



Environment Testing America

Bubble Trouble Solution



Mark Bruce August 11, 2020

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Trouble from One Bubble





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History Lesson

March 1990

Labcert Bulletin published by the Office of Drinking Water

Note: By the time the sample arrives at the lab, a small bubble may have developed. As long as this is no larger than the size of a pea the sample may be considered valid.

Courtesy of Jerry Parr

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6.0 SAMPLE COLLECTION, PRESERVATION, AND HANDLING

6.1 ... Samples should be stored in capped vials, with minimum headspace, at 4°C or less in an area free of solvent fumes. The size of any bubble caused by degassing upon cooling the sample should not exceed 5 - 6 mm.



The Solution

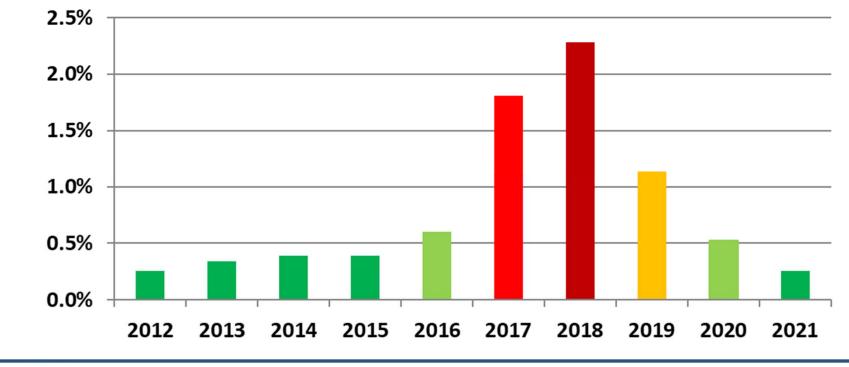






Recent History - Year

%VOC water samples with headspace non-conformance memos

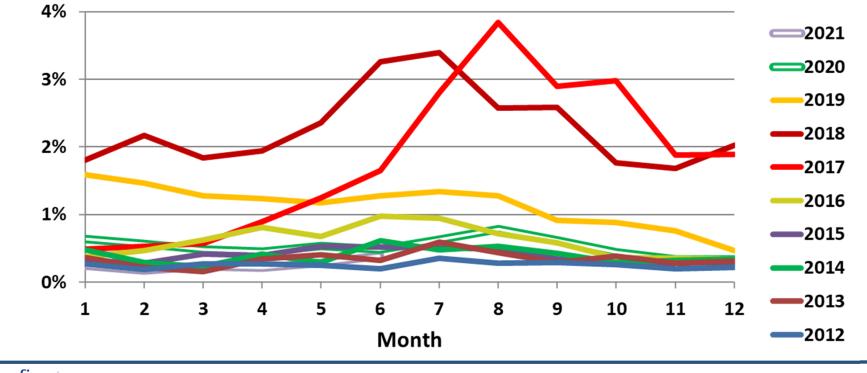


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Recent History - Month



%VOC water samples with headspace non-conformance memos



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Factors Studied

Sample	Collection Technique
Carbonate concentration	Entrained air bubbles due to vigorous pouring
Dissolved gas concentration	Water dome height above the top of bottle
Water temperature at time of sampling	Under tightened cap
	Over tightened cap





Factors Studied

Transport and Storage

Temperature drop to 0-6°C

Cap loosening due to vibration or temperature changes

Air pressure changes due to air shipment or elevation changes between site and laboratory

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Factors Studied

Container	
Temperature of bottle at time of use	Stiffness (durometer) of septum
Long term storage temperature of bottles	HCI preservative source
Diameter of septum, cap and top of glass bottle	Thickness of Teflon face on septum
Thread match between cap and bottle	Press fit vs bonded septum
Inside surface of cap that contacts the septum	Shape of septum & position in cap
Glass bottle sealing surface (rough, round, flat)	Septum position after tightening

VOC Vials Studied Extensively



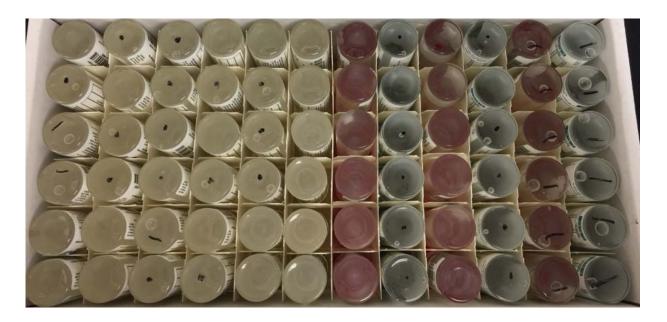
3+ Years

5 suppliers

57 Experiments

2584 Bubbles

5526 Vials



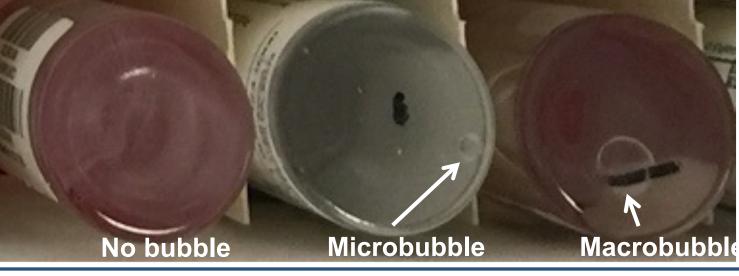
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Headspace Classification



No bubble Microbubble (<6 mm)

Macrobubble (>6 mm)



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Cap Tightness Classification

- Level 0 septum is bowed down
- Level 1 septum is flat
- Level 2 slightly bowed upward, (6-8 in-lb)
- Level 3 bowed upward, extends 1-2 mm above the cap (10-12 in-lb)
- Level 4 –bowed upward, extends > 3 mm above the top of the cap



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Investigate all previously listed potential factors

Major client with two sites with recurring bubble trouble

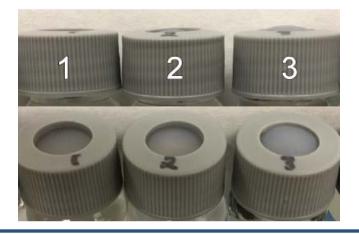
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Field Test

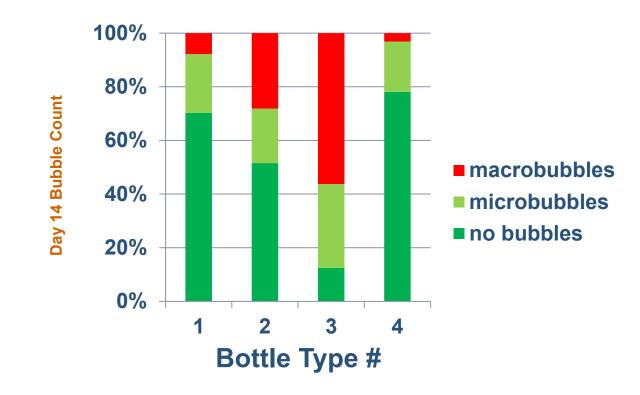
Experimental Factors

- Site (2)
- Well (10)
- Sampler (4)
- Bottle type (4)
- Cap tightness (3 levels, 1, 2 & 3)





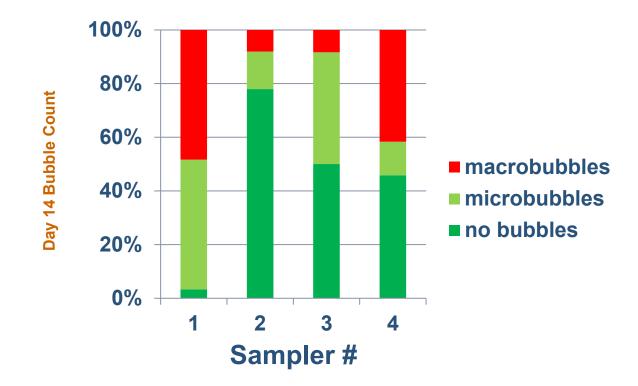
Bubbles by Bottle Type



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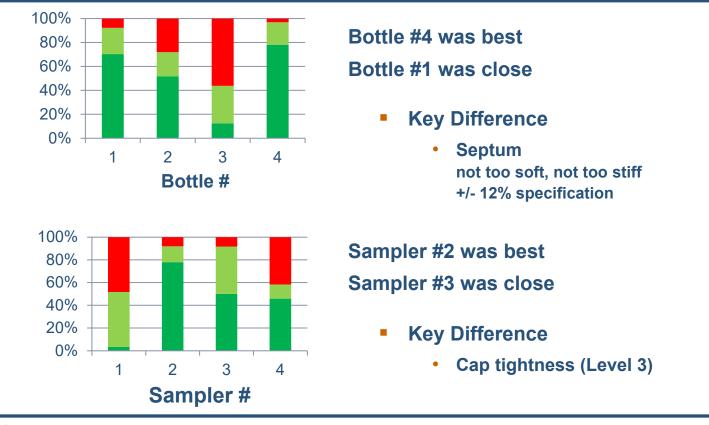
Bubbles by Field Sampler



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Interpretation





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Best Combination

Bottle #4

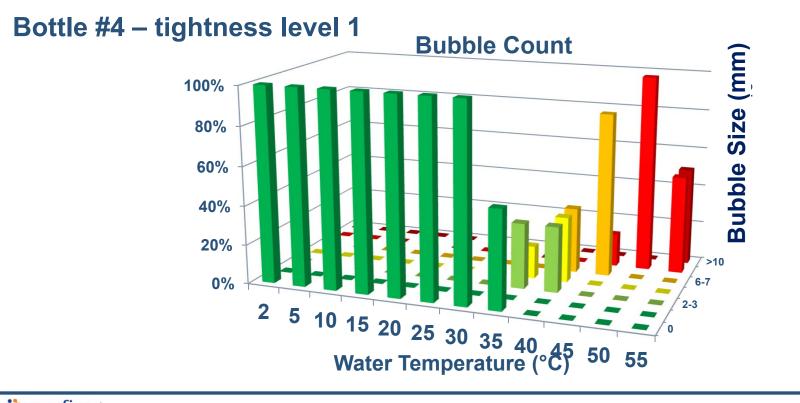
Sampler #2 (level 3 tightness)

No Bubbles 100% of vials Over 10 wells And 2 sites

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Sample Temperature

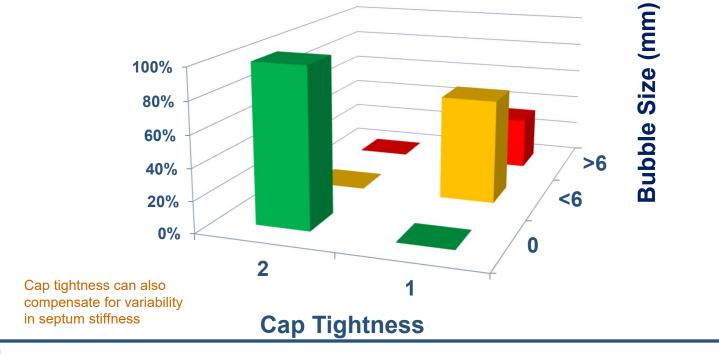


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How to Compensate for Warm Samples



Bottles 1 & 4, 35°C



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What is Happening?

Hypothesis

Sample water contracts on cooling

Pulling air past the septum seal to form a bubble

Larger the temperature drop or the less tight cap The more air pulled in

Hypothesis Test

Measure VOC increase due to air bubble

Cool vials in jar with ethyl ether in air 40 uL EE, 4 L air Low torque vial caps 4 in-lb High torque vial caps 12 in-lb



Hypothesis Test – Bubble Count



Low torque vials – all had 6-8 mm bubbles

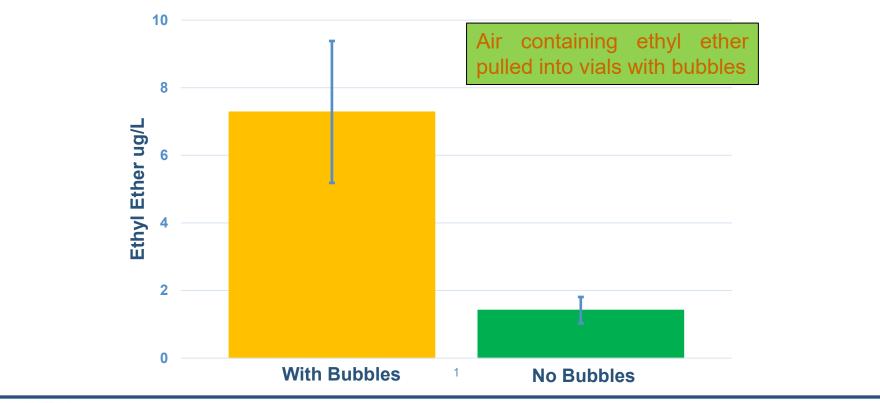
High torque vials – no bubbles



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Hypothesis Test - Results





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Conclusions



Use VOC vial with good septum and cap Cut resistant gloves / disposable gloves Water dome 2-4 mm Cap Tightness

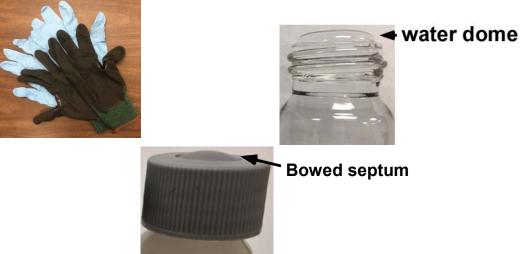
- at least Level 2
- better Level 3

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~10 in-lb

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Bubbling Questions?

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