominant peak. Also in the MS fragmentation, m/z 414.97 is predominant. The fragmentation mass error for m/z 368.98 is extremely high at 771.8 ppm.



Summary



Targeted PFAS

All Matrices – Up to 70 Compounds

Strengths: Selectivity, Sensitivity at ~1-5ppt Can be used for risk assessment Weaknesses: Limited list of compounds

Non-Target Analysis

All Matrices – Unknowns

Strengths: Ability to identify 'unknowns' with specificity Ability to conduct novel compound identification Weaknesses: Limited to current libraries Limited quantitation

Method Toolbox \mathbf{M}

TOP Assay

All Matrices – Oxidizable Precursors Strengths: Sensitivity at ~1-5ppt Specific to 'unknowns' with potential to convert to risk drivers Weaknesses: Not specific Does not complete a mass balance

Total Organic Fluorine

All Matrices – Organic Fluorine

Strengths: Closest to a mass balance Weaknesses: Sensitivity at ~1-5ppb No selectivity

🛟 eurofins

Environment Testing America 24



Thank You

Charles Neslund Charles.Neslund@Eurofinset.com 717-799-0439

